



18th Workshop on Targetry and Target Chemistry

New metallic germanium target fabrication and dissolution techniques for the cyclotron production of positron-emitting ^{71}As and ^{72}As

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University of Wisconsin School of Medicine and Public Health

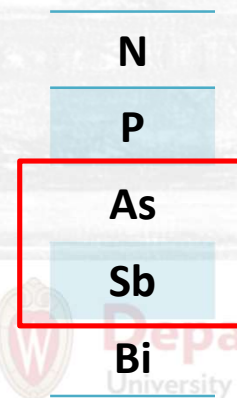
Research topic

^{77}As beta emitter and $^{71/72}\text{As}$ positron emitter

^{72}Se 8.4 d Ec only	^{73}Se 7.2 h 40 m		^{72}Se 0.89%	^{75}Se 119.8 d EC only	^{76}Se 9.4 %	^{77}Se 7.6 %	^{78}Se 23.8 %	^{79}Se 4 m $3 \times 10^5\text{y}$		
^{71}As 65.3 h β^+ : 816 keV...	^{72}As 26 h β^+ : 1117 keV.		^{73}As 80.3 h Ec only	^{74}As 17.8 d β^+ β^-		^{75}As 100 %	^{76}As 26.4 h β^- : 2962 keV		^{77}As 38.8 h β^- : 683 keV	^{78}As 90.7 m β^- : 4209 keV
^{70}Ge 20.57%	^{71}Ge 11.43d Ec only		^{72}Ge 27.45%	^{73}Ge 7.75%	^{74}Ge 36.5%	^{75}Ge 48 s 83 m		^{76}Ge 7.73%	^{77}Ge 53 s 11 h	

^{119}Sb MAe⁻ emitter

- Homologous relationship with ^{119}Sb



[1] P.A. Ellison et al. Bioconjug. Chem. 27 (2016) 179–188.
 [2] Y. Feng et al., Appl. Radiat. Isot. 143 (2019) 113–122.
 [3] Matthew D.Gott et al., Journal of Chromatography A, 1441(2016) 68-74

GeO₂ target

Pre irradiation:



GeO₂

After 4 μA*

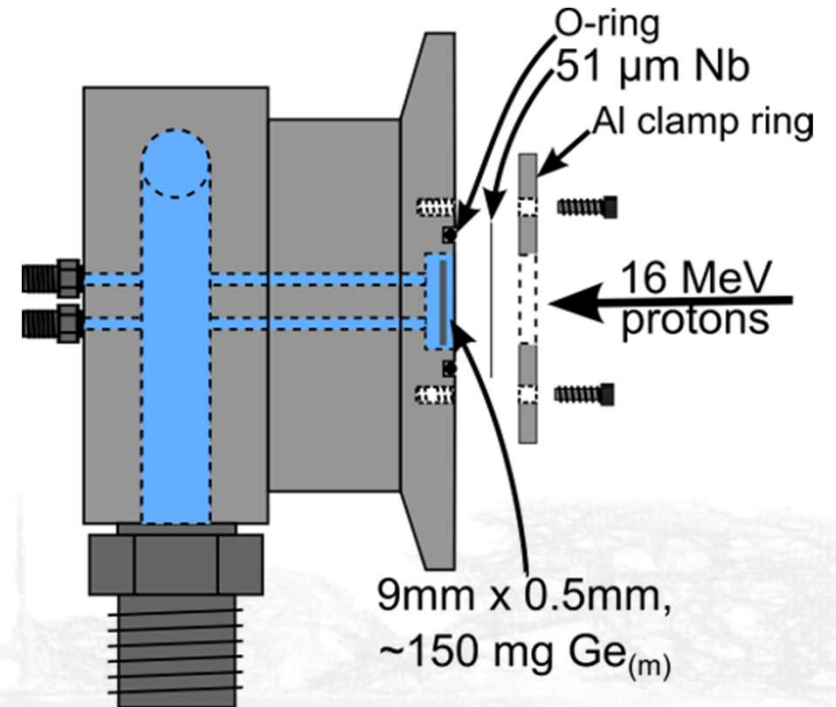


Difficult to dissolve it and As isolation yield decreased.

Problem...

Only tolerant to low beam.

4π-water-cooled target



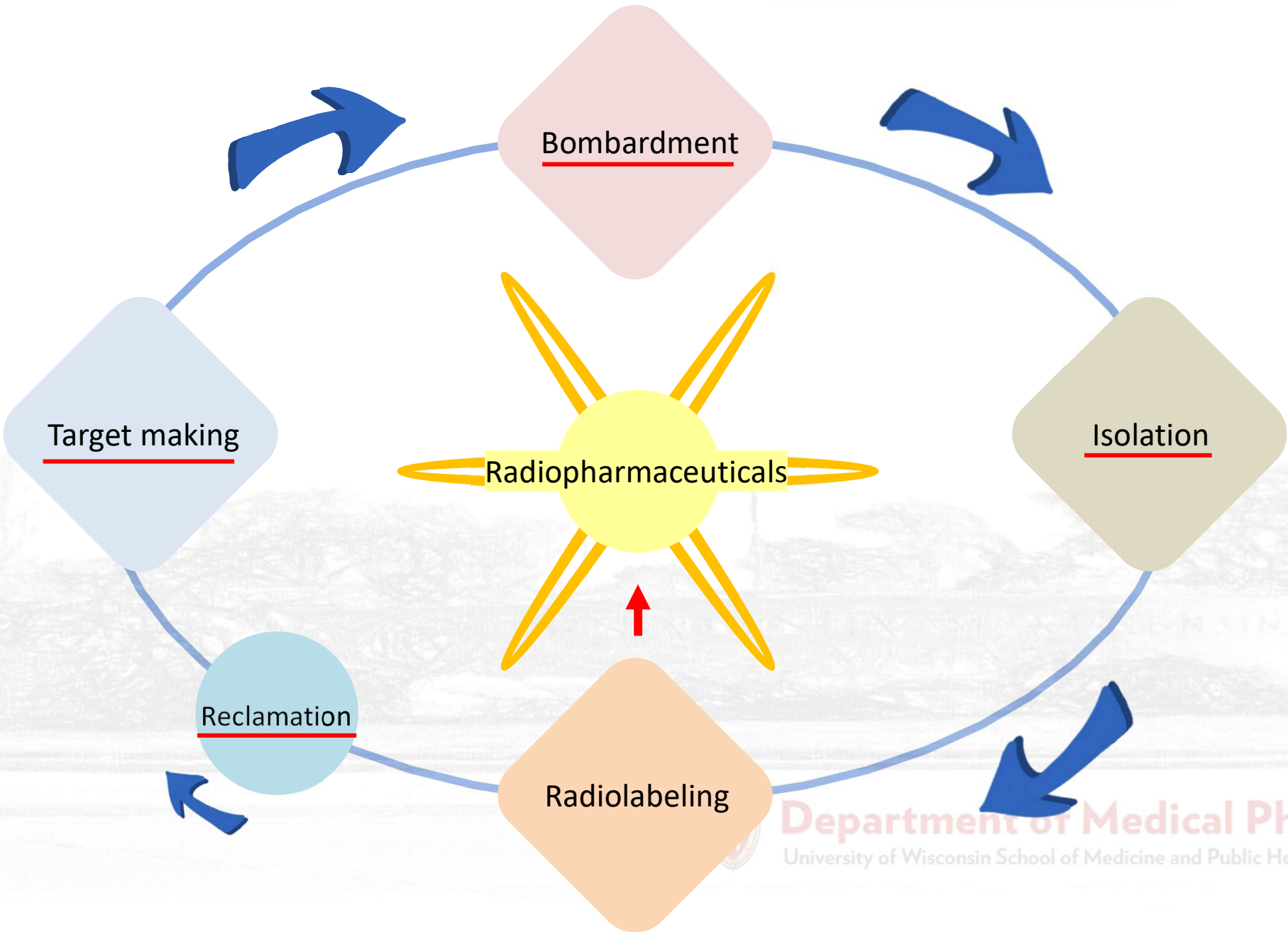
Problem...

it is incompatible with deuterons and commercial solid target irradiation systems.

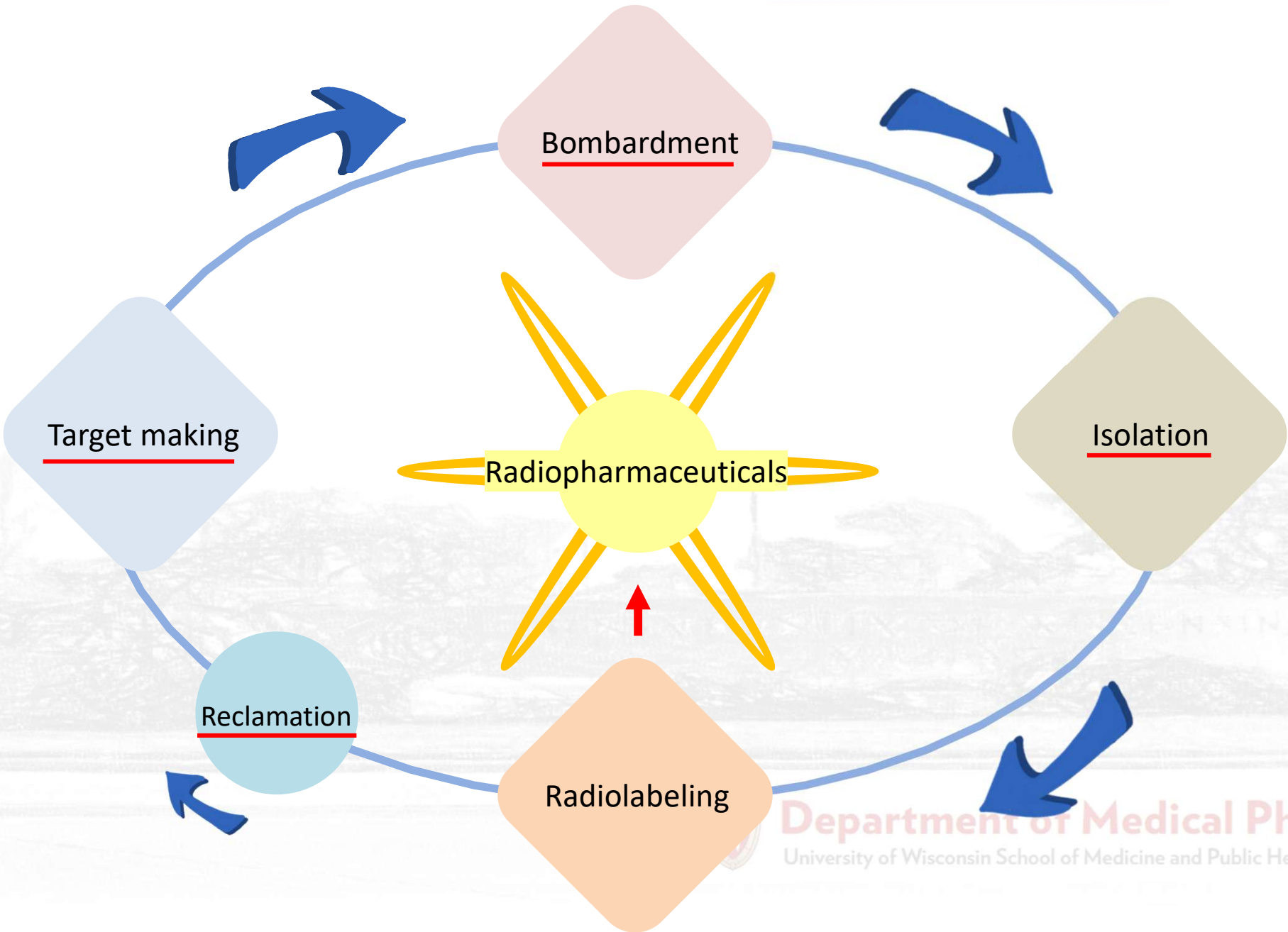
[4] P.A. Ellison et al., AIP Conf. Proc. 1509 (2012) 135-140.

[1] P.A. Ellison et al. Bioconjug. Chem. 27 (2016) 179–188.

Research topic



Research topic



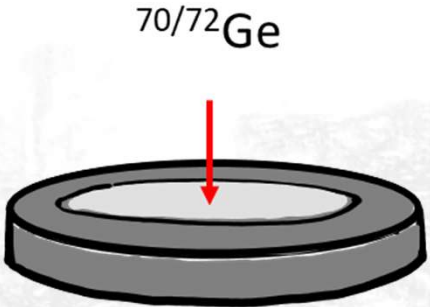
Research topic

Target making

Reclamation

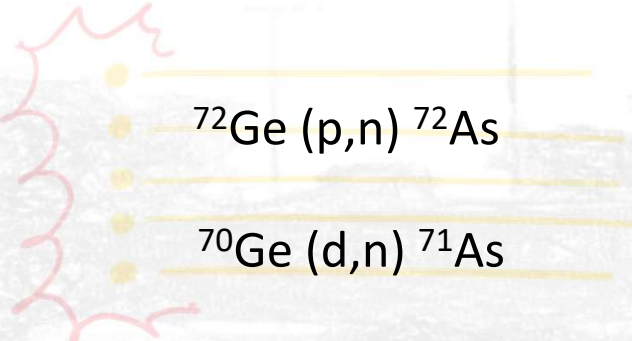
Bombardment

Isolation
(Cold experiment)

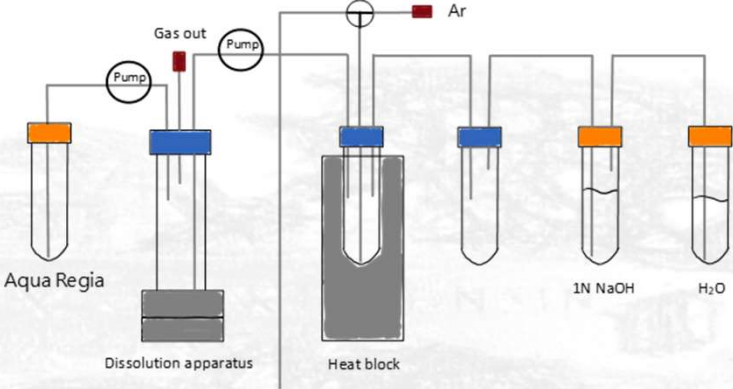


Target preparation

Ta coin
($\varnothing=10$ mm, 1 mm deep)



Bombardment



Dissolution and Separation



Target making

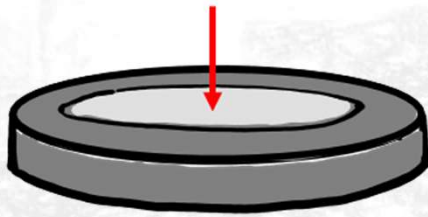
Target making

Bombardment

Isolation

Reclamation

$^{70/72}\text{Ge}$



Target preparation

Ta coin

($\varnothing=10$ mm, 1 mm deep)

Ge bb making



Hot pressing



Ge Target

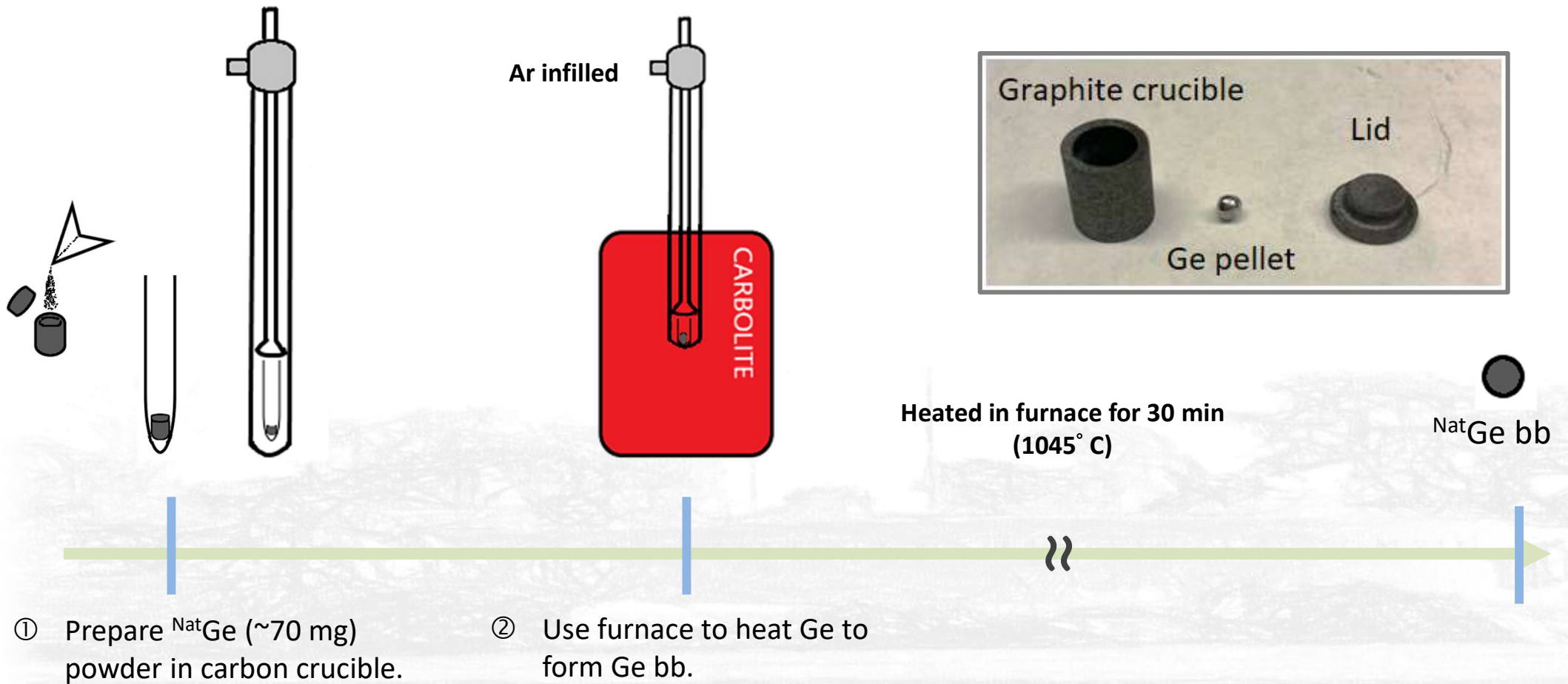


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Target making: Germanium bb making

Method

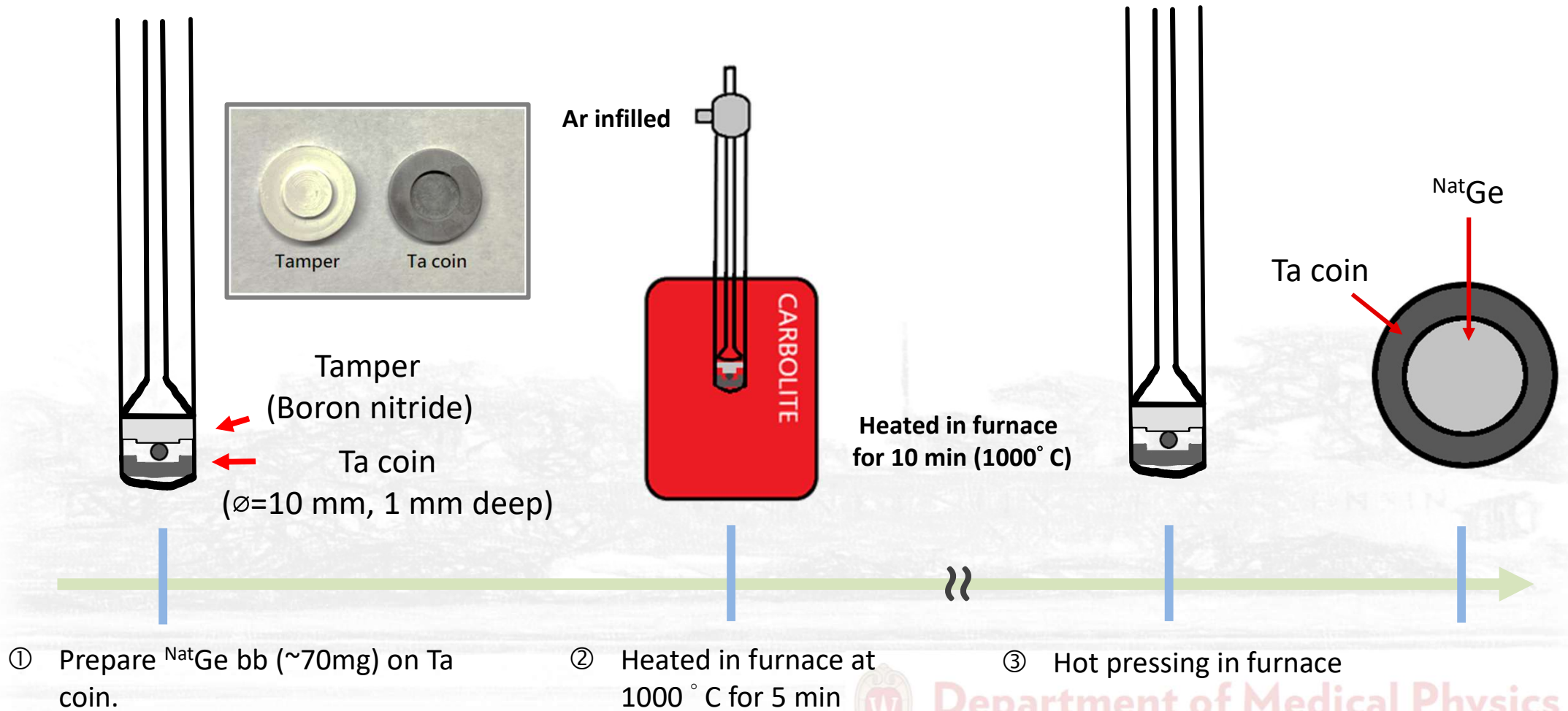


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Target making: Ge target production

Method



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Target making: Germanium bb making and Ge target production

Result

Ge bb making	Initial Ge powder mass (mg)	Ge Loss (mg)	Ge % Loss	
	~70	0.3 ± 0.27	0.5 ± 0.35	-

Ge-Ta target making	Initial Ge bb mass (mg)	Ge loss on target (mg)	Ge % Loss	Ge thickness (μm)
	~70	0.7 ± 0.73	0.9 ± 1.0	325 ± 44

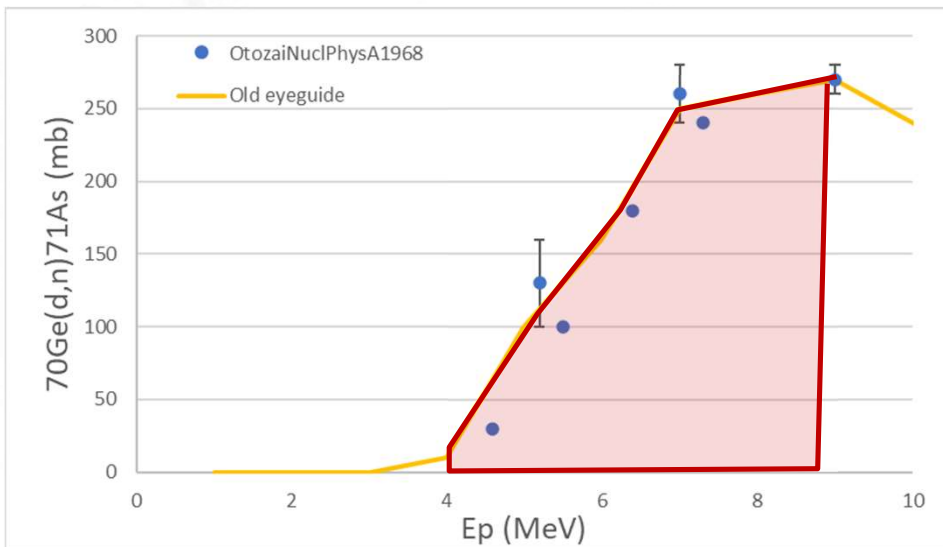


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Theoretical thickness: 166 μm

$$\text{Thickness} = \frac{\text{mass}}{\rho \times \text{area}} = \frac{0.07 \text{ g}}{5.35 \frac{\text{g}}{\text{cm}^3} \times 0.25\pi \text{cm}^2}$$



Department of Medical Physics

[5] K. Otozai, et al., Nucl. Phys. A. 107 (1968) 427–435.

[6] J.F. Ziegler et al., Nucl. Phys. B. 268 (2010) 1818–1823.

Bombardment: Irradiation study

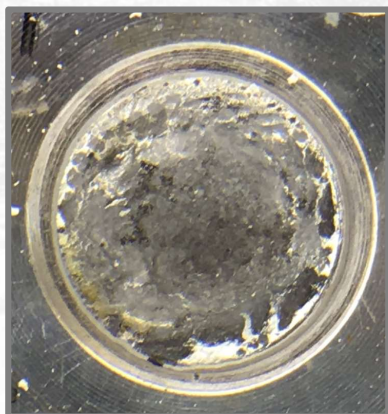
Method & Result

Irradiation condition		
Projectile	Proton	Deuteron
Energy	12 MeV	8 MeV
Current	30 μA	18.5 μA
Target	^{nat} Ge	

Targetry	Reaction	Energy (MeV)	Experiment (mCi/μAh)	Theory (mCi/μAh) [5,7]
Prototype coin-type target	⁷² Ge(p,n) ⁷² As	12	2.5	5.7
	⁷⁰ Ge(d,n) ⁷¹ As	8	0.17	0.17

*Enriched target yields extrapolated from experimental proton and deuteron irradiations of a ^{nat}Ge target

Before irradiation (proton)



After irradiation (proton)



Before irradiation (deuteron)



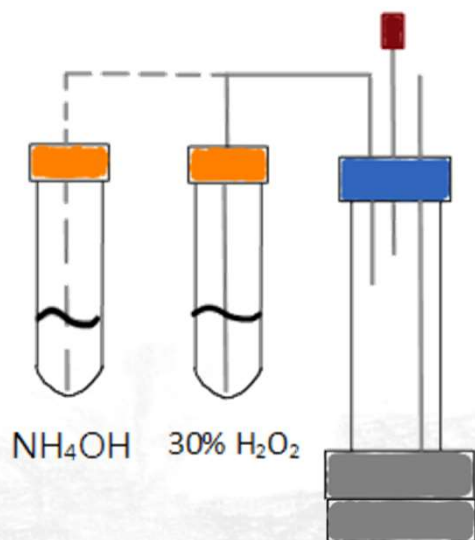
After irradiation (deuteron)



Isolation: Dissolution of Ge

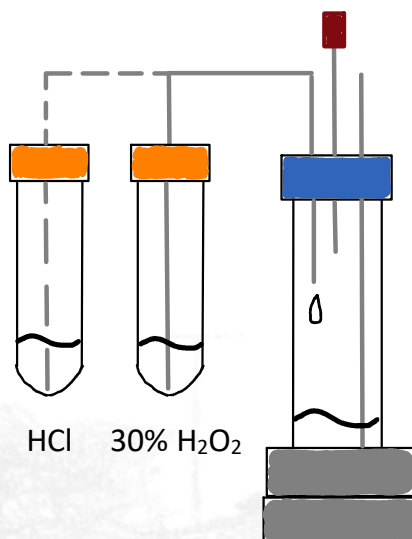
Method

NH_4OH method



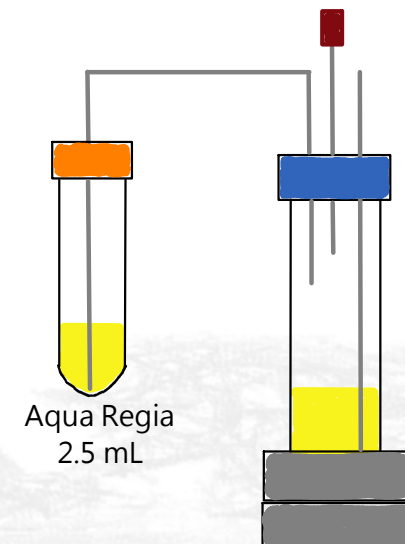
- ① Add 2 mL 0.48M NH_4OH in 30% H_2O_2
- ② Heated at 100 °C for 15 min.

HCl method



- ① Add 2 mL 37% HCl
- ② Heated at 100 °C for 15 min.
- ③ Dropwise addition of 2 ml of 30% H_2O_2

Aqua Regia method



- ① Add 2.5 mL aqua regia
- ② Heated at 100 °C for 30 min.



Isolation: Dissolution of Ge

Result

Condition	Ge-coated Ta coin		
Dissolution solution	0.48M NH ₄ OH [M] in 30% H ₂ O ₂ [7]	37% HCl with 30% H ₂ O ₂ [8]	Aqua regia [1]
Solution volume (mL)	2	2	2.5
Total reaction time (min)	15	15	30
Reaction Temp. (°C)	100	100	100
Ge mass dissolved (mg)	6.9 ± 0.7 (14±2%) n=20	20.7 ± 7.4 (25±10%) n=5	153± 12 (100%) n=6
Ta coin mass dissolved (mg)	1.1 ± 0.2 n=4	0.1± 0.1 n=3	0.0±0.1 n=3

Conclusion

- ◆ Aqua regia condition dissolved more mass of Ge than others under the same reaction time.
- ◆ No significant mass lost for Ta coin under aqua regia condition.

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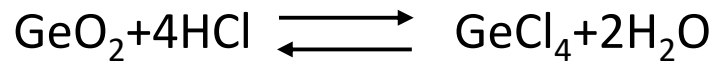
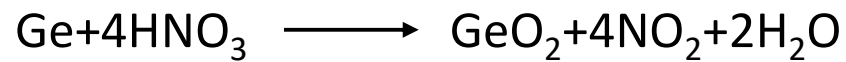
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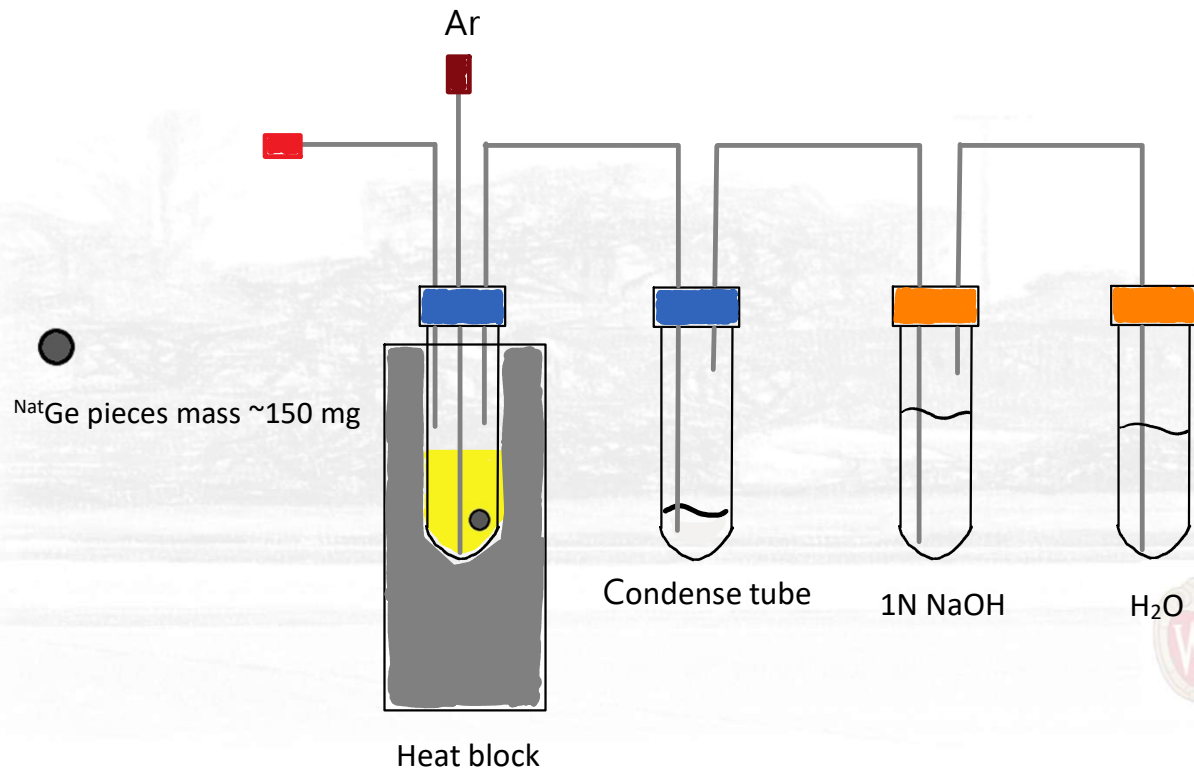
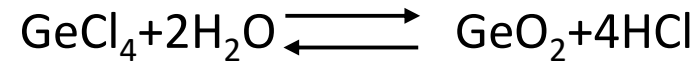
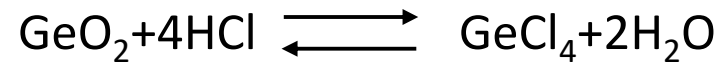
Reclamation: Dissolution and Distillation

Method

Dissolution:



Distillation and Hydrolysis:



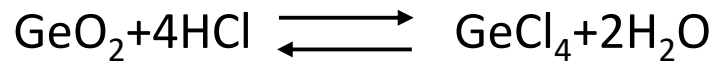
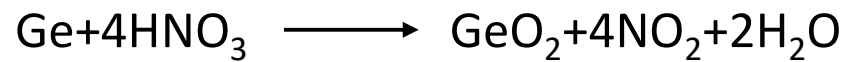
^{71/72}AsCl₅ would remain in heating tube during distillation



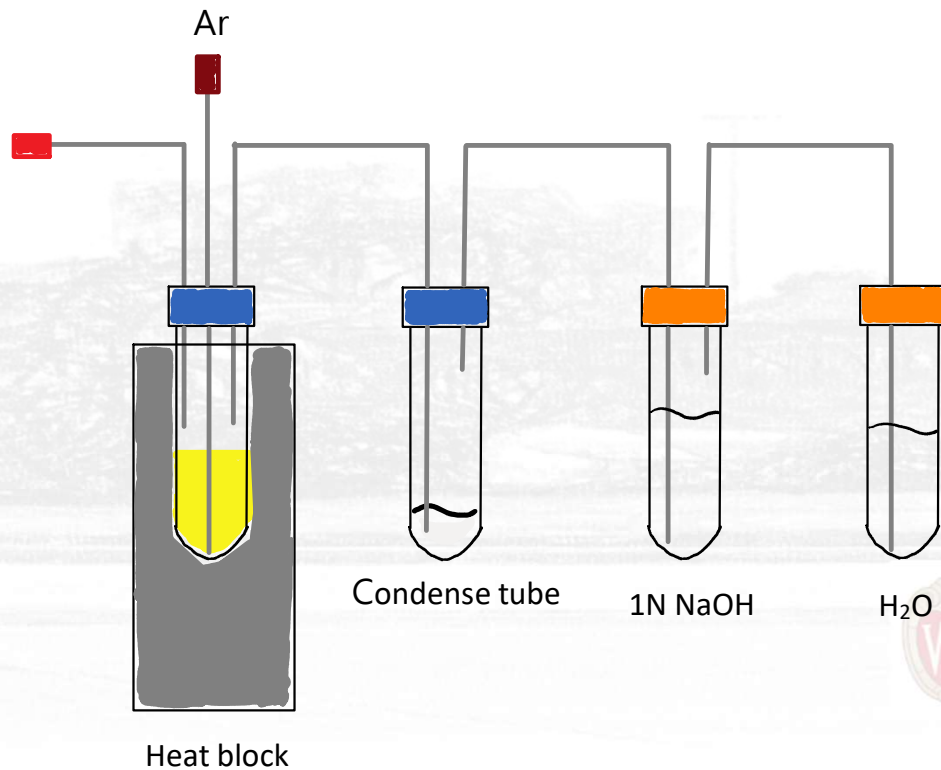
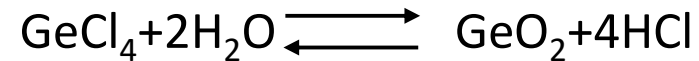
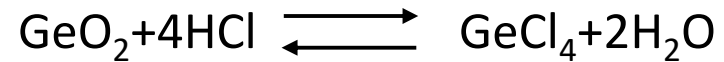
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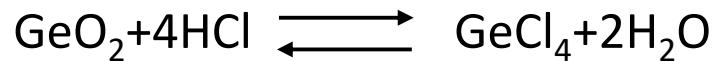
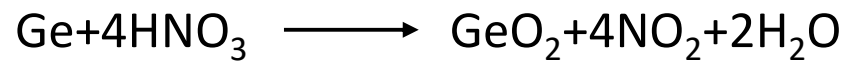
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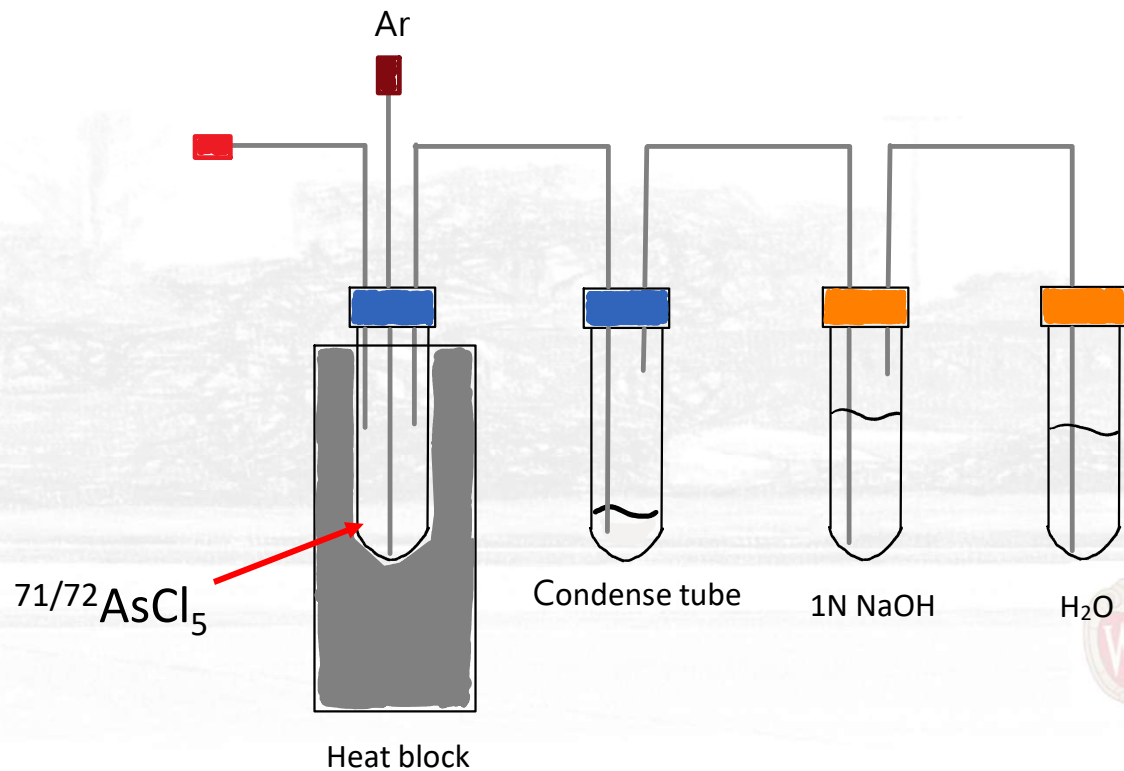
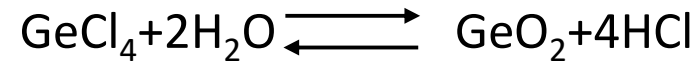
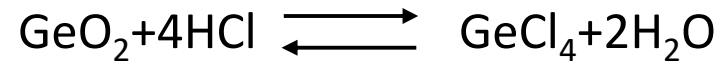
Reclamation: Dissolution and Distillation

Method

Dissolution:



Distillation and Hydrolysis:

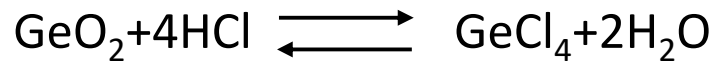
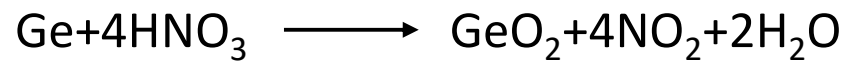


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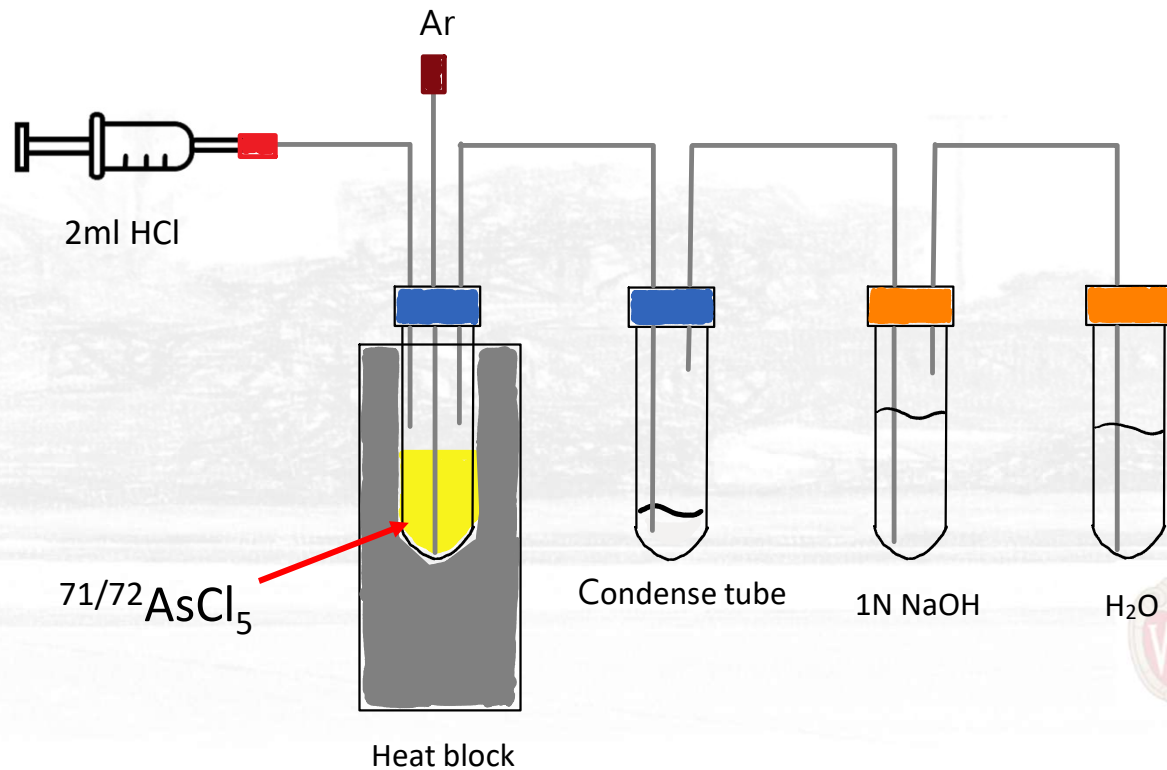
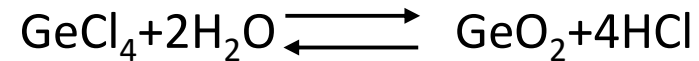
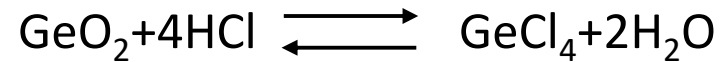
Reclamation: Dissolution and Distillation

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Dissolution:



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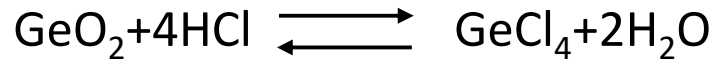
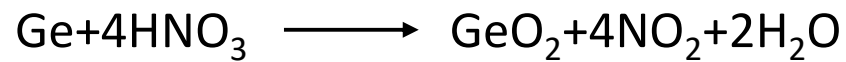
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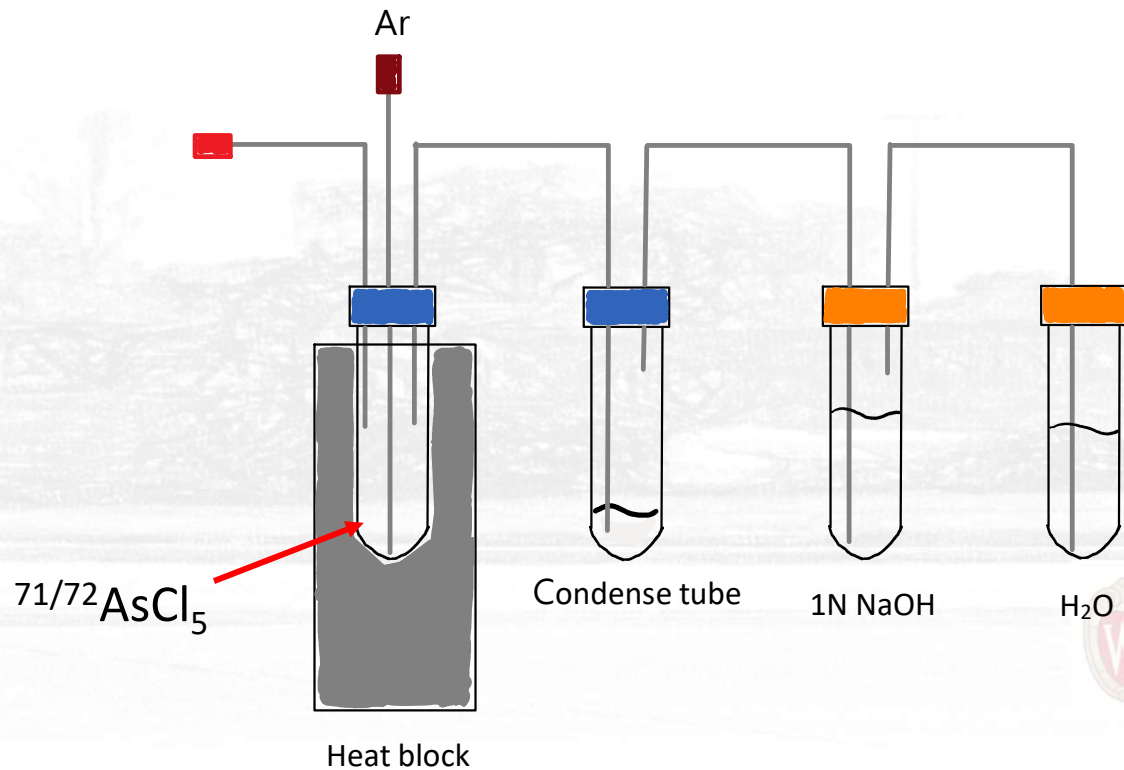
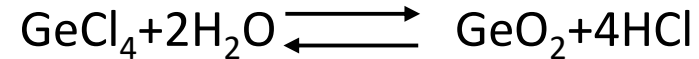
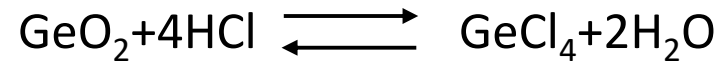
Reclamation: Dissolution and Distillation

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Distillation and Hydrolysis:



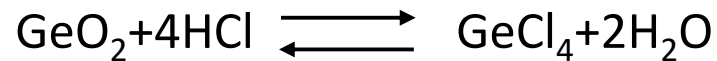
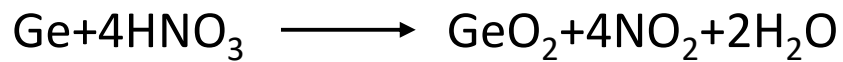
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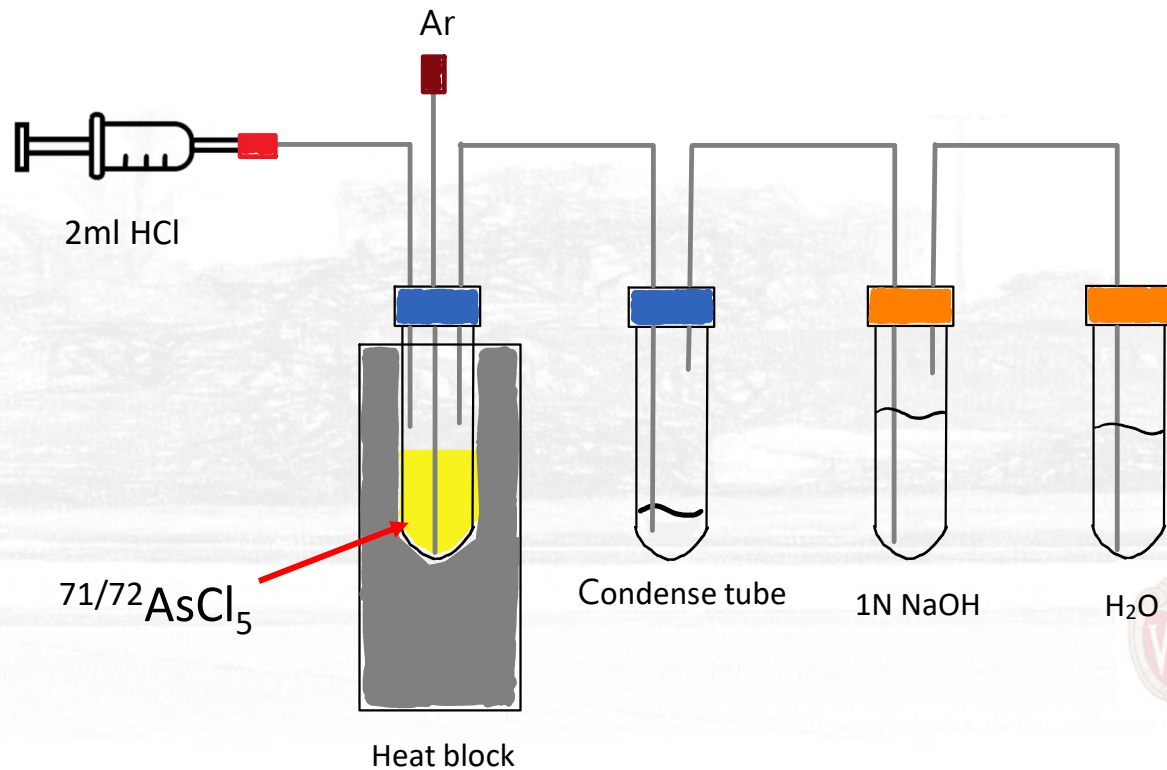
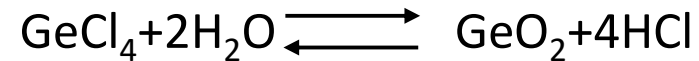
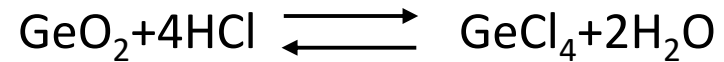
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Distillation and Hydrolysis:



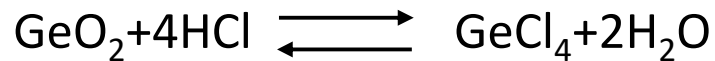
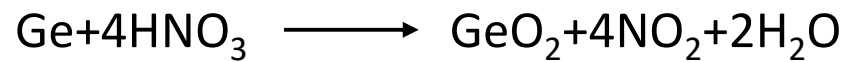
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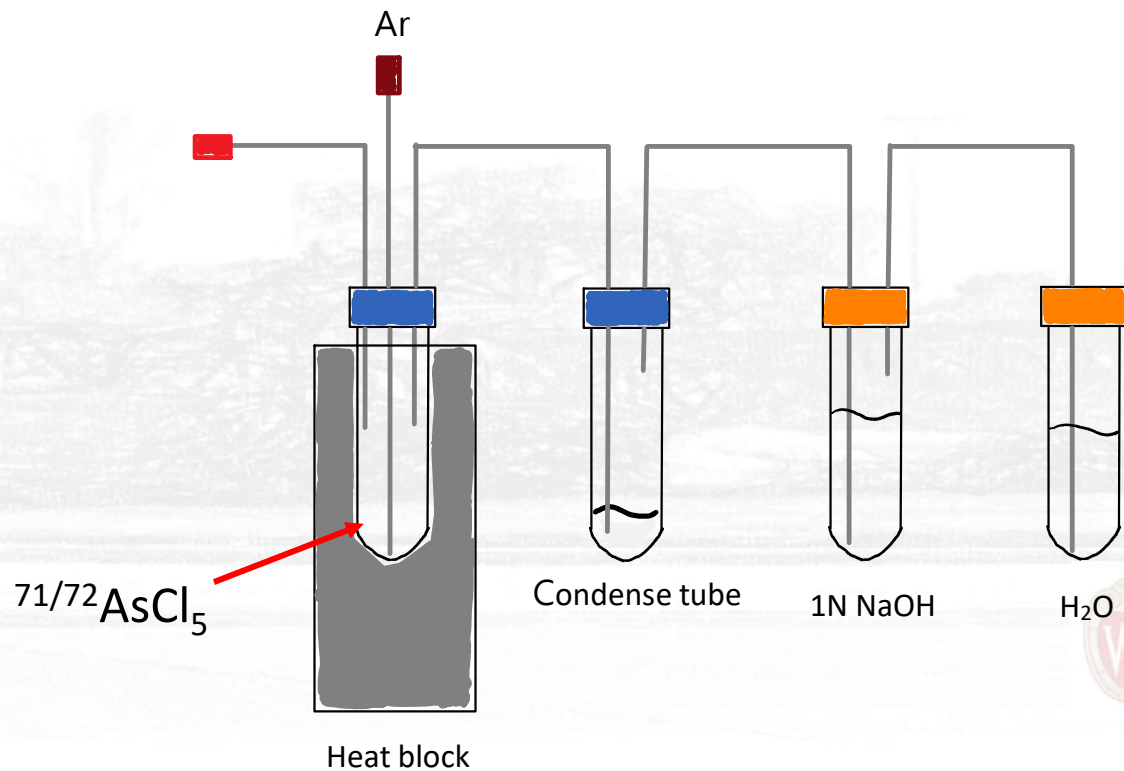
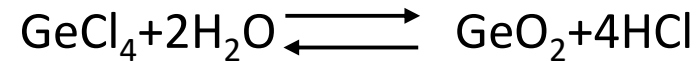
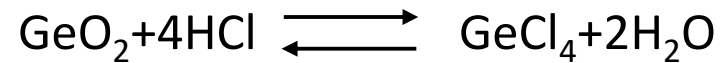
Reclamation: Dissolution and Distillation

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Dissolution:



Distillation and Hydrolysis:

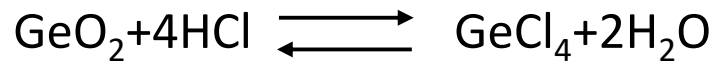
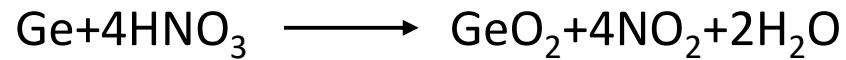


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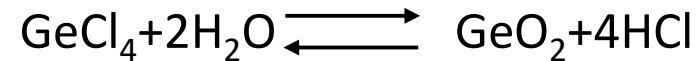
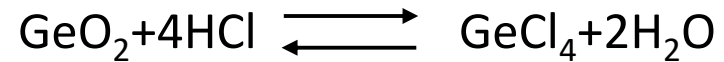
Reclamation: Dissolution and Distillation

Method

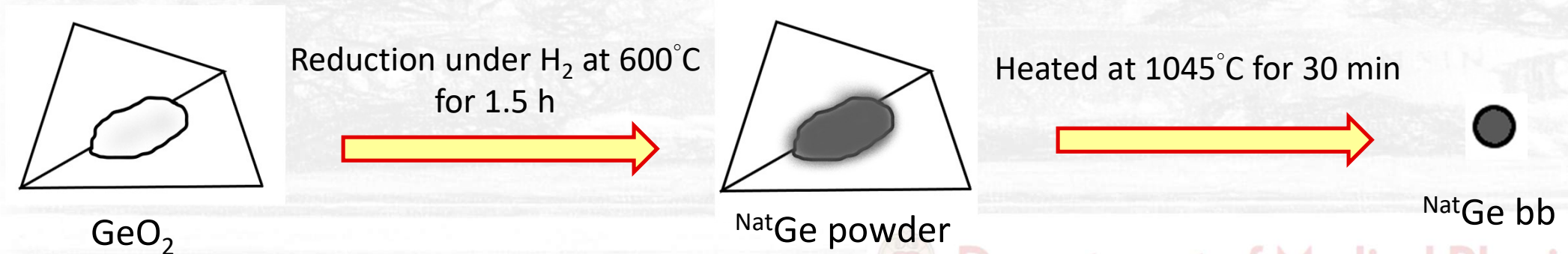
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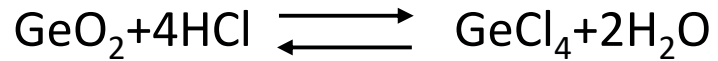
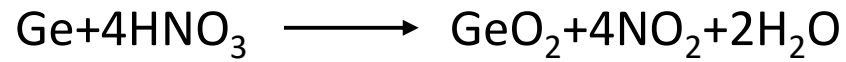
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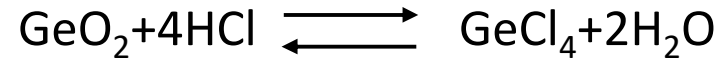
Reclamation: Dissolution and Distillation

Method

Dissolution:



Distillation:



Reclamation:

HCl capture method

NH₄OH neutralization method

NaOH neutralization method

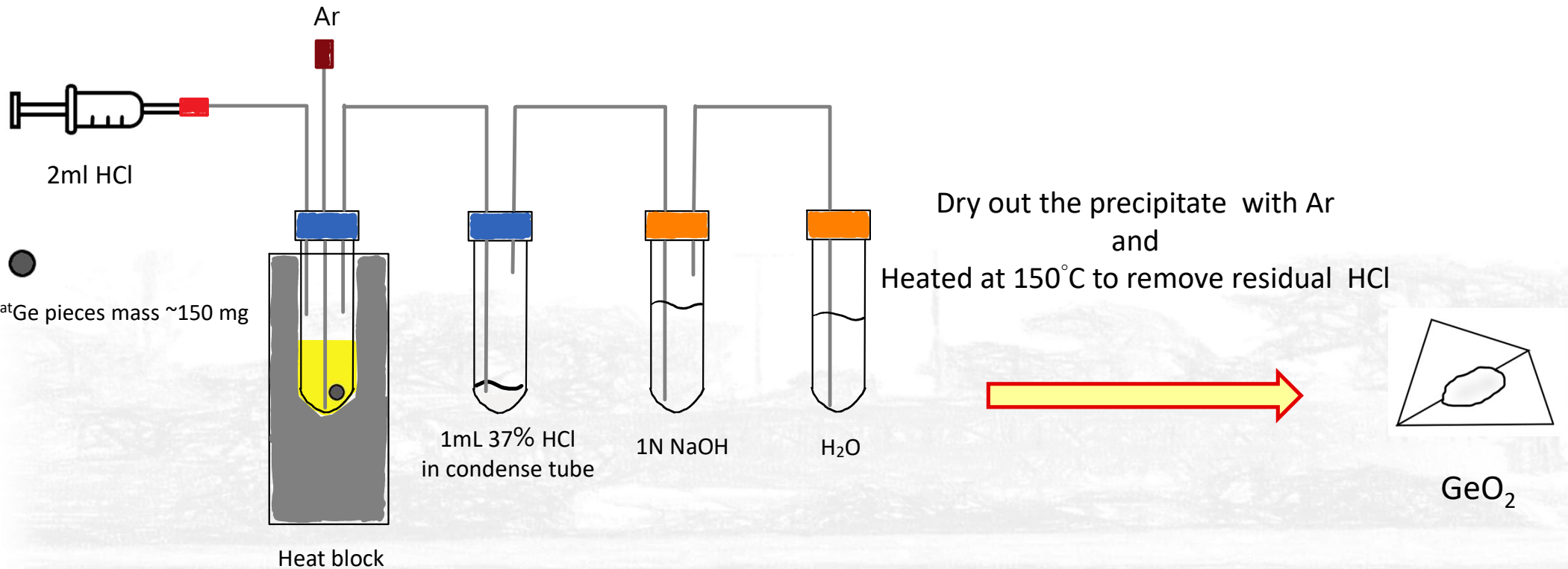


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Reclamation: Distillation and Reclamation

Method HCl capture method

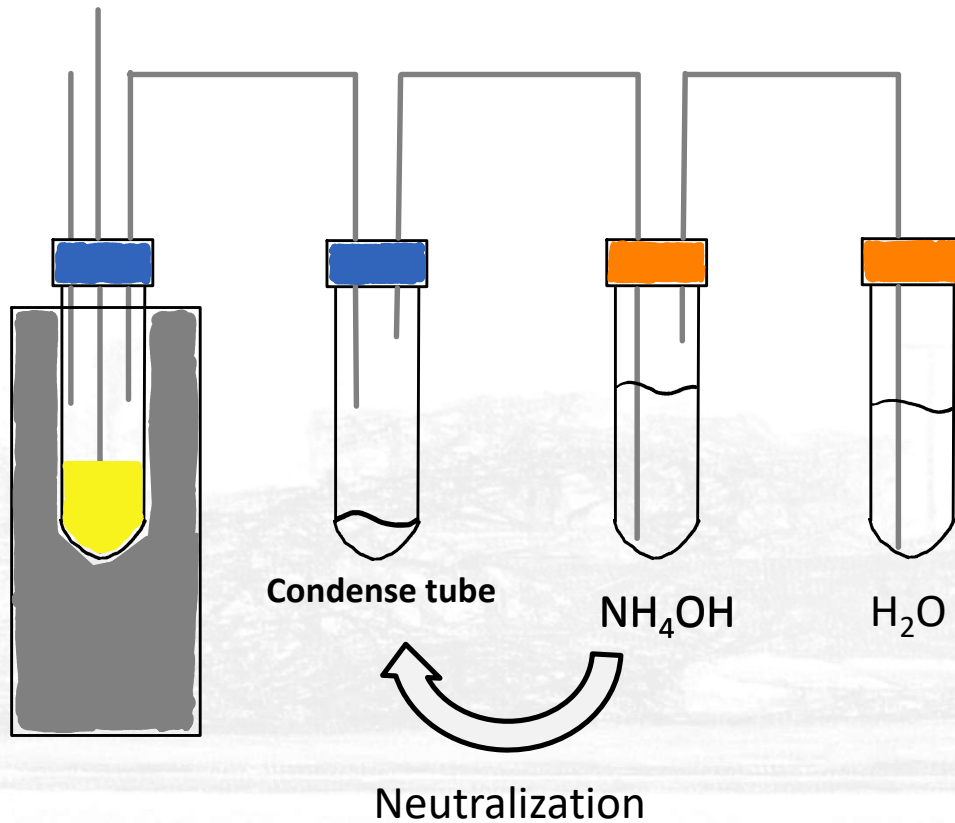


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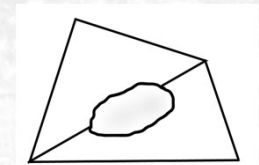
University of Wisconsin School of Medicine and Public Health

Reclamation: Distillation and Reclamation

Method NH_4OH neutralization method



Dry out the precipitate
and
Heated at $350^{\circ}C$ to remove salt



GeO_2



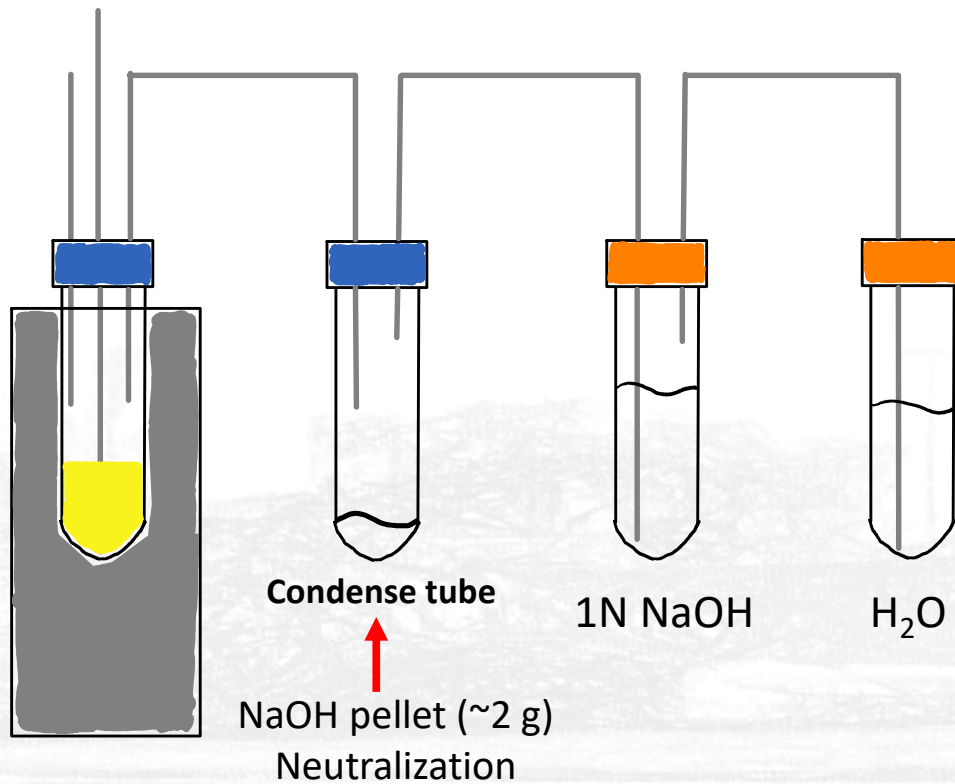
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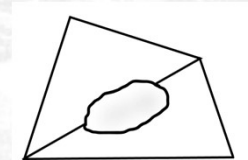
Reclamation: Distillation and Reclamation

Method

NaOH neutralization method



Wash with H₂O
and
Dry out the precipitate



GeO₂



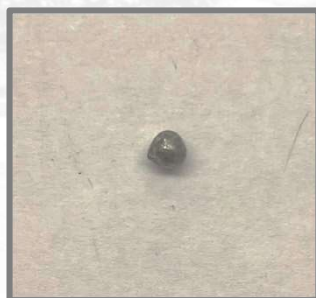
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Reclamation

Result

Condition	HCl capture	NH ₄ OH neutralization	NaOH neutralization
Ge piece mass (mg)	142.2	153.6	151.7
After removing residual HCl or salt			
^{Nat} GeO ₂ mass (mg)	103.5	86.4	53.8
After reduction by H ₂			
Ge mass	67.5	55.3	19.6
After bb making			
Ge bb mass	68.7	58.6	14.8
Reclamation efficiency %	48.3%	38.2%	9.8%



Conclusion

Target making

Reclamation

- Target production: Improve the target production procedure to get fully covered Ge target.
- Reclamation: HCl capture method. Optimization for the procedure to increase the efficiency is needed.

Bombardment

- Target cooling problem: The Ge is melted during irradiation. We will use thinner Ta coin and try to make the thickness of Ge thinner.

Isolation

- Separation of As from Ge: Aqua regia method displayed higher Ge mass dissolved and shorter dissolving time.

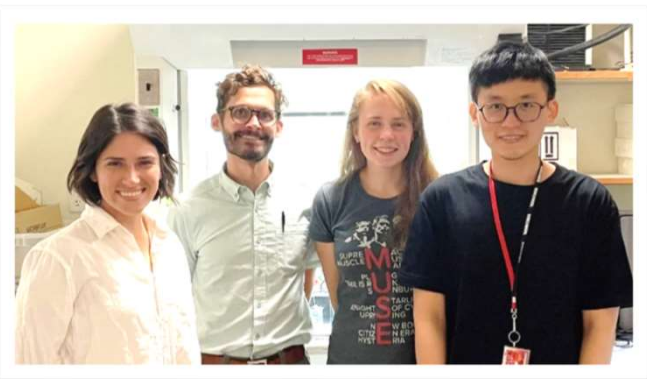


Acknowledgement

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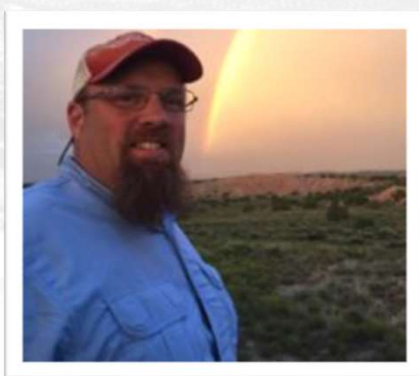
Paul Ellison's Group



Cyclotron Group

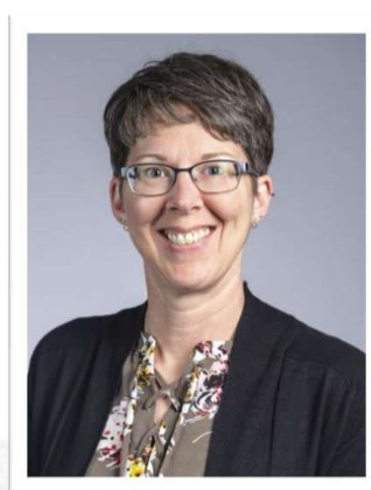


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Thanks for your attention



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Bombardment: Irradiation study

Method & Result

Irradiation condition		
Projectile	Proton	Deuteron
Energy	12 MeV	8 MeV
Current	30μA	18.5 μA
Target	^{nat} Ge	

Targetry	Reaction	Energy (MeV)	Experiment (mCi/μAh)	Theory (mCi/μAh) [1,4]
Prototype coin-type target	⁷² Ge(p,n) ⁷² As	12	2.5	5.7
	⁷⁰ Ge(d,n) ⁷¹ As	8	0.17	0.17

*Enriched target yields extrapolated from experimental proton and deuteron irradiations of a ^{nat}Ge target

Energy (MeV)	Thickness (μm)	Cross section (bar)	Activity
8	0.036	0.26	1.6×10 ⁵
.	.	.	.
.	.	.	.
.	.	.	.
4	0.024	0.01	4.2×10 ³

Sum up!

$$A(\text{Bq}) = \rho \left(\frac{\text{g}}{\text{cm}^3} \right) \times x(\text{cm}) \times \sigma(\text{cm}^2) \times \frac{1}{A_w \left(\frac{\text{g}}{\text{mol}} \right)} \times A_N \times I \left(\frac{\text{particles}}{\text{s}} \right) \times (1 - e^{-\lambda t})$$

ρ = density

x = thickness

σ = cross section

A_w = atomic weight

A_N = Avogadro's

I = current intensity

t = irradiation time



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Conclusion

Target making

Reclamation

- Target production
Improve the target production procedure to get fully covered Ge target.
- Reclamation
HCl capture method.
Optimization for the procedure to increase the efficiency is needed.

Bombardment

Isolation

Metal, Metallic Element or Alloy	Temperature		Thermal Conductivity - k - (W/m K) (Btu/(ft h °F))
	- t -		
	(°C)		
	(K)	(°F)	
Tantalum	-73		57.5
"	0		57.4
"	127		57.8
"	327		58.9
"	527		59.4
"	727		60.2
"	927		61
Titanium	-73		24.5
"	0		22.4
"	127		20.4
"	327		19.4
"	527		19.7
"	727		20.7
"	927		22

Research topic

^{77}As beta emitter $^{77}\text{Ge}(n,\gamma)^{77}\text{As}$

- End point energy ≈ 683 keV
- $T_{1/2} = 38.8$ h
- Problem...

The lack of imaging property which can provide in patient selection, target verification and dosimetry.

$^{71/72}\text{As}$ positron emitter $^{72}\text{Ge}(p,n)^{72}\text{As}$ --UWisc PETtrace cyclotron

- $T_{1/2}$: $^{71}\text{As} = 65.3$ h and $^{72}\text{As} = 26$ h
- End point energy: $^{71}\text{As} = 816$ keV and $^{72}\text{As} = 2500$ keV

^{119}Sb MAe⁻ emitter

- Homologous relationship with ^{119}Sb

