

Tuning Pharmacokinetic Properties of Sulfur-Rich Macrocyclic Chelators for Complexation of Theranostic Meitner-Auger Electron Emitter ^{197m}gHg

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Introduction

- The rapidly growing field of radiopharmaceuticals for treatment of cancer contains two main categories: therapeutics and diagnostics. **Therapeutic** agents incorporate radionuclides with decays that damage cancer cells while **diagnostic** agents make use of decay modes compatible with imaging (see **Figure 1**). The combination of these techniques into one pharmaceutical has been coined **theranostics**.

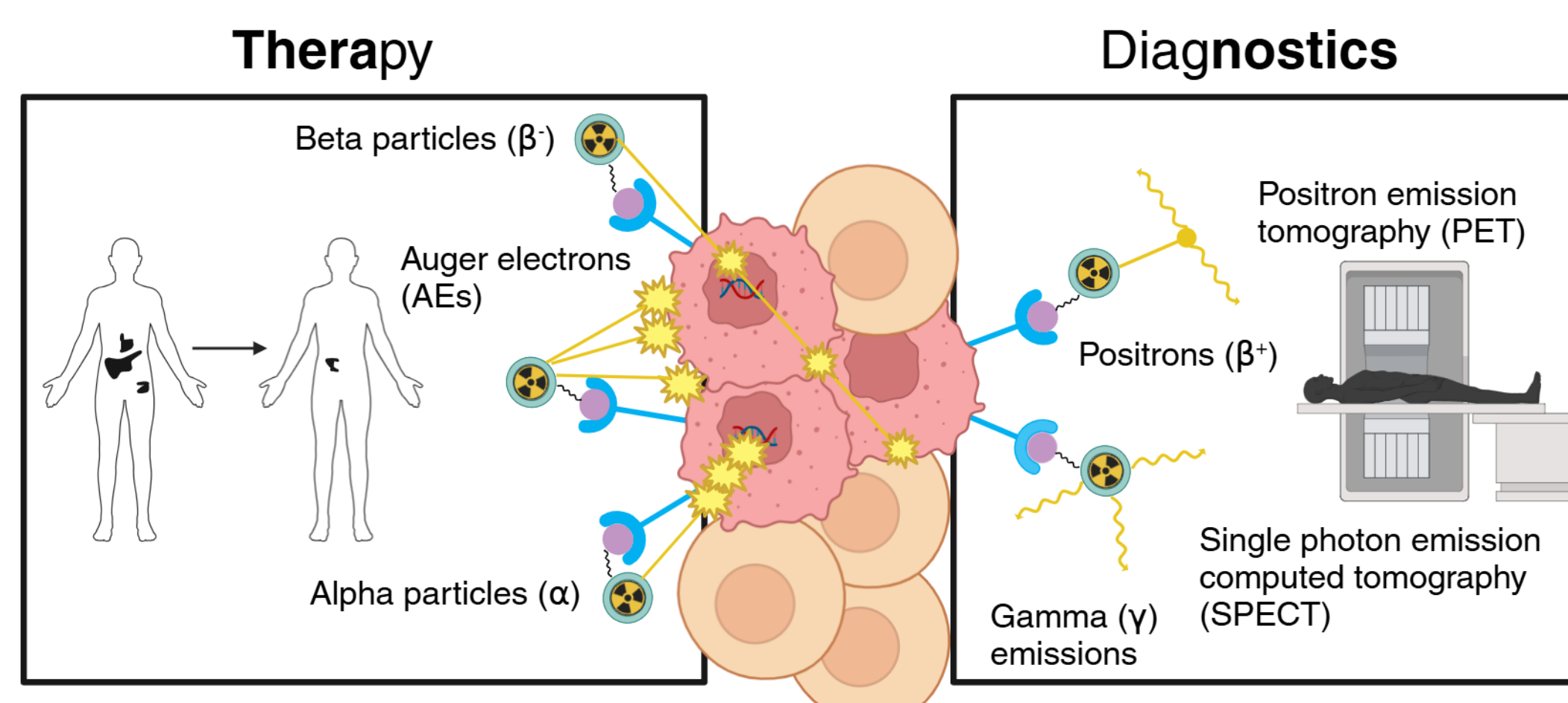


Figure 1. Types of radioactive decays utilized in therapeutic agents and diagnostic agents respectively.

- Mercury-197m/g** ^{197m}Hg ($t_{1/2} = 23.8$ h, IT (91%)) and ^{197g}Hg ($t_{1/2} = 64.14$ h, EC (100 %)) has only recently been investigated for use in **theranostic radiopharmaceuticals** and presents a promising pair of radioactive isomers to be used in a chemically matched theranostic drug, exploiting the SPECT-compatible gamma ray emissions and cytotoxic Meitner-Auger electrons.^[1]

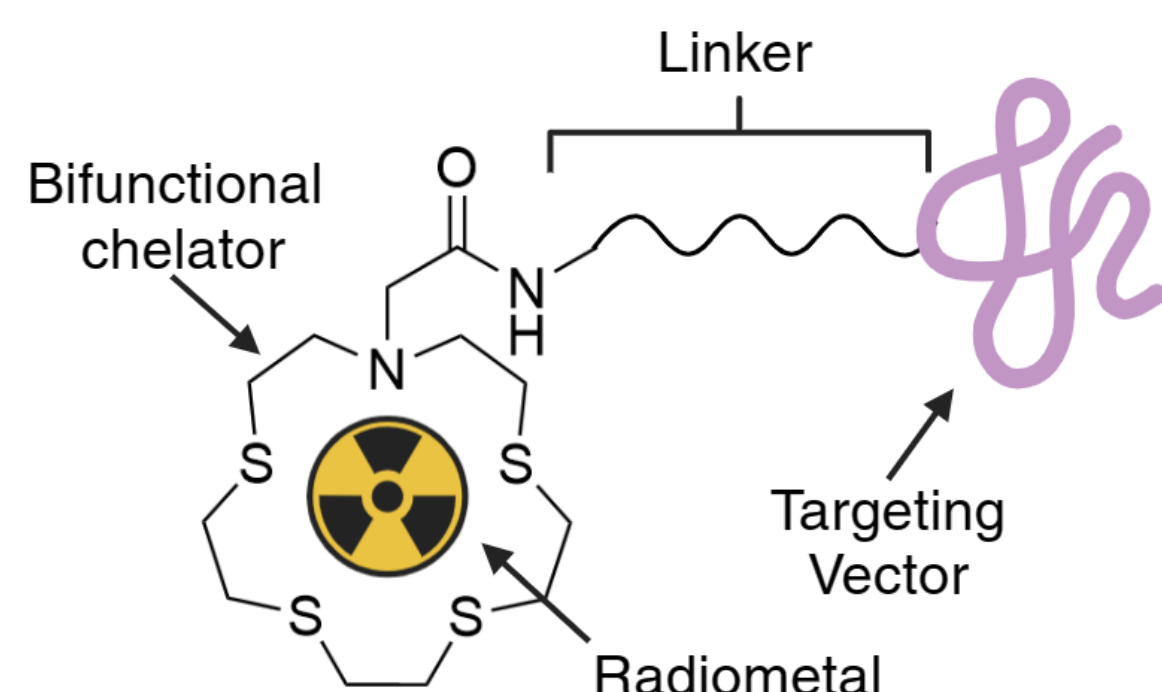


Figure 2. Structure of an inorganic radiopharmaceutical incorporating the NS₄ chelator backbone.

- Our group has previously developed a suitable chelator backbone, NS₄ (see **Figure 2**), that **selectively** forms **stable** ^{197m}gHg complexes for radiolabeling purposes.
- While selective and stable, bifunctional derivatives of this chelator are highly lipophilic, resulting in solubility issues and high liver uptake during *in vivo* studies.^[1]

Hypothesis: lipophilicity of NS₄ chelator can be tuned with polar linkers while providing position for conjugation of a targeting vector.

Novel Bifunctional Chelators

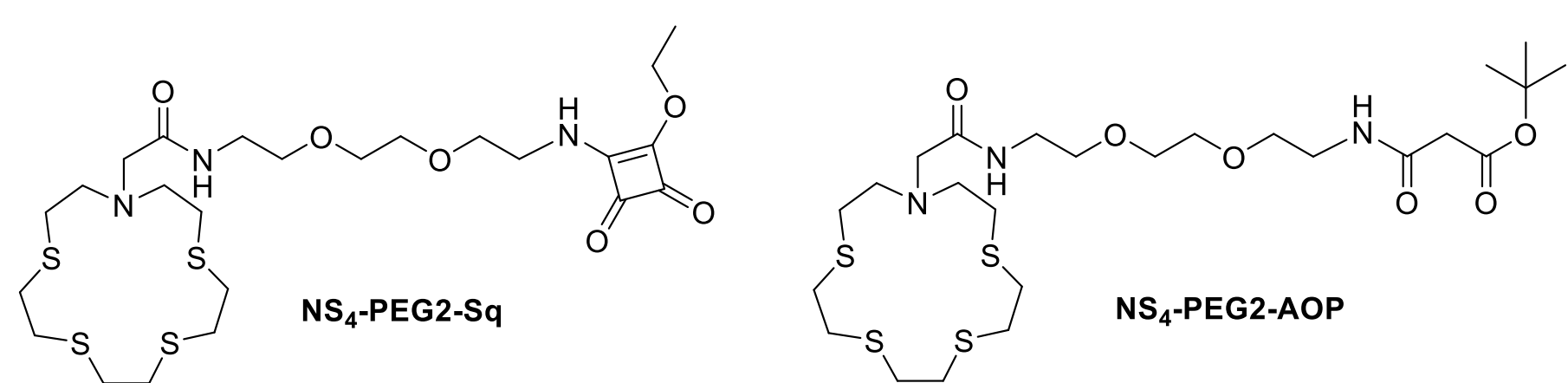


Figure 3. Chemical structures NS₄ chelators bearing first generation linker modifications.

- Two novel bifunctional chelators (BFCs) (**Figure 3**) were designed by the integration of a polyethylene glycol (PEG2) linker chain and two bifunctional handles for the targeting vector.
- The PEG2 chain, squarmino ester (Sq) and protected amino oxopropanoic acid (AOP) are all highly polar.
- The Sq and AOP groups can both be easily deprotected and coupled to exposed lysine residues on a targeting vector.^[2]

Novel BFC Synthesis

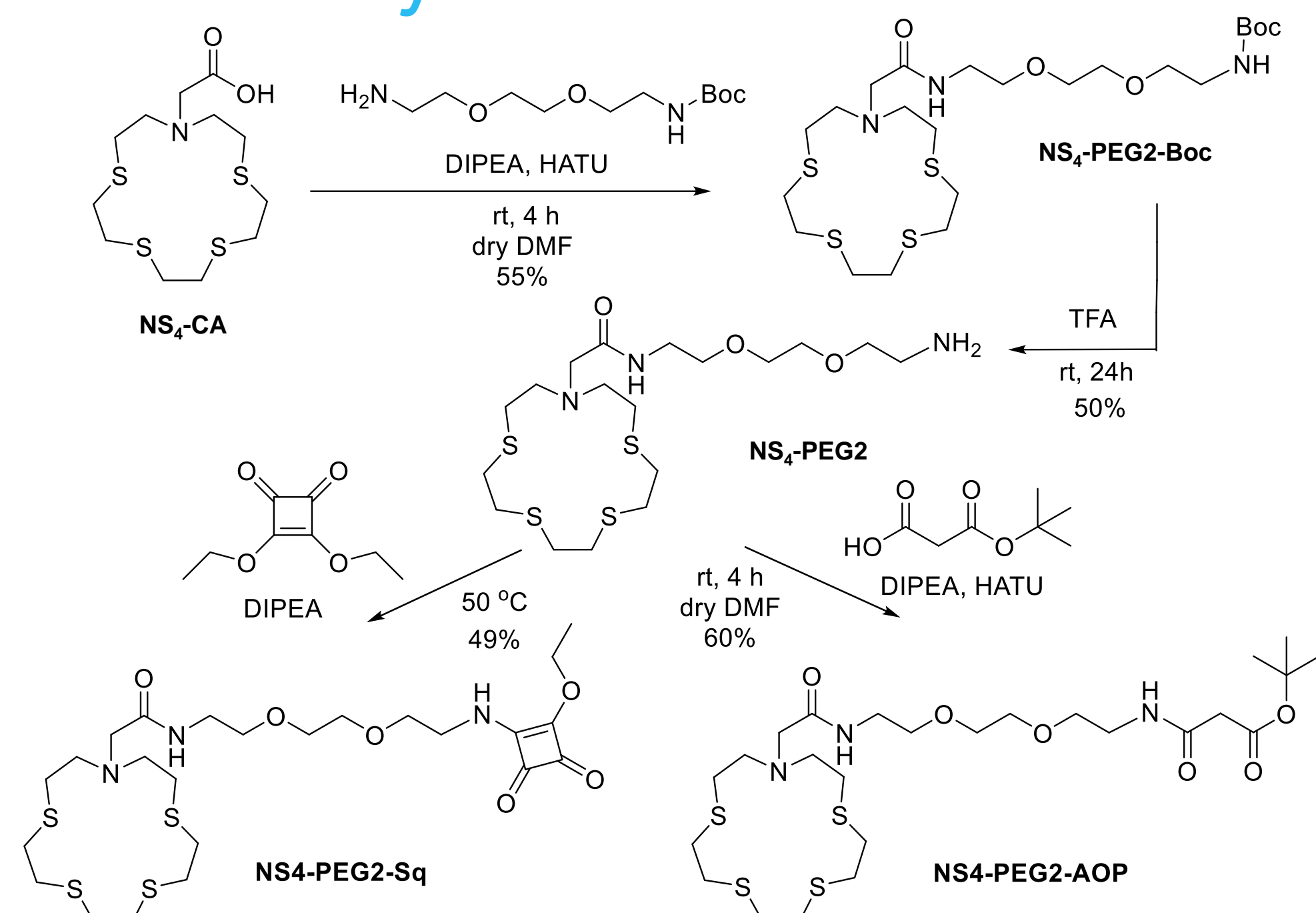


Figure 4. Synthetic route for NS₄-PEG2-(Sq/AOP).^[2]

- Both NS₄-PEG2-(Sq/AOP) have been successfully synthesized and characterized.

Non-Radioactive Mercury Study

- Noting the shared methyl acetamide (MA) moiety included in all proposed BFCs, a representative compound, NS₄-MA, was synthesized and characterized.
- A complexation study with non-radioactive Hg was then conducted to investigate the ability of the chelator to complex Hg.

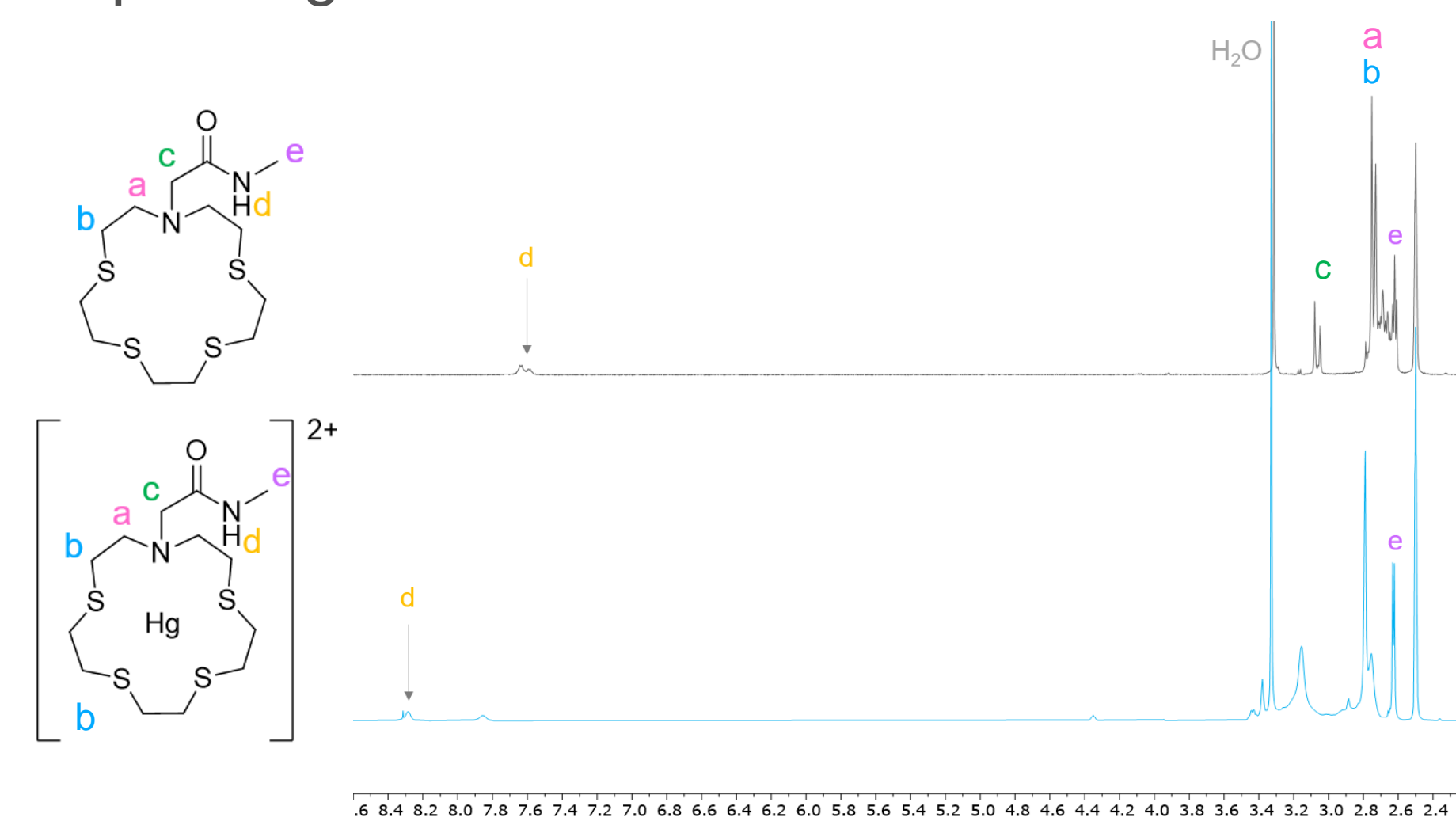


Figure 5. ¹H NMR (DMSO-*d*₆, 500 MHz) of NS₄-MA and [Hg(NS₄-MA)]²⁺.

- Shifting of amide proton (**d**) while methyl protons (**e**) remain unchanged is indicative of the carbonyl oxygen acting as a donor atom for Hg complexation; further corroboration (via DFT or crystallography) is required.

Future Work

- Conduct ^{197m}gHg radiolabeling studies with both BFCs, evaluate the stability of the formed complexes via incubation in biologically relevant media (i.e., glutathione and human serum) and determine the logD_{7.4} of radiometal-complexes to assess the lipophilicity.

- In vivo* animal studies to evaluate impact on biodistribution.

Acknowledgments

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References

- Randhawa, P.; et al. *Chem. Eur. J.* **2023**, *29* (21).
- Rudd, S.; et al. *Chem. Commun.* **2016**, *52* (80), 11889.
- Tosato, M.; et al. *Molecules* **2022**, *27* (13), 4158.