



Potential Moderator Materials for Very Cold Neutron Sources

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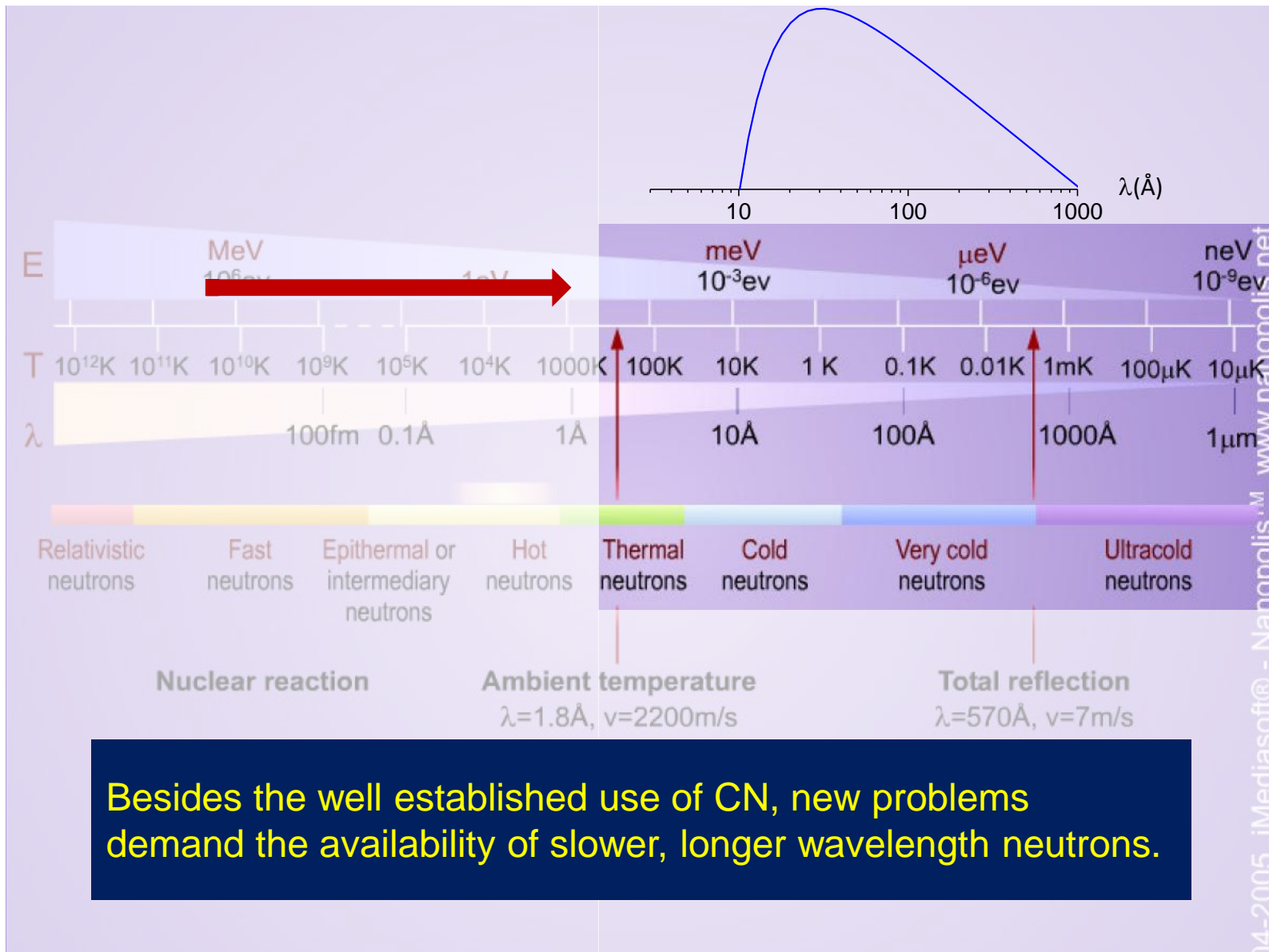


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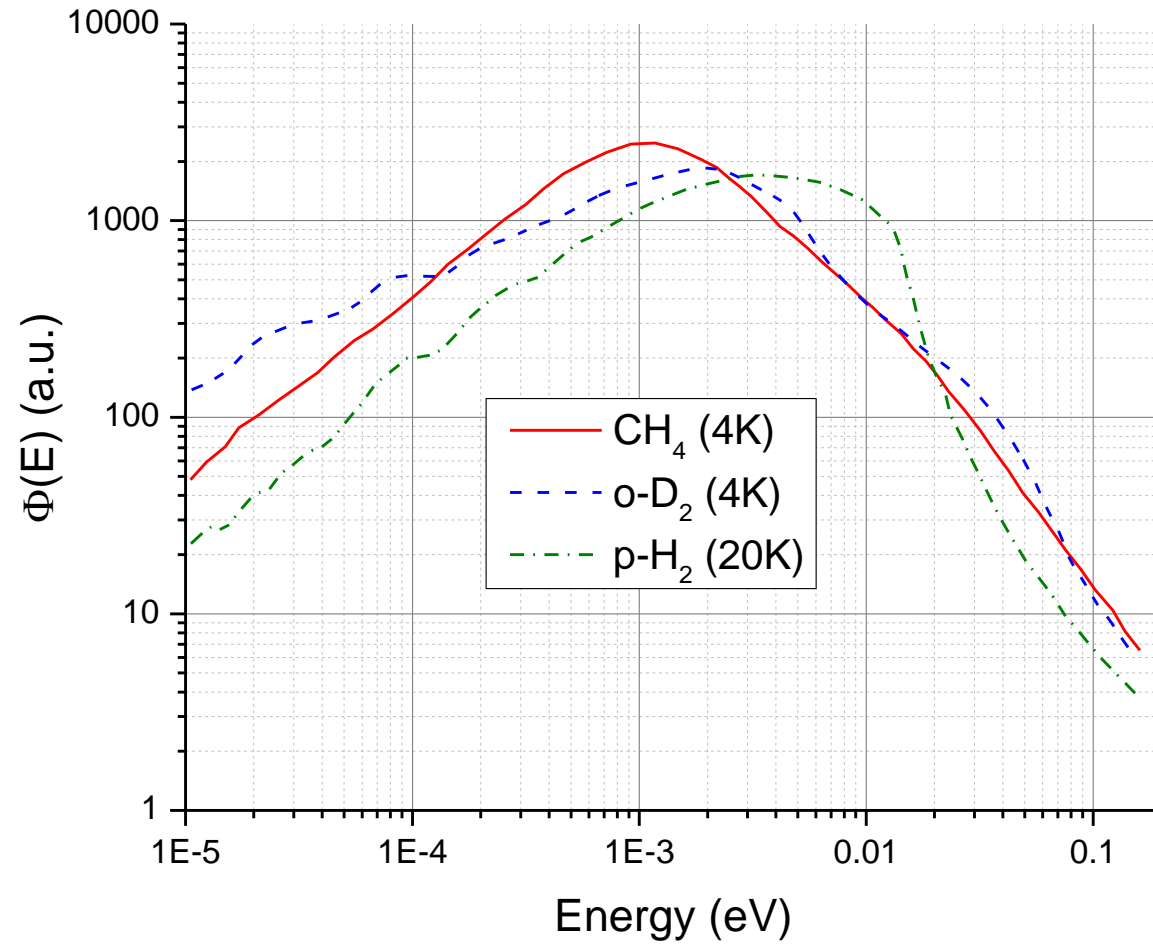
MOTIVATIONS

- ✓ The increasing demand for **long wavelength neutrons** is being motivated by the need to study larger objects and slower motions.
- ✓ Many modern neutron scattering techniques will benefit from **more intense VCN fluxes** if they could be realized.
- ✓ Results of a preliminary study aimed at identifying materials that look promising as **potential moderators for VCN** sources.
- ✓ Explore systems that present significant contributions of low-energy modes in their generalized frequency spectrum, and that implied searching for systems with **very low rotational transitions**.



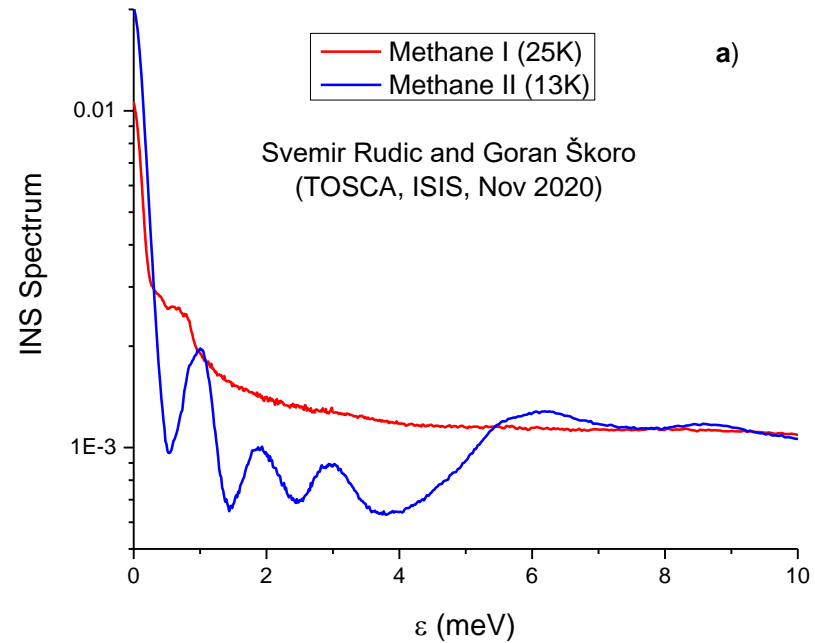
Besides the well established use of CN, new problems demand the availability of slower, longer wavelength neutrons.

H: 50x52x88 mm³
D: 140x140x200 mm³



Adapted from F. X. Gallmeier et al 2018 J. Phys.: Conf. Ser. 1021 012083

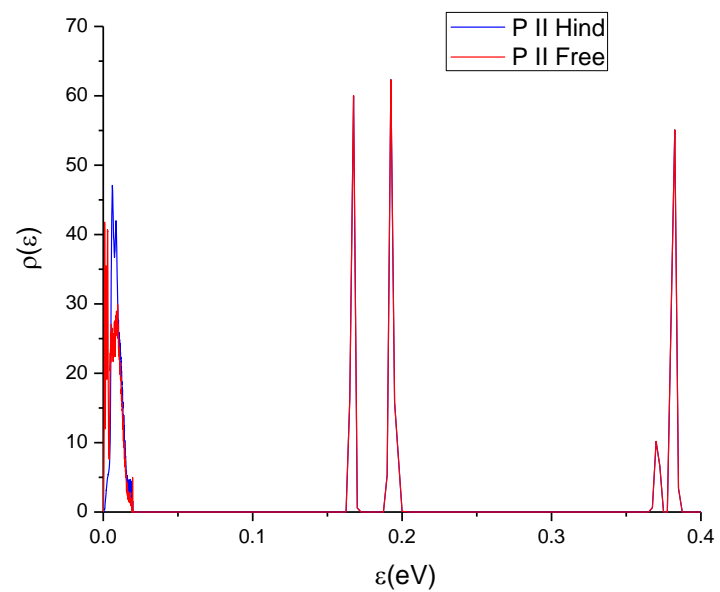
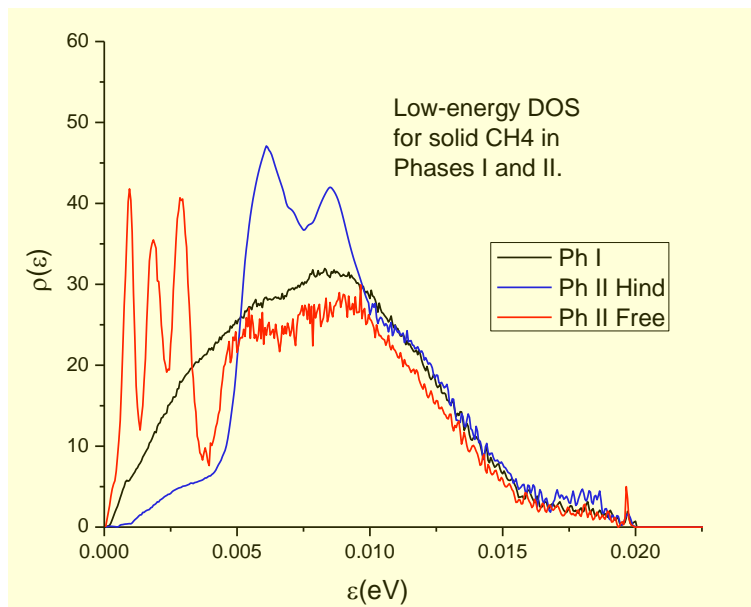
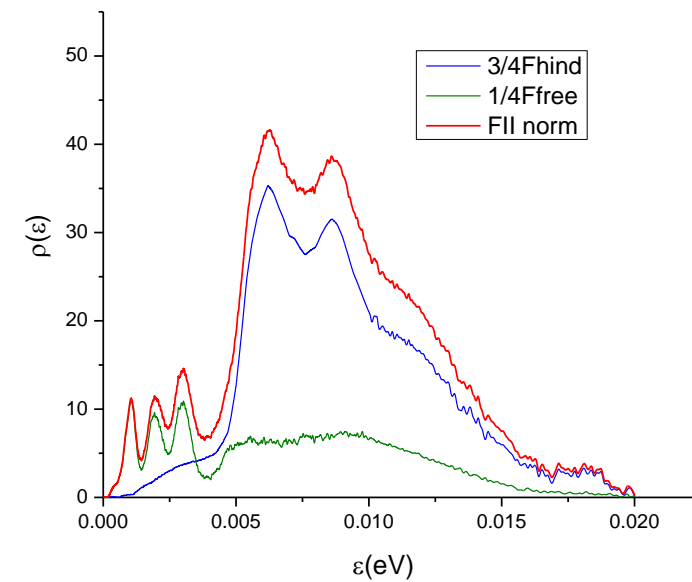
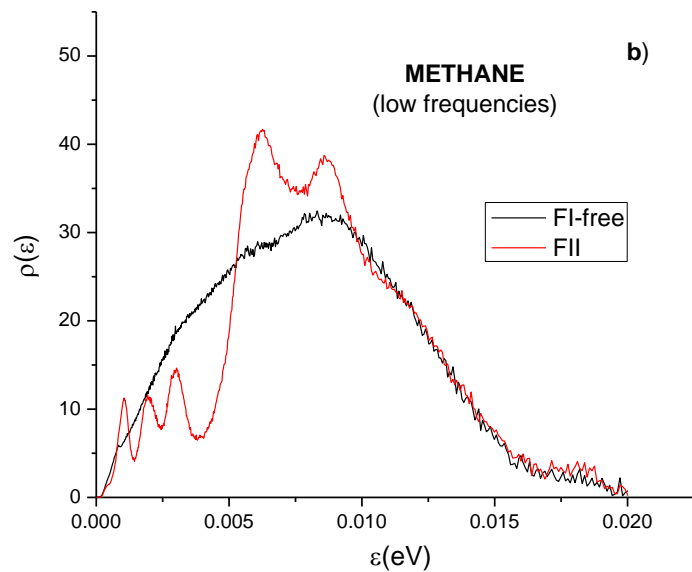
METHANE II



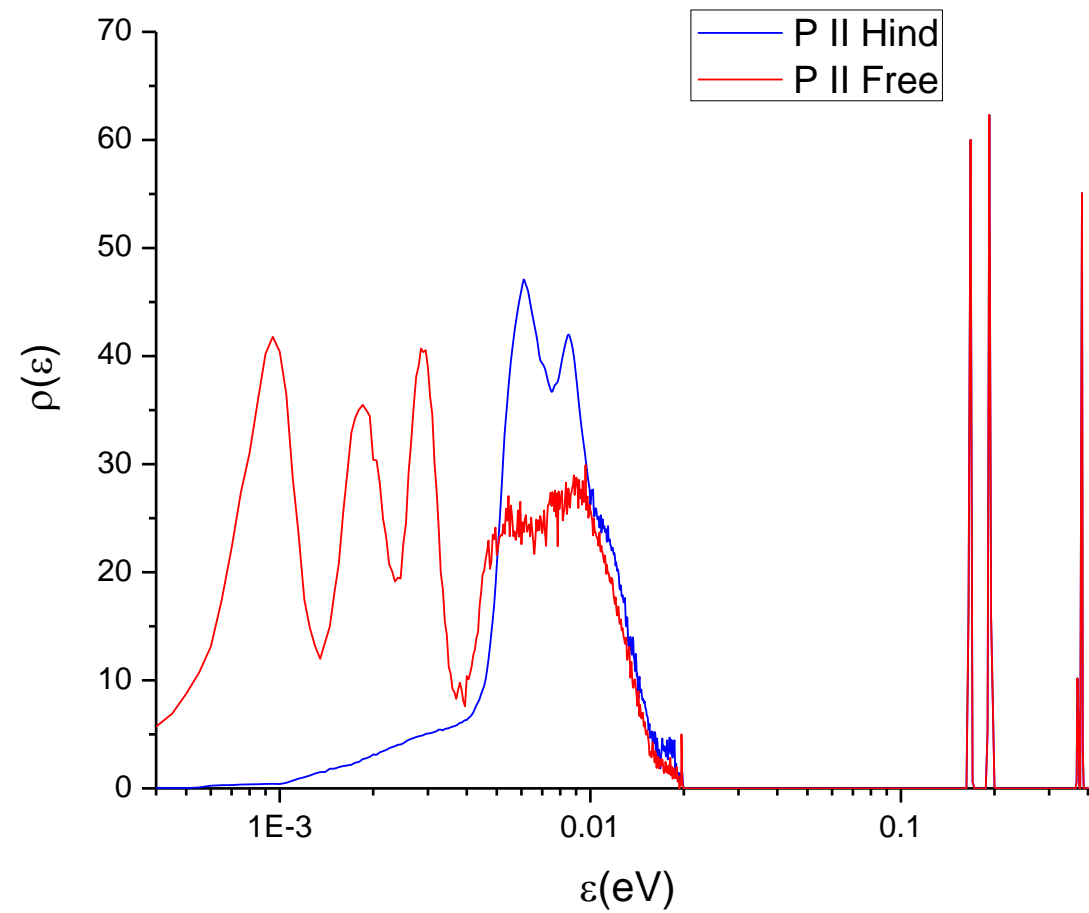
The measured INS spectra of solid Methane I (25K) and II (13K) are shown over a small energy transfer range.

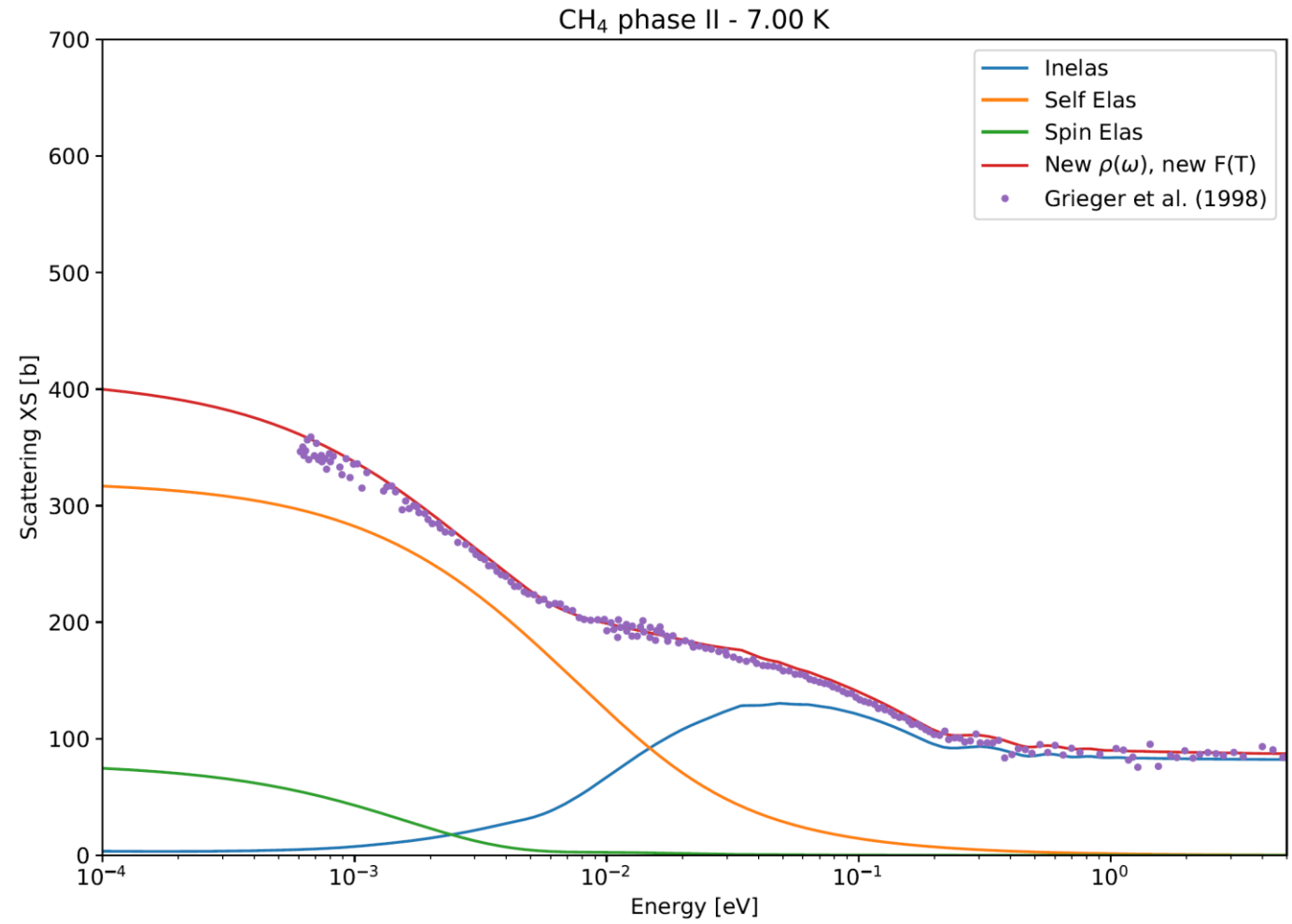
$$\left(\frac{d^2 \sigma}{d\Omega dE'} \right)_{inc+1ph} = N \frac{\sigma_{inc}}{4\pi} \frac{k}{k_0} \frac{Q^2}{4M} \exp(-2W) \frac{Z(\omega)}{\omega} [n(\omega) + 1]$$

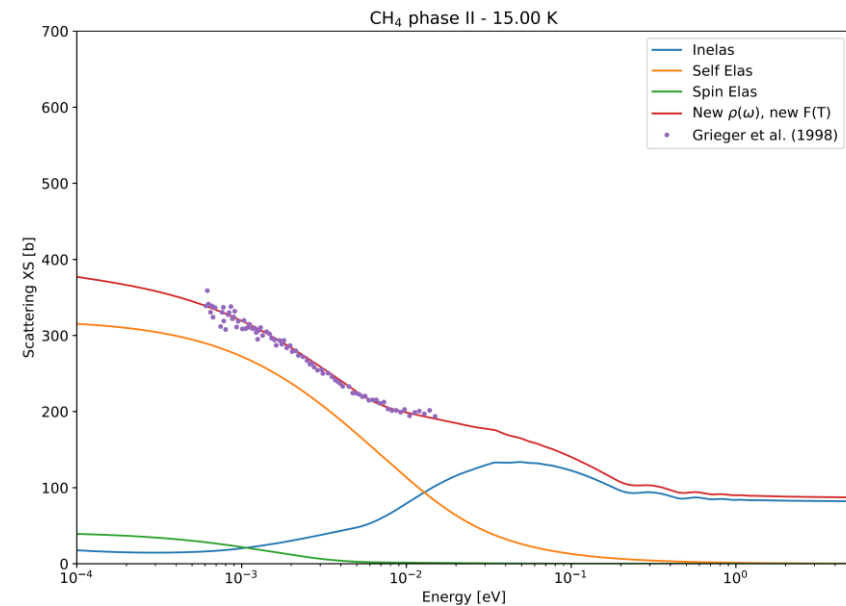
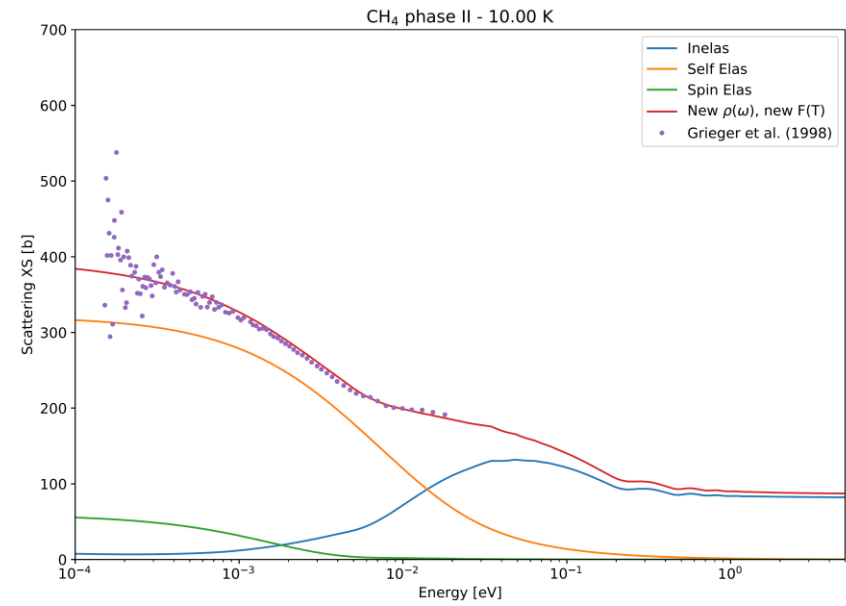
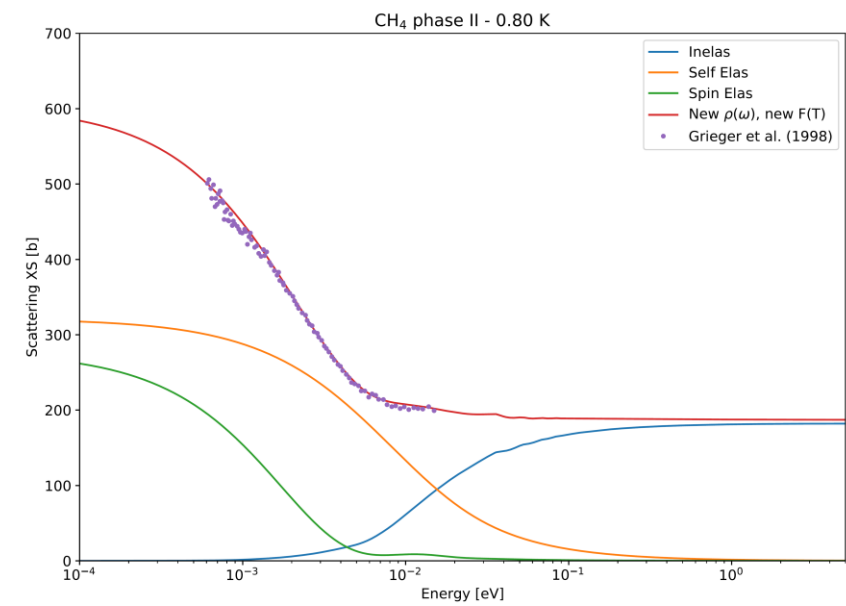
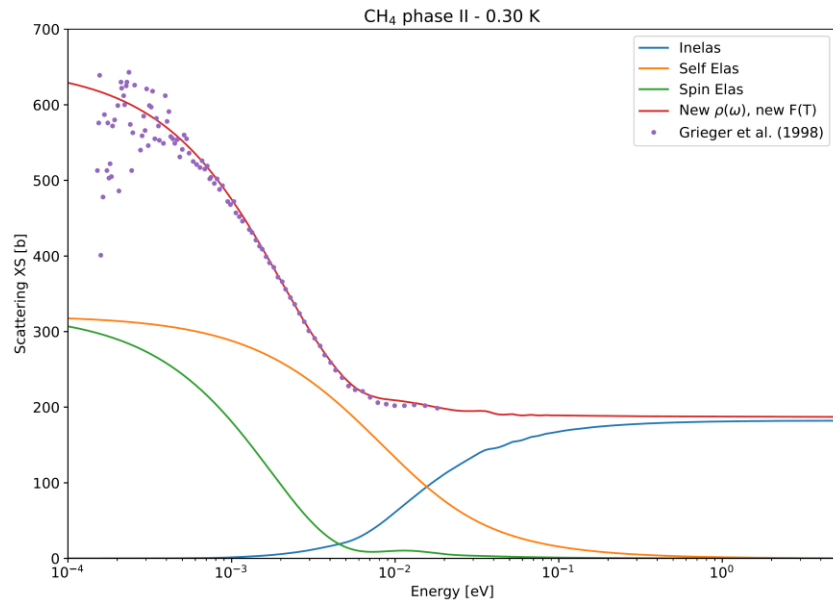
DOS of solid CH₄



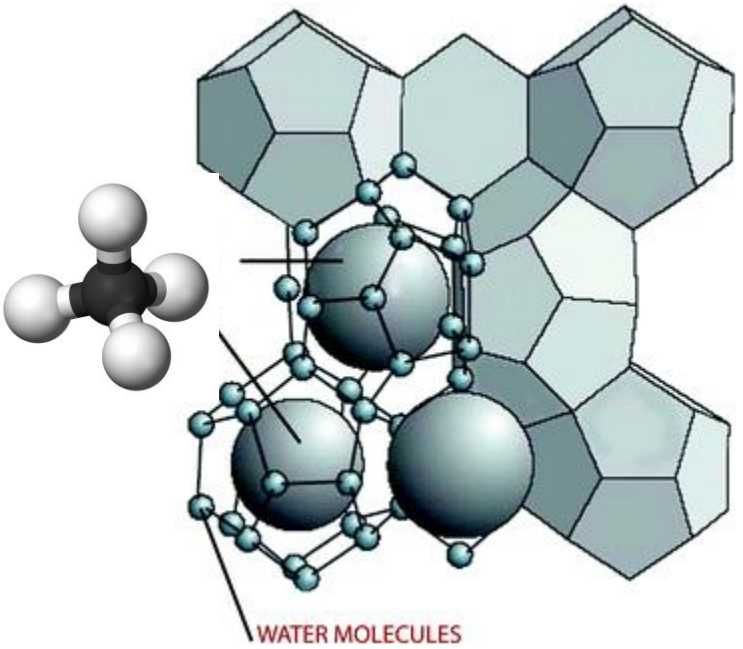
DOS of solid CH₄ (II)

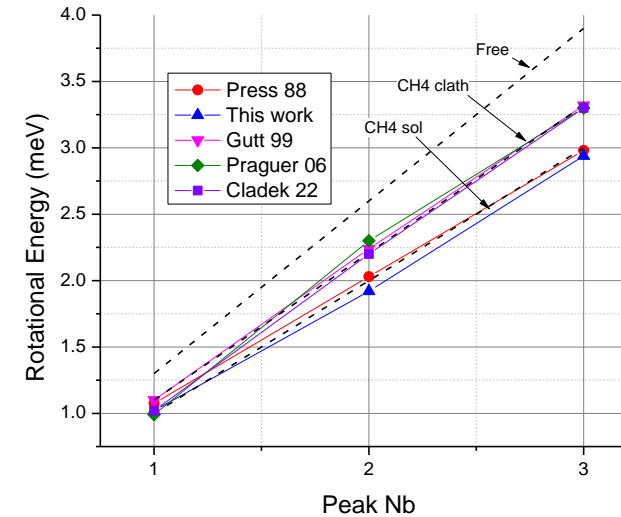
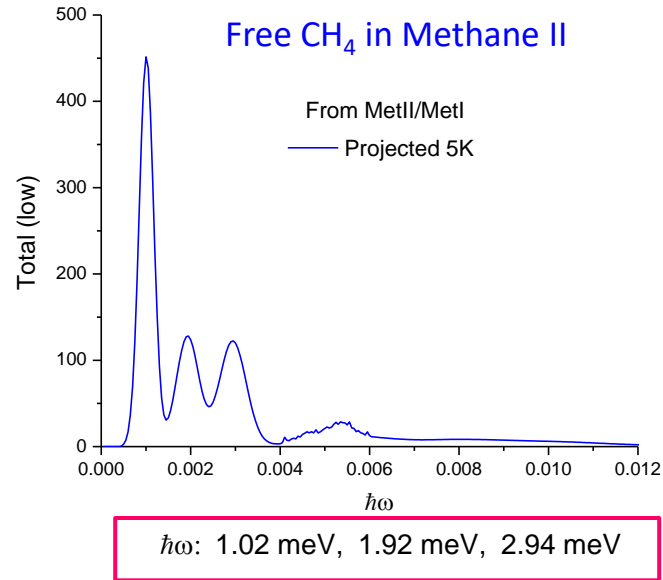






METHANE d-CLATHRATE



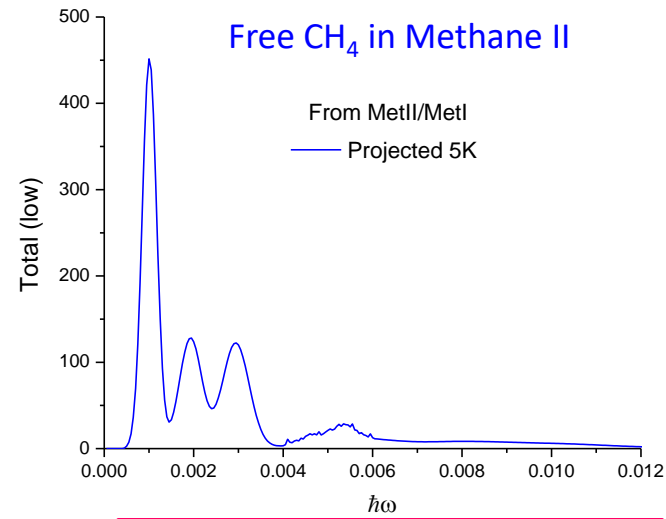
DOS of CH₄ in *d*-Clathrate

The eigenvalues of the Hamiltonian for *free* rotating CH₄ molecules are

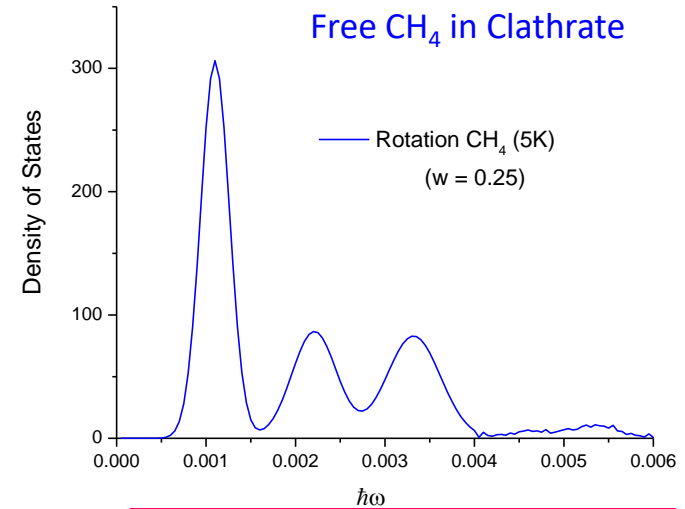
$$E_j = \hbar^2 j(j+1) / (2I_0)$$

with the value of the rotational constant $\hbar^2 / (2I_0) = 0.655$ meV.

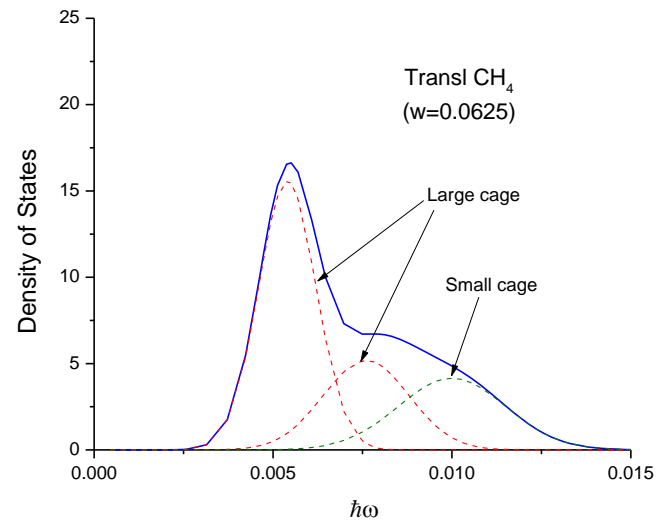
DOS of CH₄ in d-Clathrate



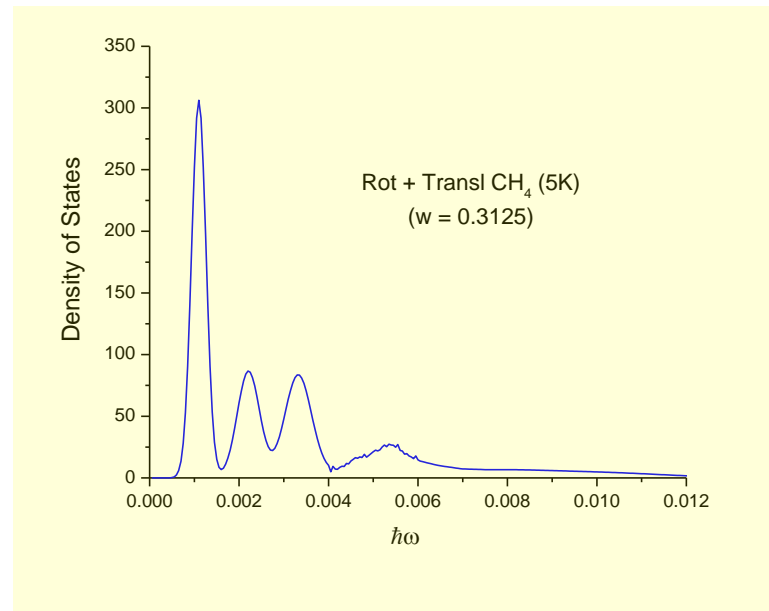
$\hbar\omega$: 1.02 meV, 1.92 meV, 2.94 meV

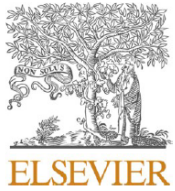


$\hbar\omega$: 1.1 meV, 2.21 meV, 3.32 meV



J.R. Granada, from measured data [J. Baumert, Thesis, 2003]
(5.4 meV, 7.6 meV, 10 meV)





Full Length Article

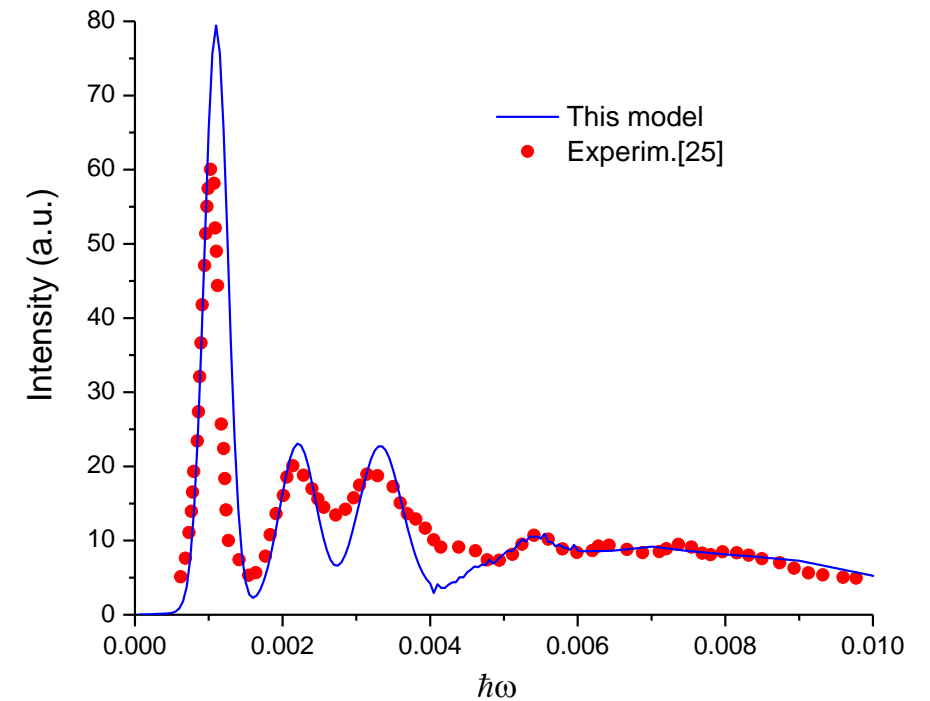
In situ inelastic neutron scattering of mixed CH₄-CO₂ hydrates

Bernadette R. Cladek^{a,b}, A.J. Ramirez-Cuesta^b, S. Michelle Everett^b, Marshall T. McDonnell^c,
 Luke Daemen^b, Yongqiang Cheng^b, Paulo H.B. Brant Carvalho^d, Christopher Tulk^b,
 Matthew G. Tucker^b, David J. Keffer^a, Claudia J. Rawn^{a,*}



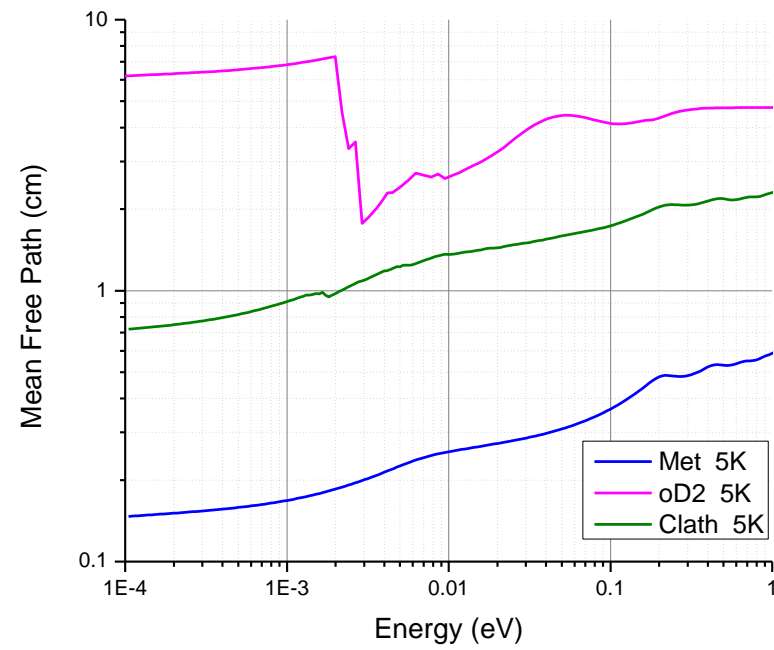
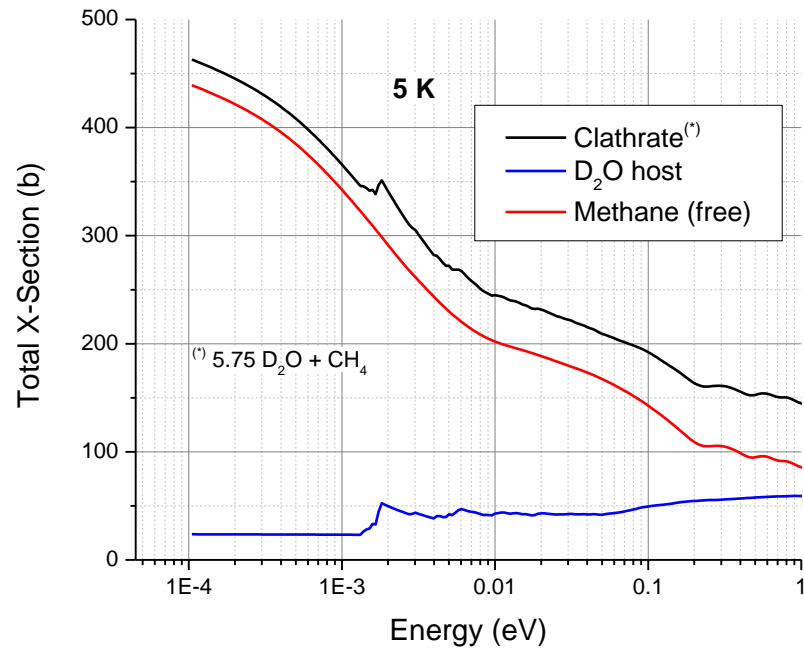
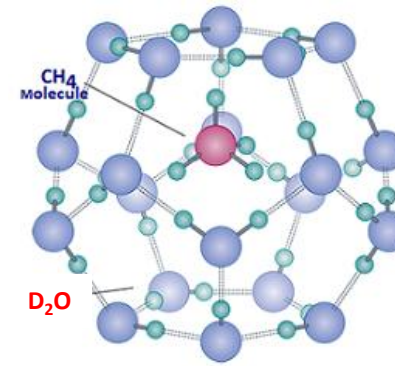
INS data compared with the *weighted DOS* of the system:

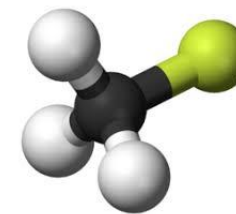
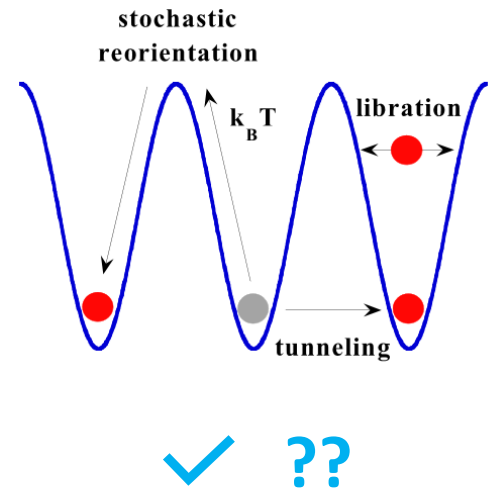
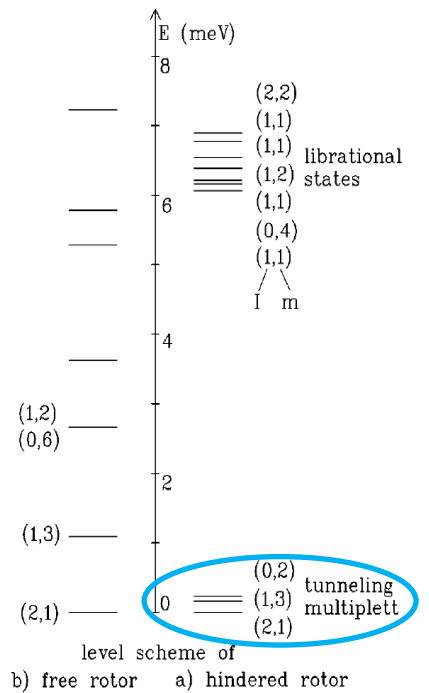
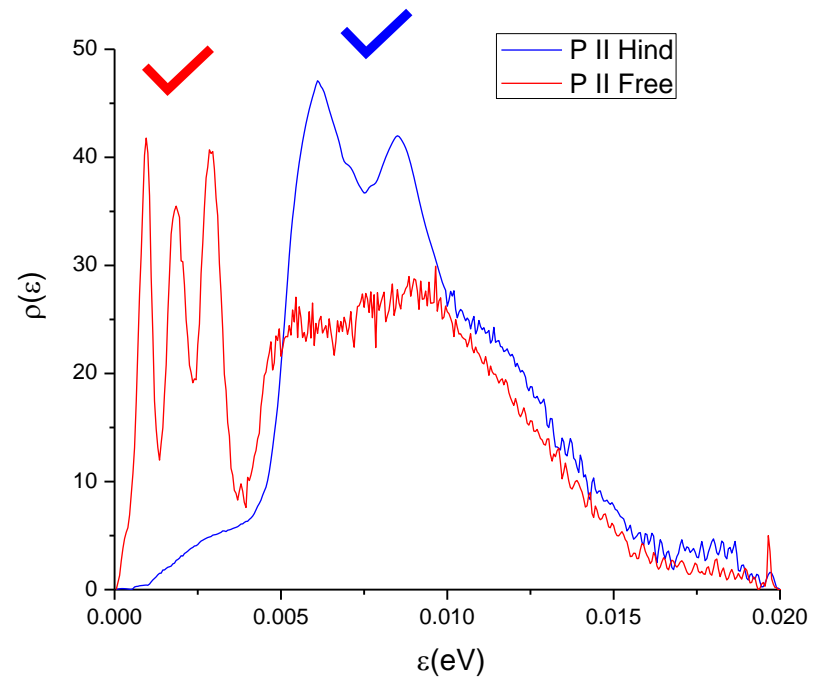
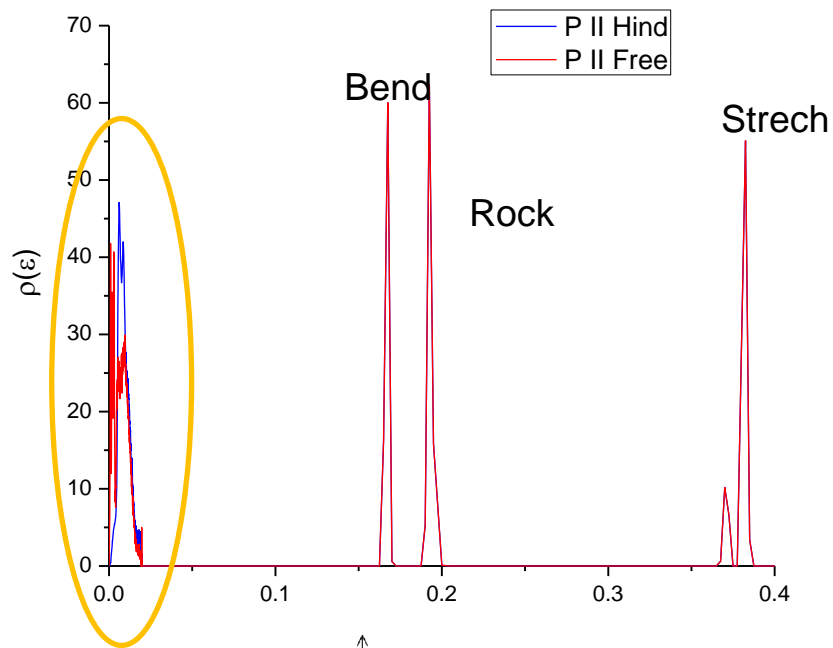
$$\cong \sigma(4H) \rho(H \text{ in CH}_4) + 5.75 \sigma(2H) \rho(H \text{ in H}_2\text{O ice})$$



Protonated Methane Clathrate, 5K

XS CH₄ d-CLATHRATE

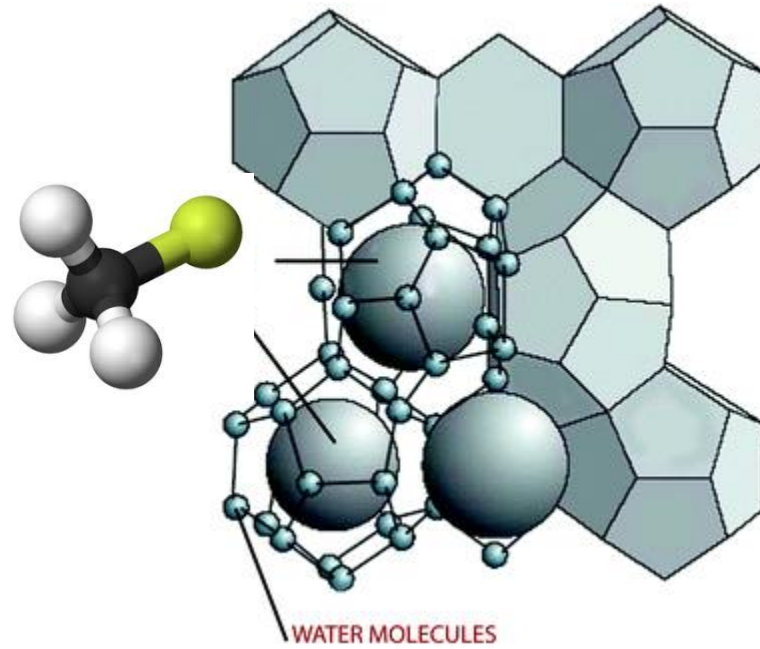




Dynamics of methyl groups in molecular solids

??

METHYL FLUORIDE IN CLATHRATE



Adsorption sites and rotational tunneling of methyl groups in cubic I methyl fluoride water clathrate†

M. Prager,^a J. Baumert,^b W. Press,^b M. Plazanet,^b J. S. Tse^c and D. D. Klug^c

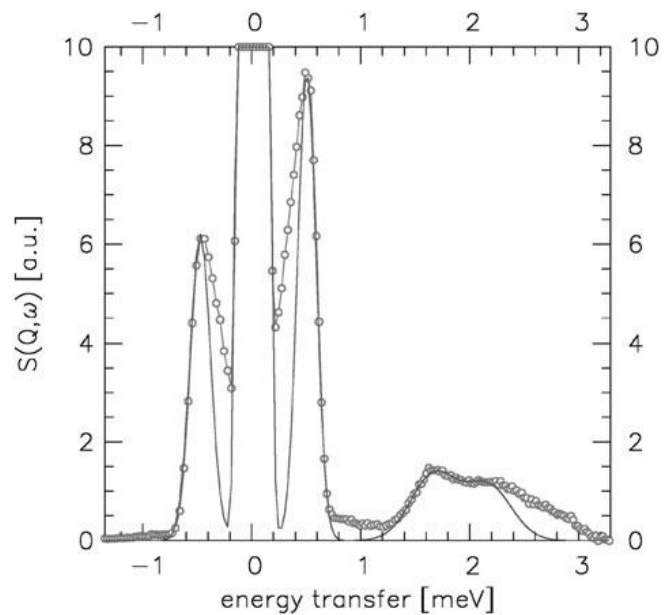
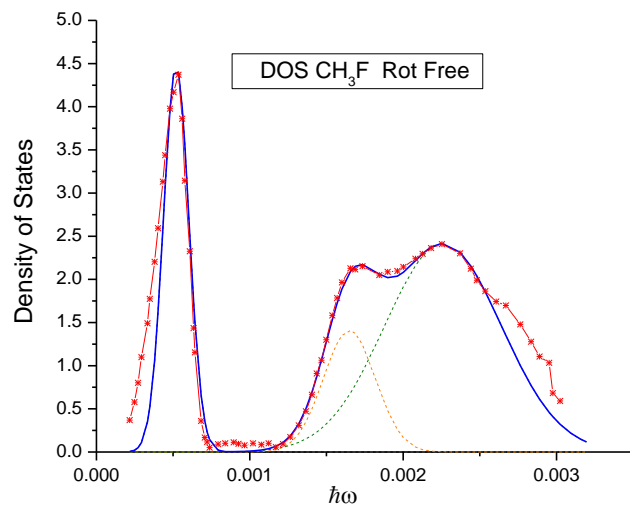
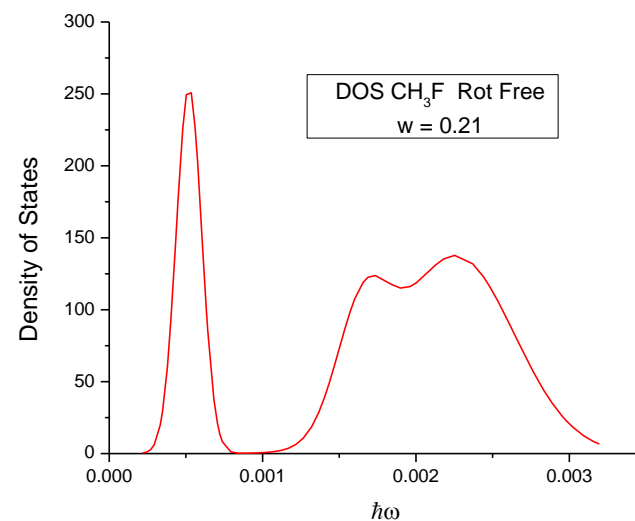


Fig. 9 Lower resolution spectrum including higher rotational levels of methyl fluoride water clathrate. Fully protonated sample. Sample temperature $T = 2.0$ K. Spectrometer: IN5 of ILL. Wavelength $\lambda = 4.5$ Å. Average momentum transfer $Q = 2.3$ Å⁻¹.

$$S(Q, \omega) = \bar{\sigma} \frac{\hbar^2 Q^2}{2\bar{m}} \exp(-2W) \frac{G(\omega)}{\hbar\omega} [n(\omega) + 1]$$



$\hbar\omega$: 0.52 meV, 1.65 meV, 2.26 meV



Adsorption sites and rotational tunneling of methyl groups in cubic I methyl fluoride water clathrate†

M. Prager,^a J. Baumert,^b W. Press,^b M. Plazanet,^b J. S. Tse^c and D. D. Klug^c

Fig. 2 Neutron scattering spectrum taken in the regime of lattice modes and summed over all scattering angles. Spectrometer: SV29 at Forschungszentrum Jülich. Wavelength $\lambda = 1.81$ Å. Sample temperature $T = 4.2$ K. Average momentum transfer $Q = 4.9$ Å⁻¹.

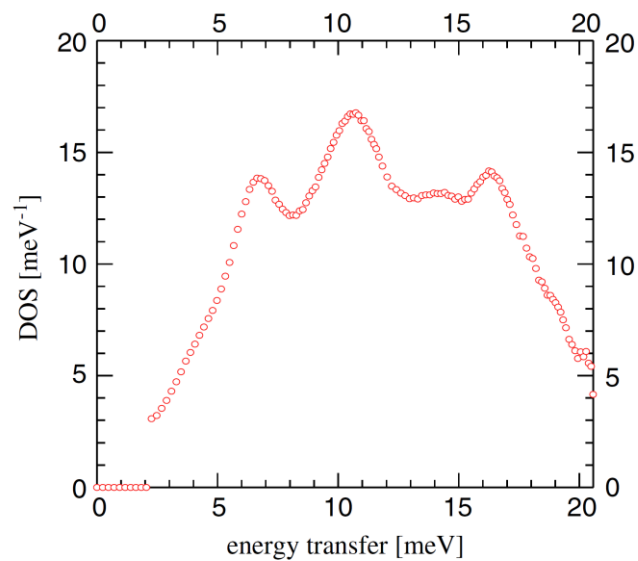
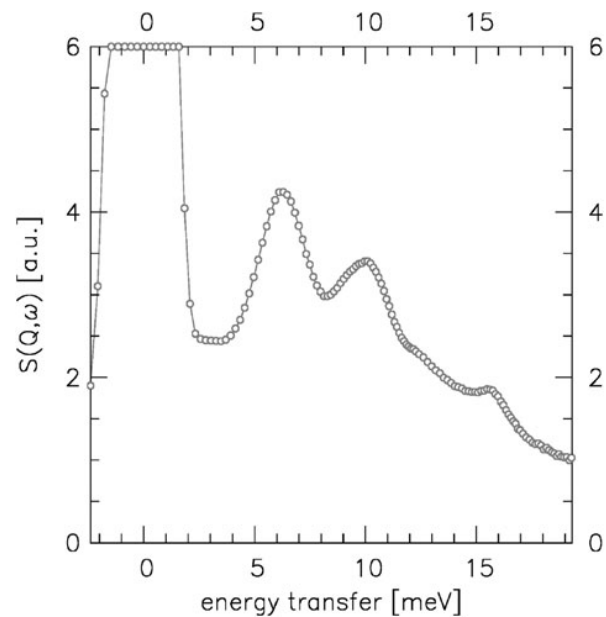
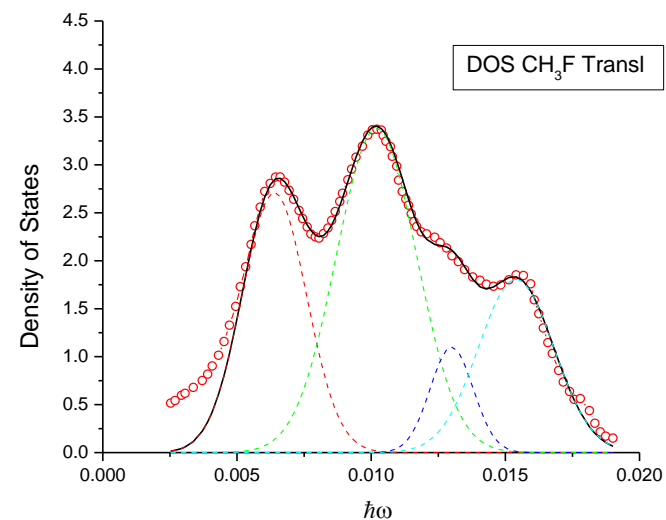


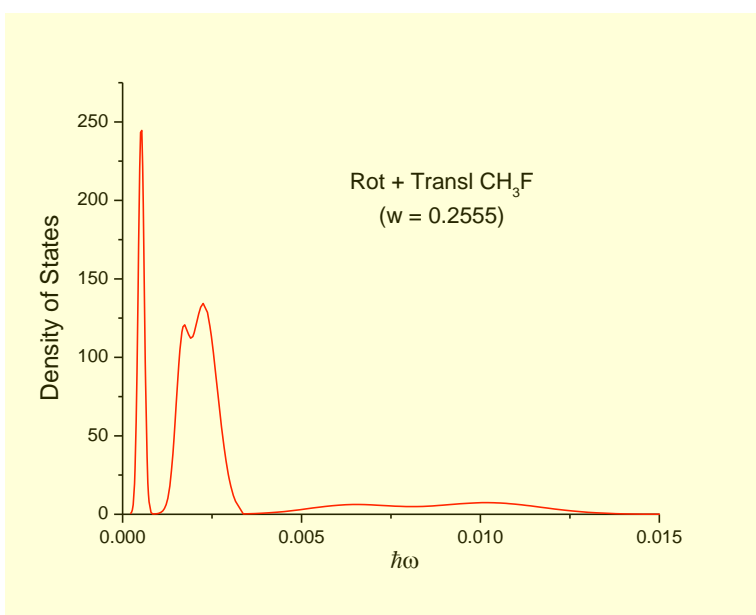
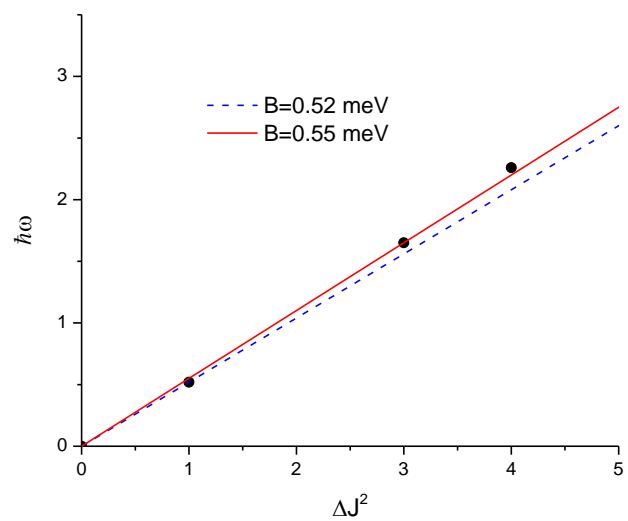
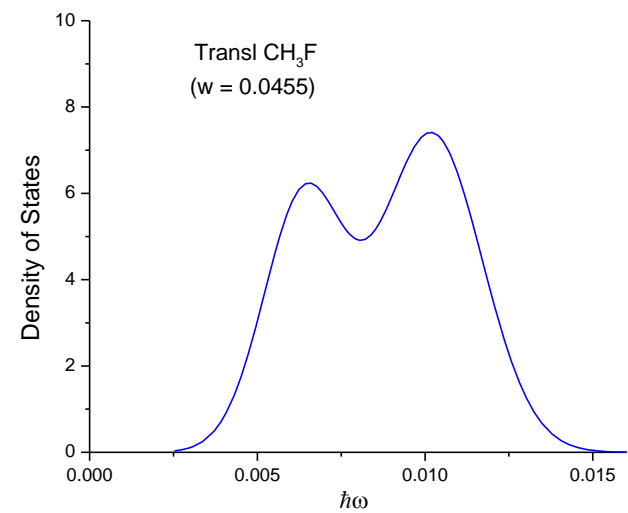
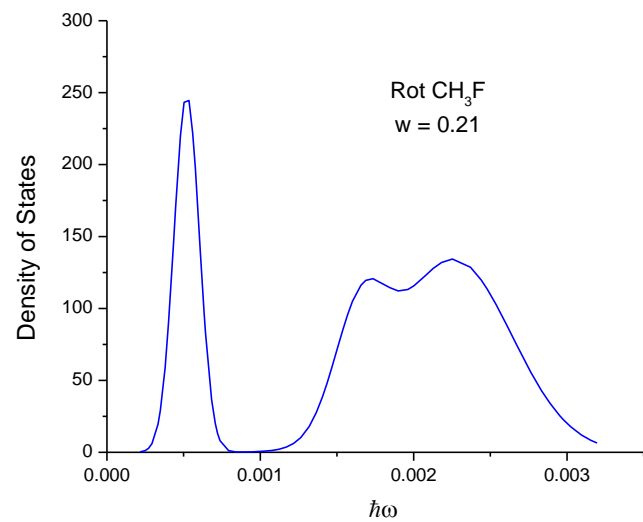
Figure 7. Generalized density of states of $\text{CH}_3\text{I}\cdot 17\text{D}_2\text{O}$ in the meV range.

$$S(Q, \omega) = \bar{\sigma} \frac{\hbar^2 Q^2}{2\bar{m}} \exp(-2W) \frac{G(\omega)}{\hbar\omega} [n(\omega) + 1]$$



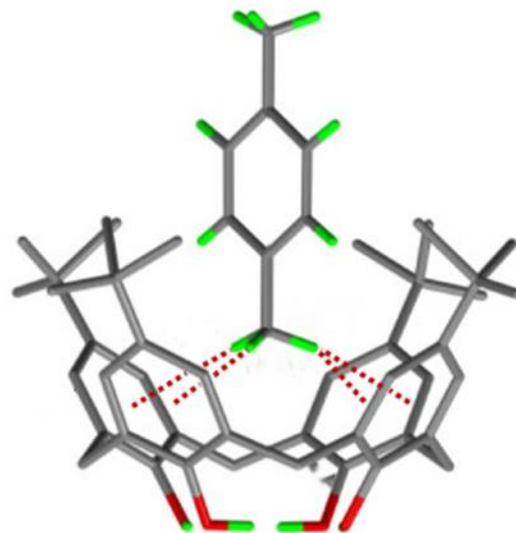
$\hbar\omega$: 6.39 meV, 10.2 meV, 13 meV, 15.4 meV

DOS – CH₃F in Clathrate

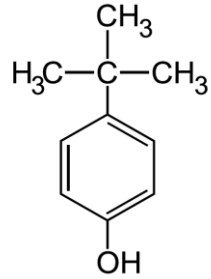


**METHYL ROTATIONS in
CALIXARENE SUPRAMOLECULAR COMPLEXES**

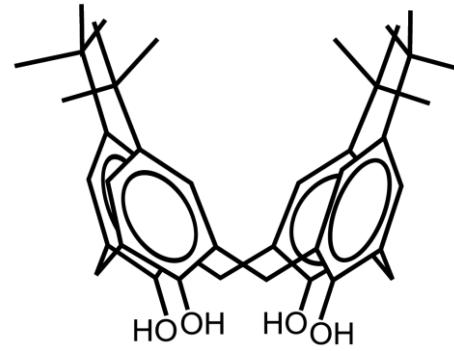
p -tert-butyl-calix[4]arene(2:1) p -xylene



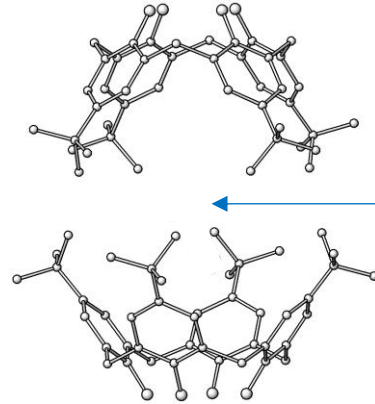
p-tert-butylphenol



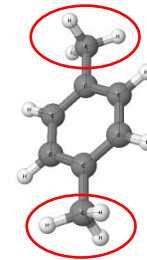
p-tert-butylcalix[4]arene



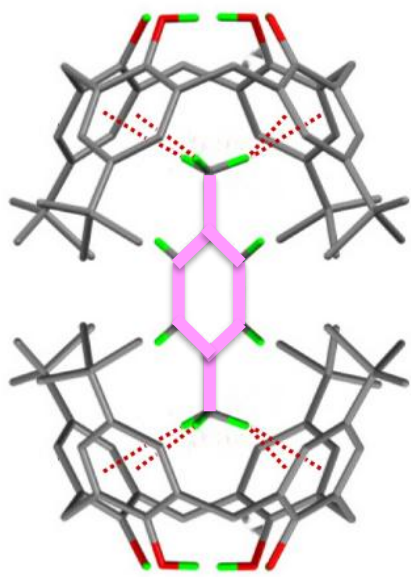
p-tert-butyl-calix[4]arene(2)



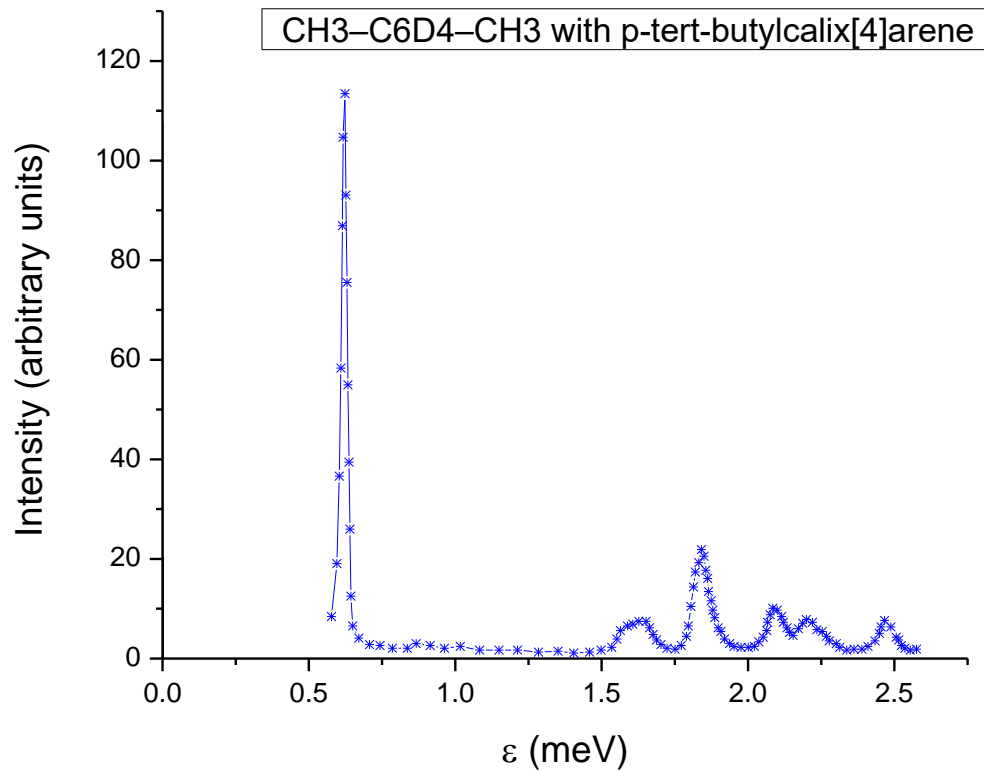
p-xylene



The molecular units, the conformation, and the actual structure of the calixarene complex.



p -tert-butyl-calix[4]arene(2:1) p -xylene



P. Schiebel *et al.*,
 J. Phys.: Condens. Matter **10** (1998) 2221

| Transition | Energy (meV) | Obs. intensity | Calc. intensity |
|-----------------|--------------|----------------|-----------------|
| 0 → I | 0.626 | 100 (±3) | 100 |
| I → II | 1.24 | 2 (±1) | 2 |
| I → III | 1.60 | 11 (±3) | 12 |
| 0 → II + I → IV | 1.85 | 39 (±4) | 39 |
| I → V | 2.10 | 15 (±1) | 15 |
| 0 → III | 2.21 | 19 (±2) | 17 |
| 0 → IV | 2.48 | 13 (±1) | 12 |
| 0 → V | 2.73 | 4 (±3) | 12 |

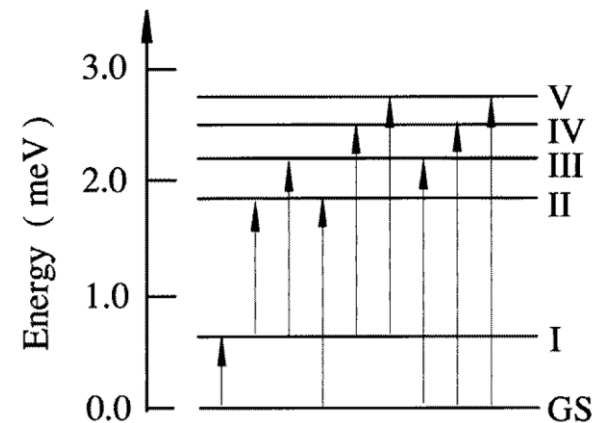
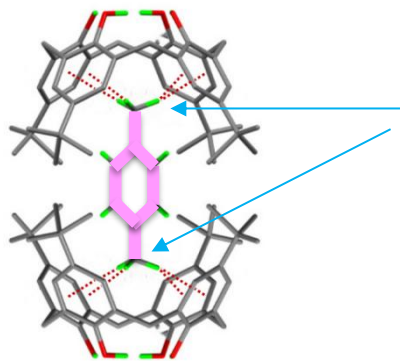
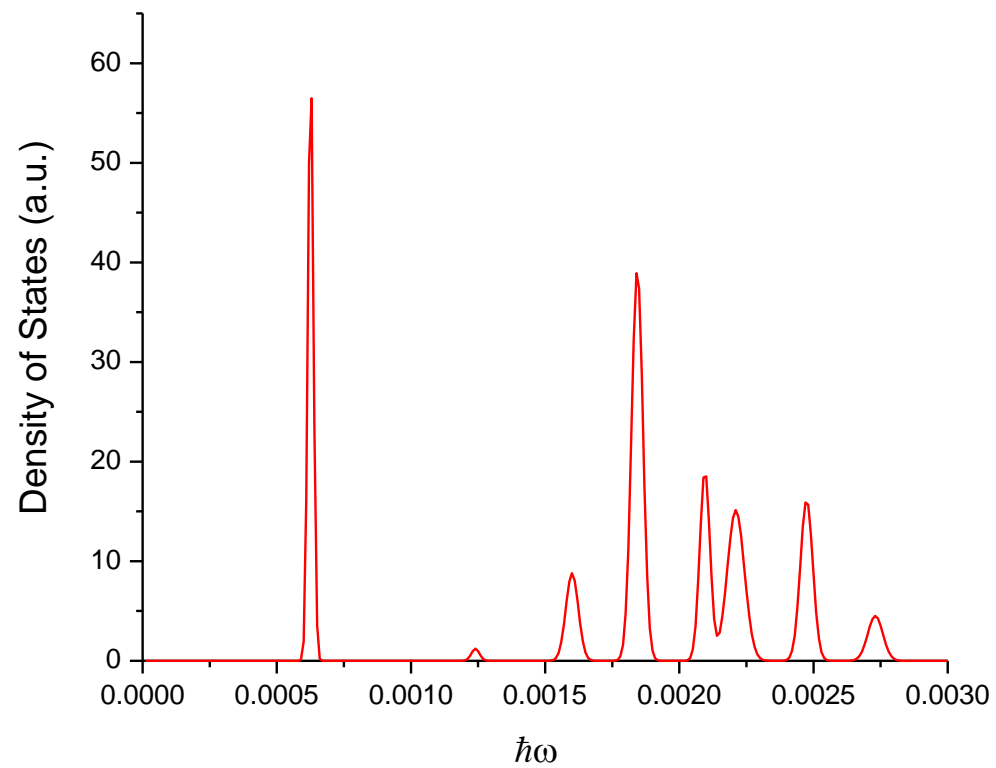


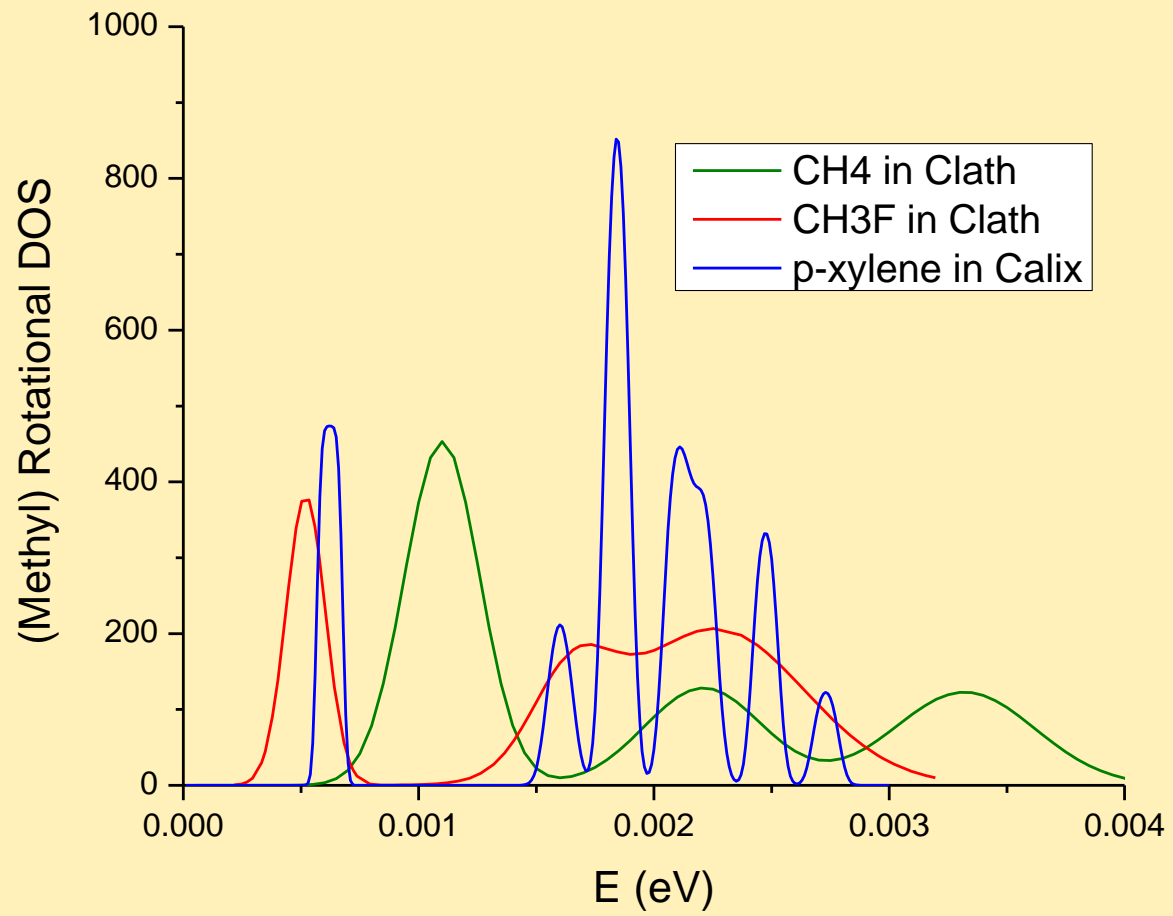
Figure 5. Energy level scheme for the rotation–translation motion of the *p*-xylene methyl groups in the calixarene complex. The arrows indicate transitions with non-zero intensity.



Methyl rotations in
p-tert-butyl-calix[4]arene(2:1) p-xylene



Derived from P. Schiebel *et al.* (1998)



CONCLUSIONS

We have explored and proposed a few molecular systems that show dynamical features of interest to act as moderators to produce **very cold neutrons**. Their densities of states have been derived from available experimental information, which are the relevant input for the elaboration of reliable scattering kernels, using modern computer packages.

It is important to emphasize that at the few Kelvins temperature range that those VCN moderators should operate, spin correlations develop amongst the protons in the CH_3 and CH_4 units and they must properly be accounted for in the cross sections' calculations.

CONCLUSIONS

A very interesting material for producing VCN's is the supramolecular complex *p-tert-Butyl-Calix[4]arene(2:1)p-Xylene* at low temperatures. The reason is the very rich low energy excitations' spectrum associated with the dynamics of the methyl groups in the p-xylene molecule. An additional, highly important property associated with this system is its expected strong radiation resistance due to the capacity of the aromatic rings to redistribute the energy of incident photons or fast neutrons, thus avoiding the radiolytic hydrogen production within a moderator immersed in strong radiation fields.

CONCLUSIONS

Finally, I wish to emphasize that the ideas presented here are the result of an exploratory study, aimed at showing a possible road rather than defining new materials that could perform well as VCN moderators.

This kind of ideas must be investigated with extensive use of experimental information, typically inelastic and other neutronic measurements. For this purpose, besides the availability of good spectrometers in all major neutron sources, smaller, highly flexible CANS (e.g. Hokkaido LINAC, LENS) will be needed to test new materials and validate their scattering kernels and cross section libraries.

THANKS FOR YOUR ATTENTION

Proceedings of the Workshop on
Applications of a Very Cold Neutron Source



ANL-05/42

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August 21-24, 2005

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