



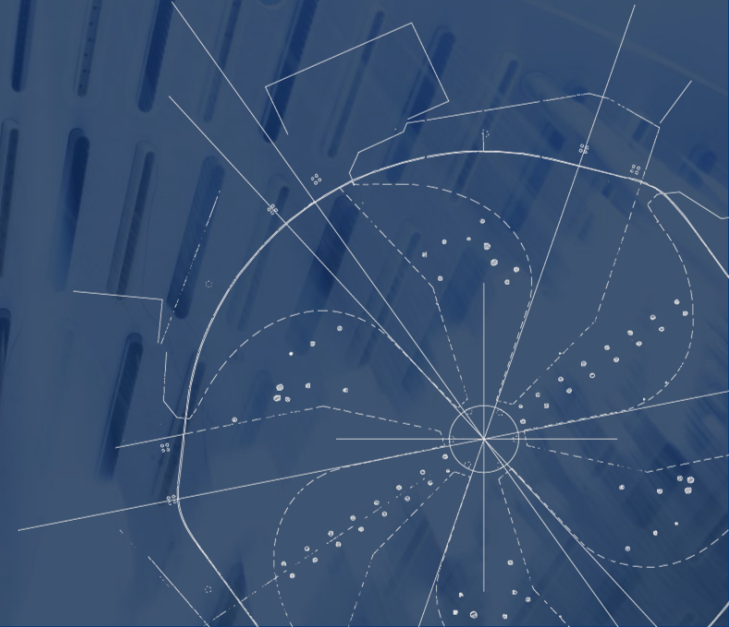
TRIUMF

Canada's national laboratory
for particle and nuclear physics
and accelerator-based science

Ion traps at rare-isotope-beam facilities

Ania A. Kwiatkowski
Research Scientist, TRIUMF
Adjunct Asst. Professor, Univ. Victoria

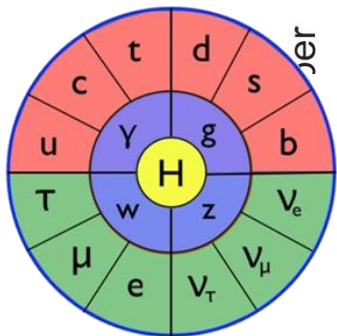
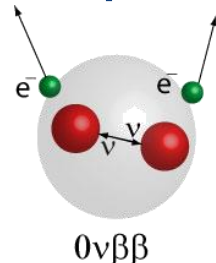
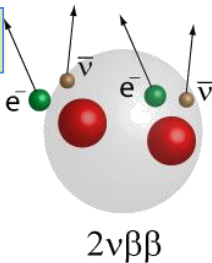
Winter Nuclear and Particle Physics Conference
17 February 2017



The atomic mass is needed input to understand ...

Neutrino physics

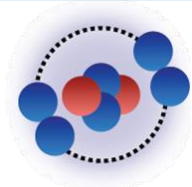
$$\delta m/m \leq 10^{-9}$$



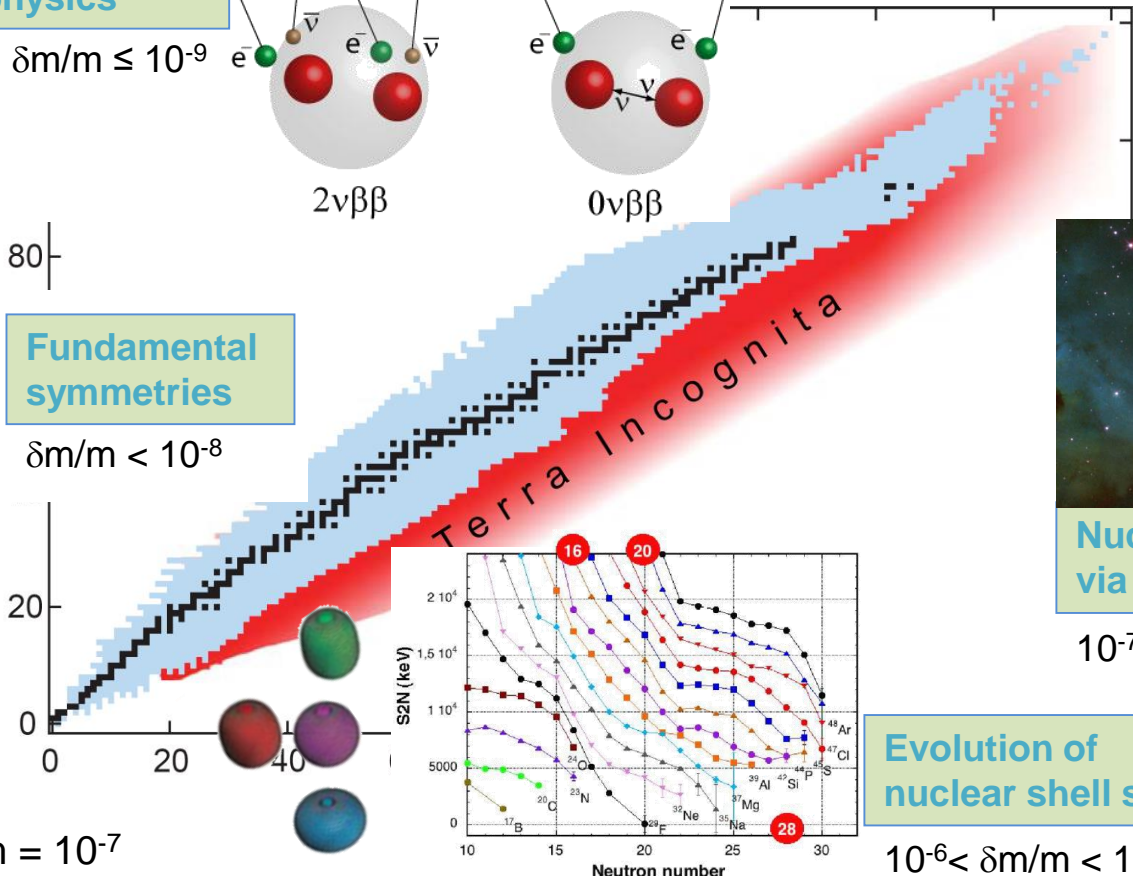
Fundamental symmetries

$$\delta m/m < 10^{-8}$$

Halos and skins



$$\delta m/m = 10^{-7}$$



Nucleosynthesis via r process

$$10^{-7} < \delta m/m < 10^{-6}$$

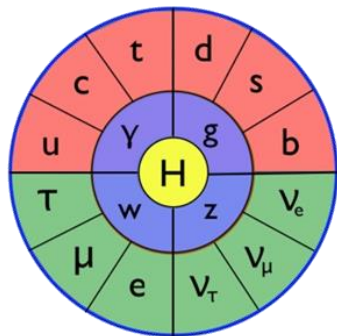
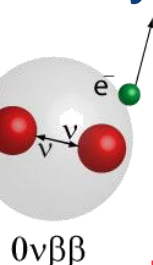
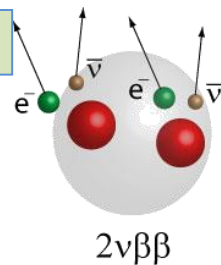
Evolution of nuclear shell structure

$$10^{-6} < \delta m/m < 10^{-5}$$

The desired precision is not always there.

Neutrino physics

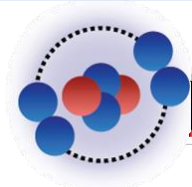
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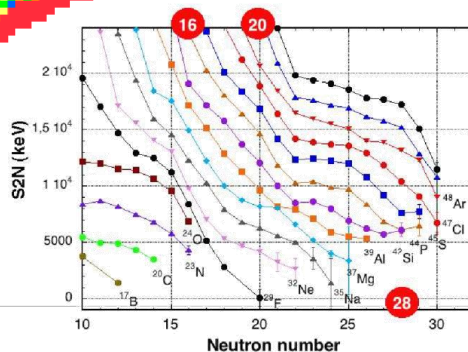
Fundamental symmetries

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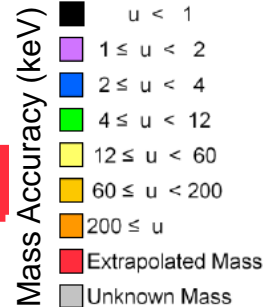


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Evolution of nuclear shell structure

$$10^{-6} < \delta m/m < 10^{-5}$$



Nucleosynthesis via r process

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How can we make high-precision measurements?

My wish list for an ideal laboratory

- easy & exact manipulation
- well-defined volume
- universal
- infinite observation time

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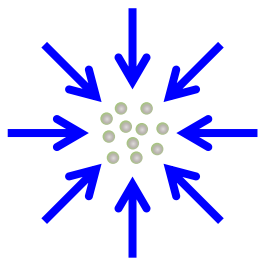


relies on
electromagnetic forces
+
good vacuum

My wish list for an ideal laboratory

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Want $\vec{F} = -q\nabla\varphi \propto -\vec{r}$



& satisfy $\Delta\varphi = 0$

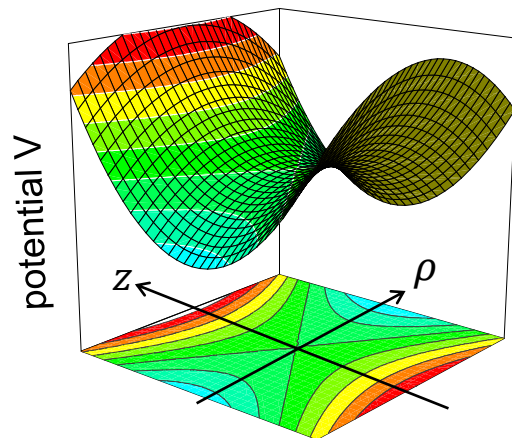
\rightarrow find $\varphi \propto \frac{\varphi_0^2}{d^2} (\rho^2 - z^2)$

relies on

electromagnetic forces

+

good vacuum



Four principal ion storage devices at RIB facilities.

Moving ions

“Stationary” ions

Four principal ion storage devices at RIB facilities.

Moving ions

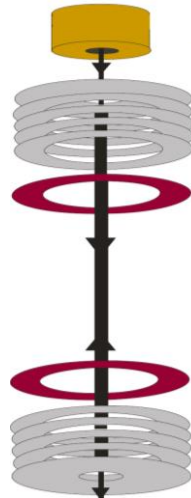
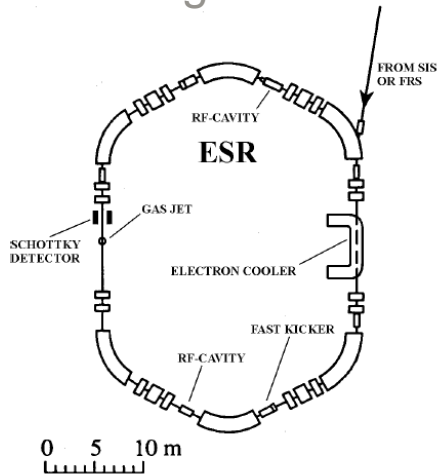
“Stationary” ions

Storage Rings

Multi-reflection/
electrostatic traps

Electromagnetic
forces restrict ions
to a ring.

Electrostatic
mirrors trap ions.



Four principal ion storage devices at RIB facilities.

Moving ions

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Storage Rings

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Paul trap

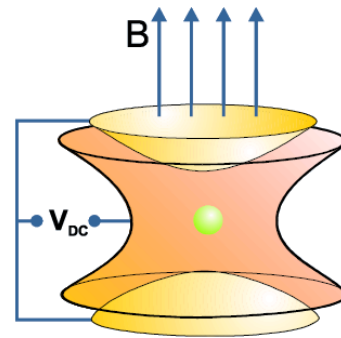
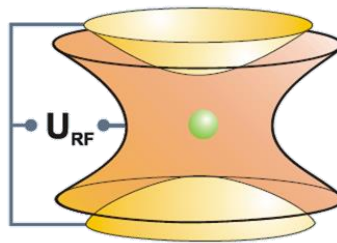
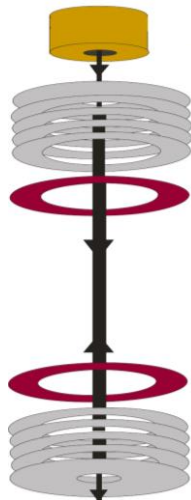
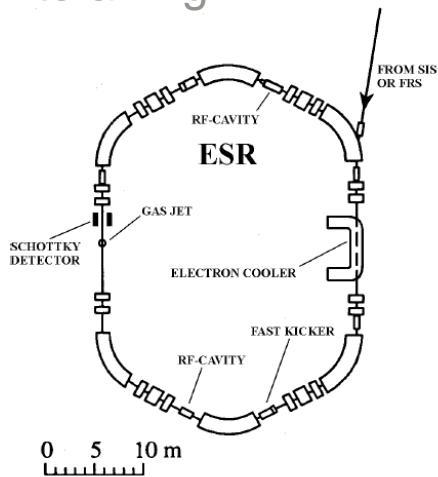
Penning trap

Electromagnetic forces restrict ions to a ring.

Electrostatic mirrors trap ions.

Oscillating (RF) electric field between ring & end caps

Superposition of magnetic field (z) & electrostatic quadrupolar field



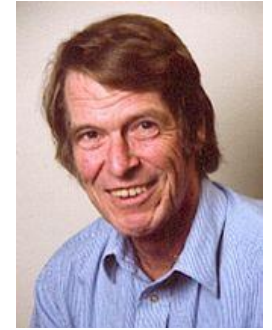
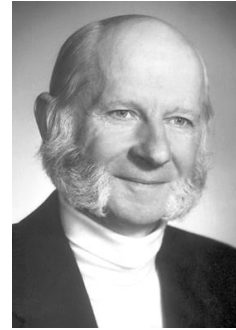
Ion traps are proliferating for a variety of purposes.



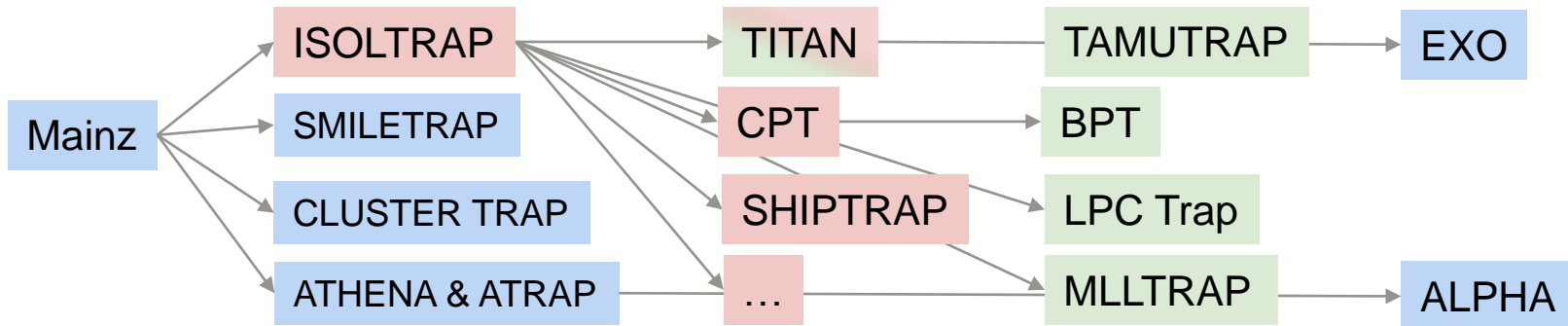
Frans Michel Penning
Foundation for Penning trap



W. Paul
Nobel Prize 1989
Development of ion trap technique



H.-J. Kluge
Adaptation for RIB facilities



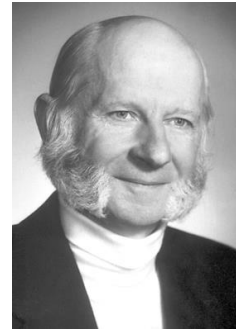
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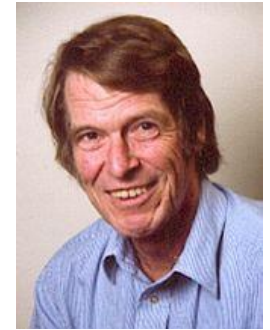


W. Paul

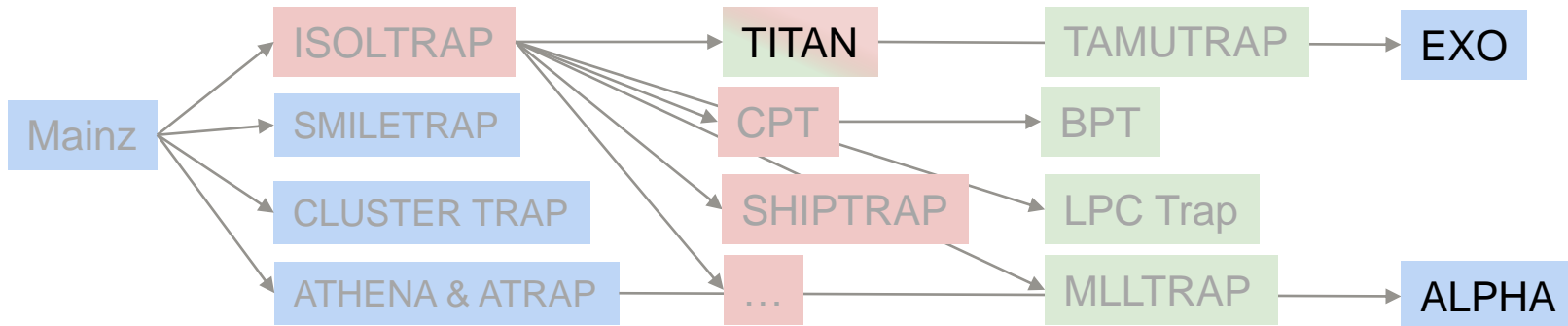


H. Dehmelt

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Development of ion trap technique



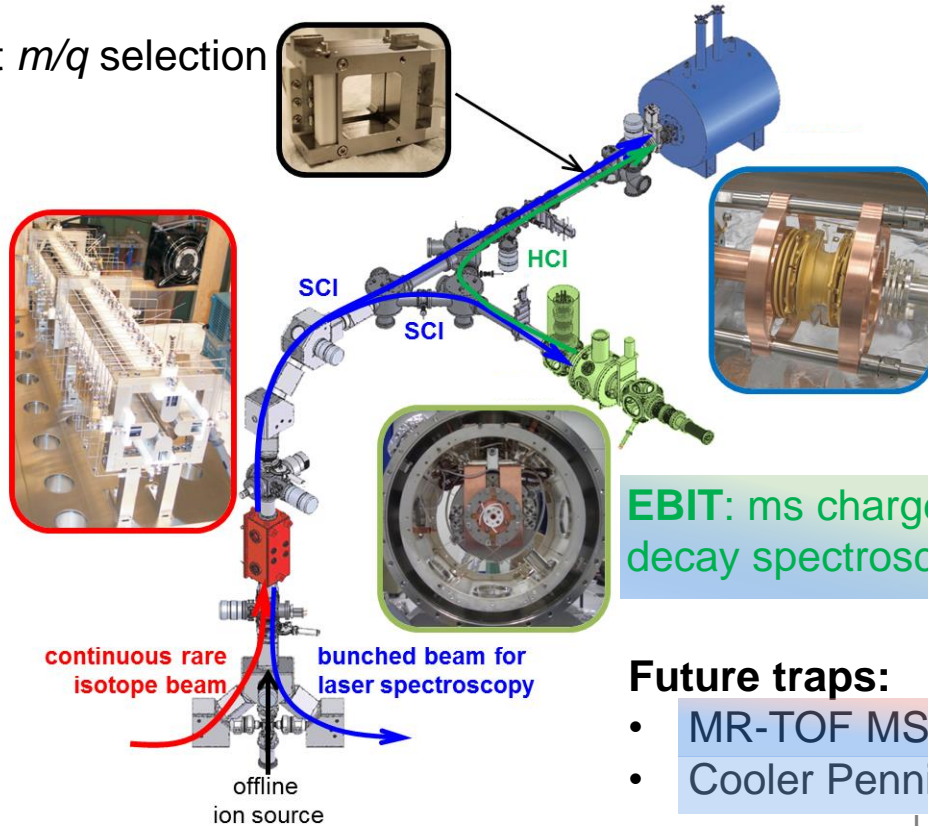
H.-J. Kluge
Adaptation for RIB facilities



TRIUMF's Ion Trap for Atomic and Nuclear science uses 5 traps to prepare & measure RIB.

BNG: fast m/q selection

Cooler-Buncher:
accumulation,
cooling, and
bunching



MPET: mass
measurement via
cyclotron frequency
determination

EBIT: ms charge breeding
decay spectroscopy

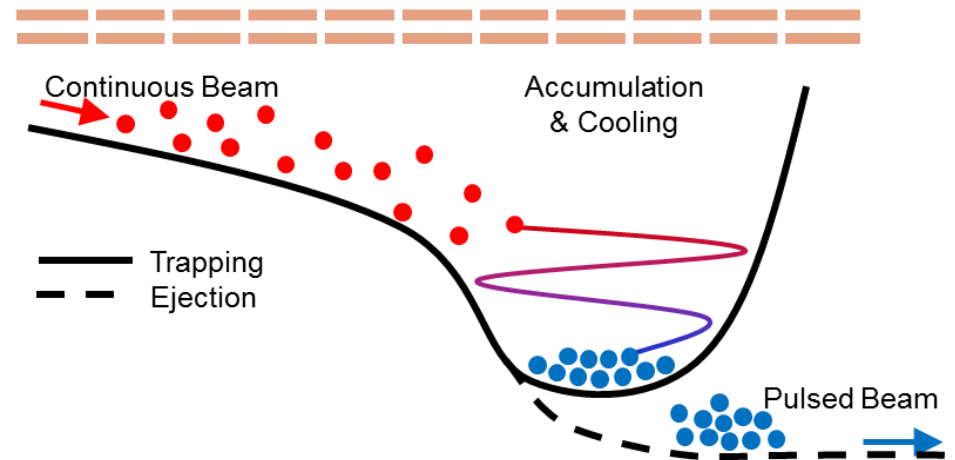
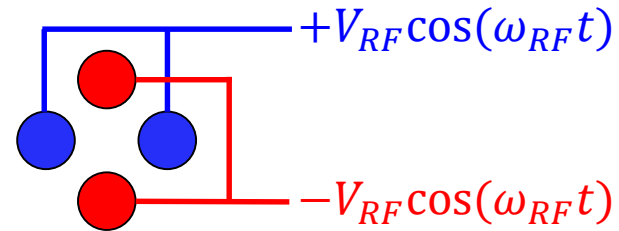
Future traps:

- MR-TOF MS for purification
- Cooler Penning trap to cool HCI



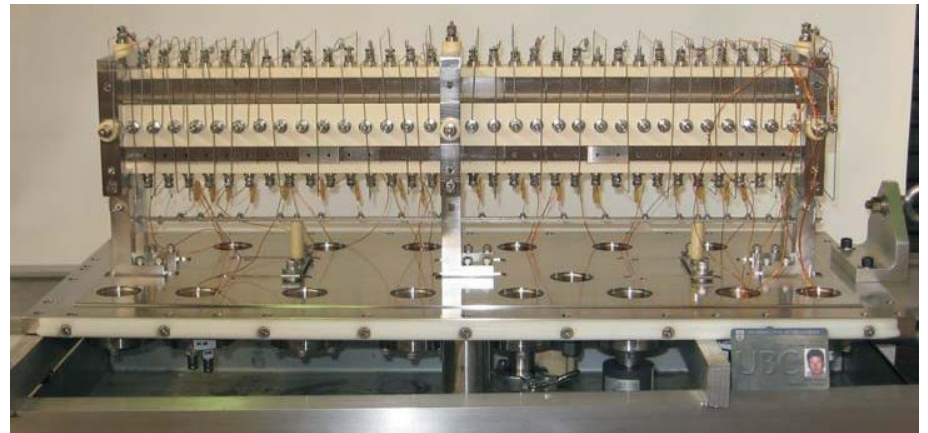
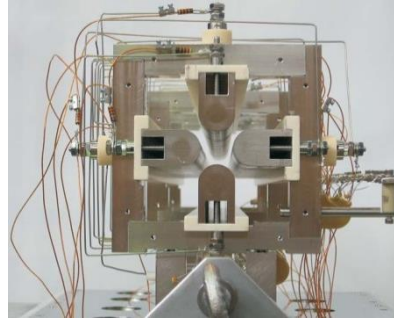
The buffer-gas-filled linear Paul trap accumulates, cools, & bunches the RIB.

- RadioFrequency Quadrupole \rightarrow transverse confinement
- Segmentation \rightarrow axial trapping
- Buffer gas \rightarrow cooling



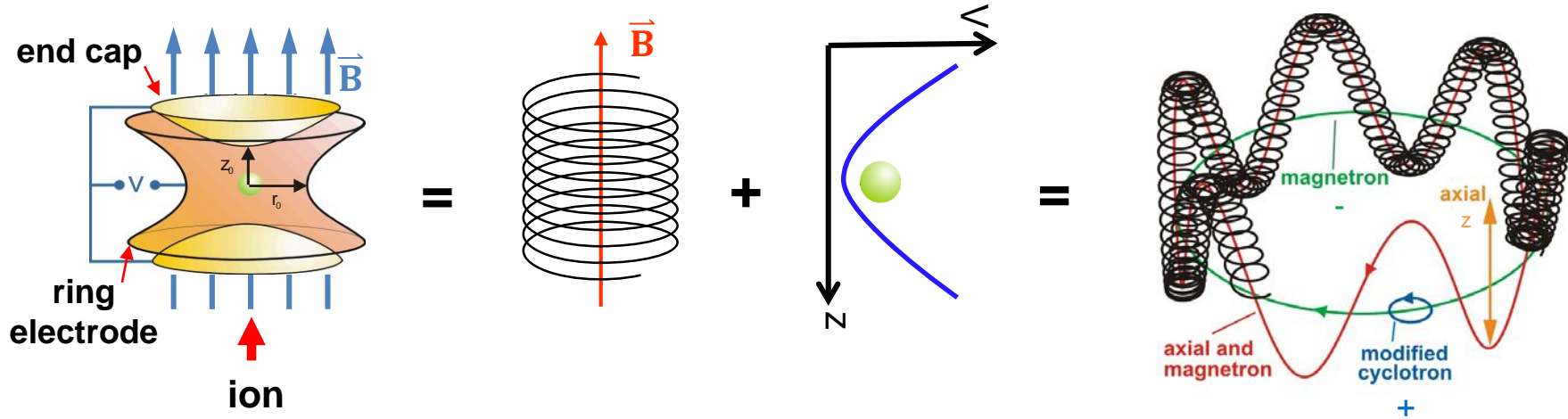
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A Penning trap accesses the cyclotron frequency & therefore the ion's mass.

$$2\pi\nu_c = (qe/m) \cdot B$$



RIB mass measurements with precisions up to $\sim 10^{-9}$
and for half-lives as low as 9 ms ($^{11}\text{Li}^+$ @ TITAN)

Measurement Penning Trap accesses the cyclotron frequency & therefore the ion's mass.

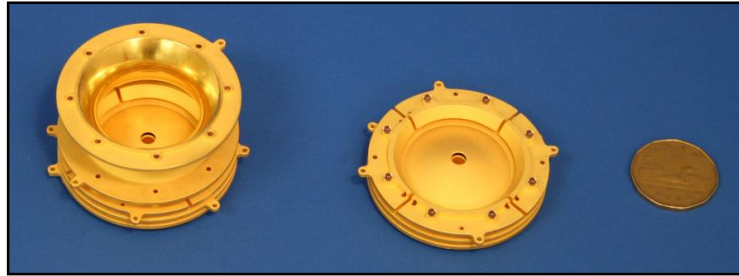
$$2\pi\nu_c = (qe/m) \cdot B$$

$$\nu_c = \nu_+ + \nu_-$$

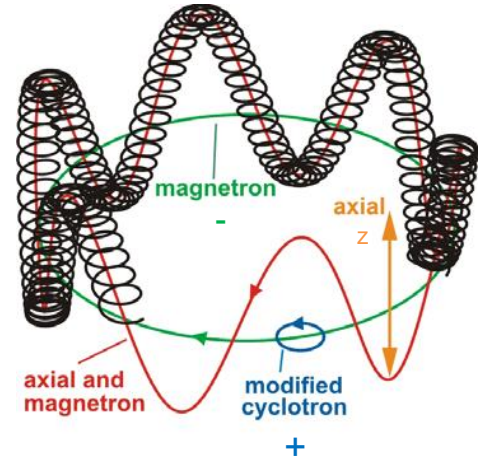


3.7 T

+

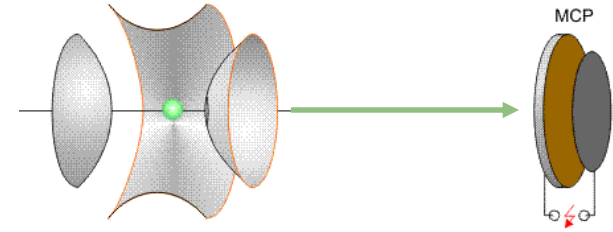
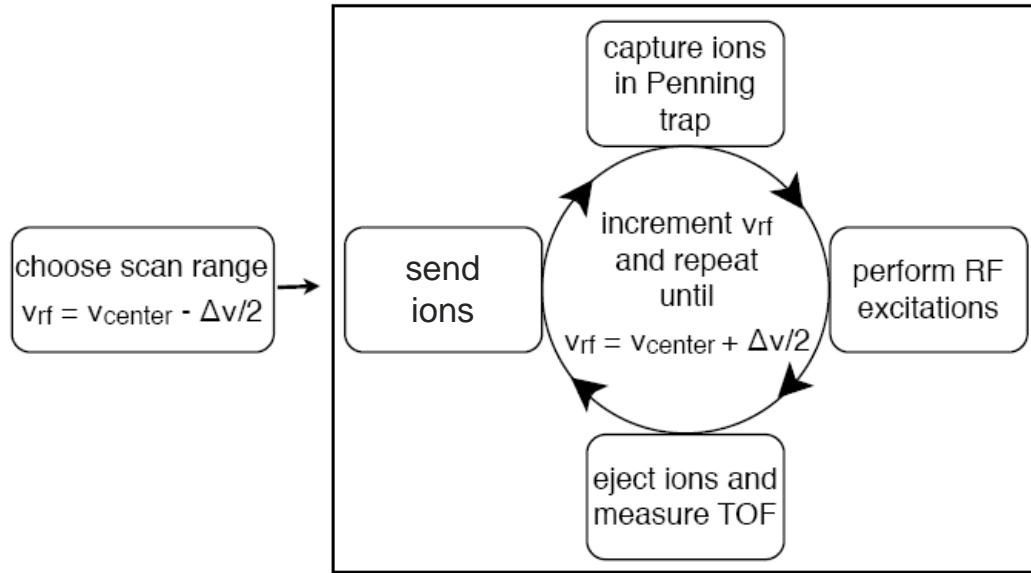


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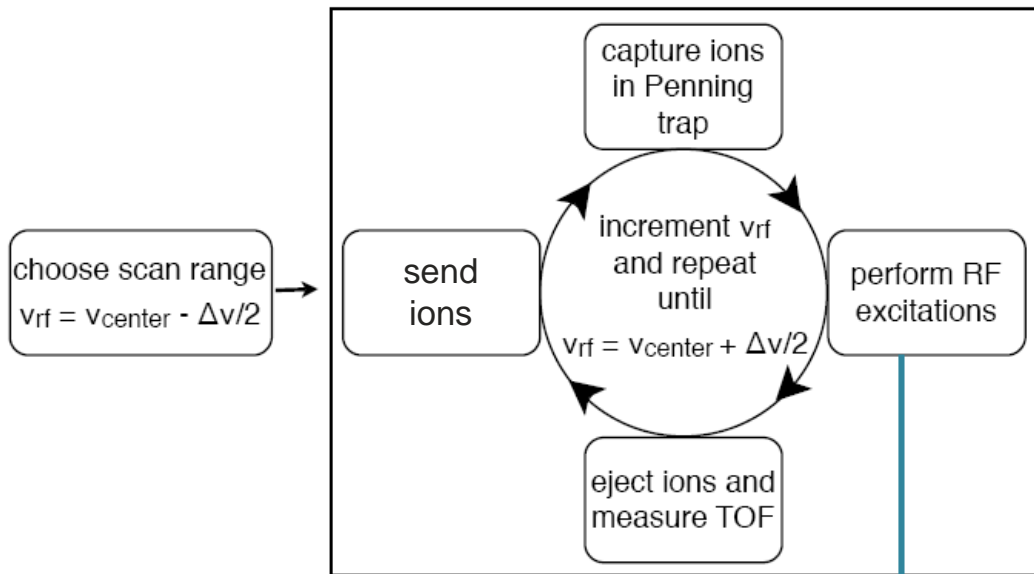


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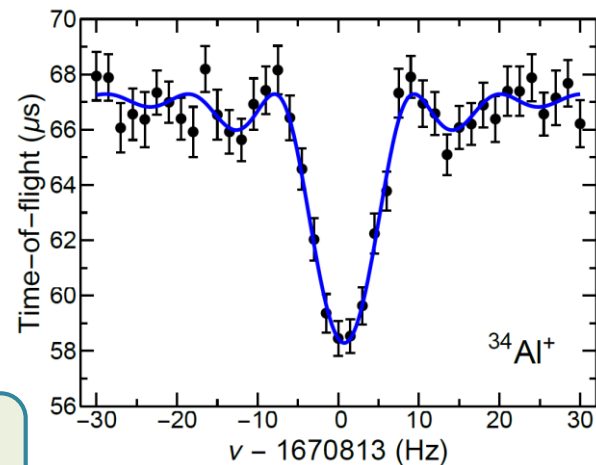
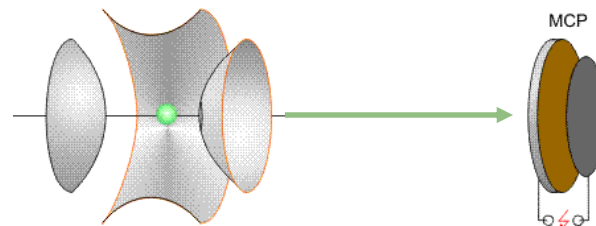
Cyclotron frequency can be determined via Time-of-Flight Ion-Cyclotron-Resonance technique.



Cyclotron frequency can be determined via Time-of-Flight Ion-Cyclotron-Resonance technique.



If $\nu_{rf} = \nu_c$, the ion's energy increases



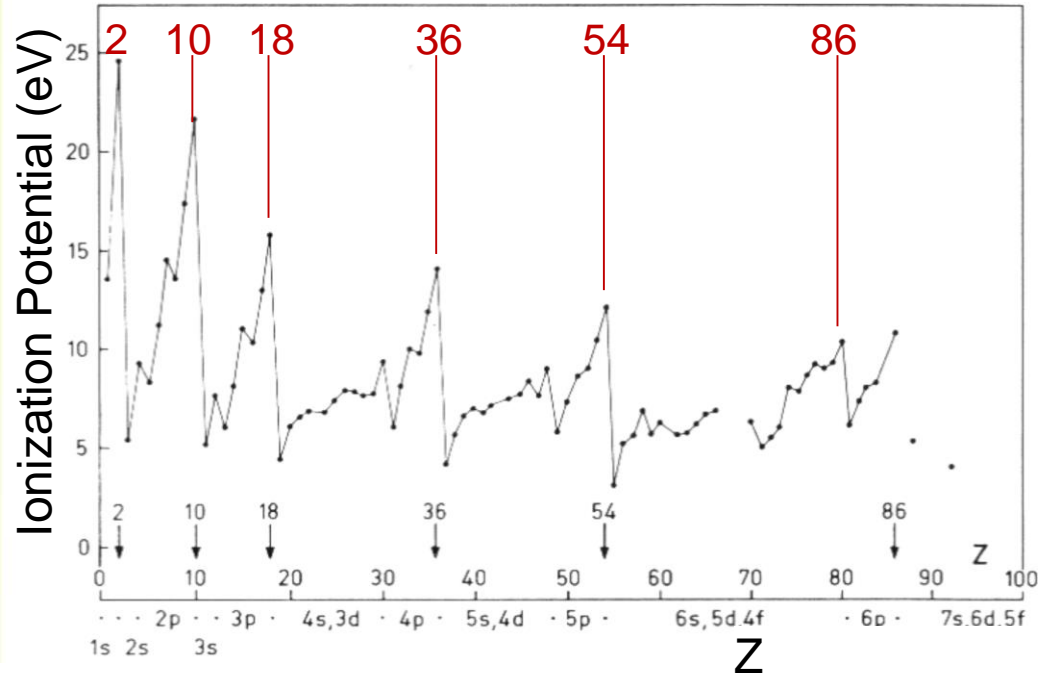
What is special about ^{34}Al and its
neighbours?

A brief aside in atomic physics: electrons occupy shells.

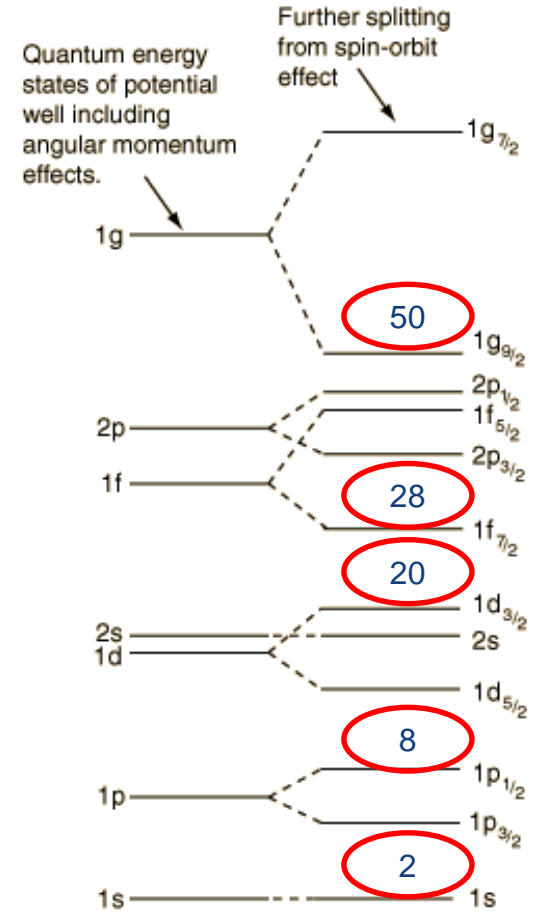
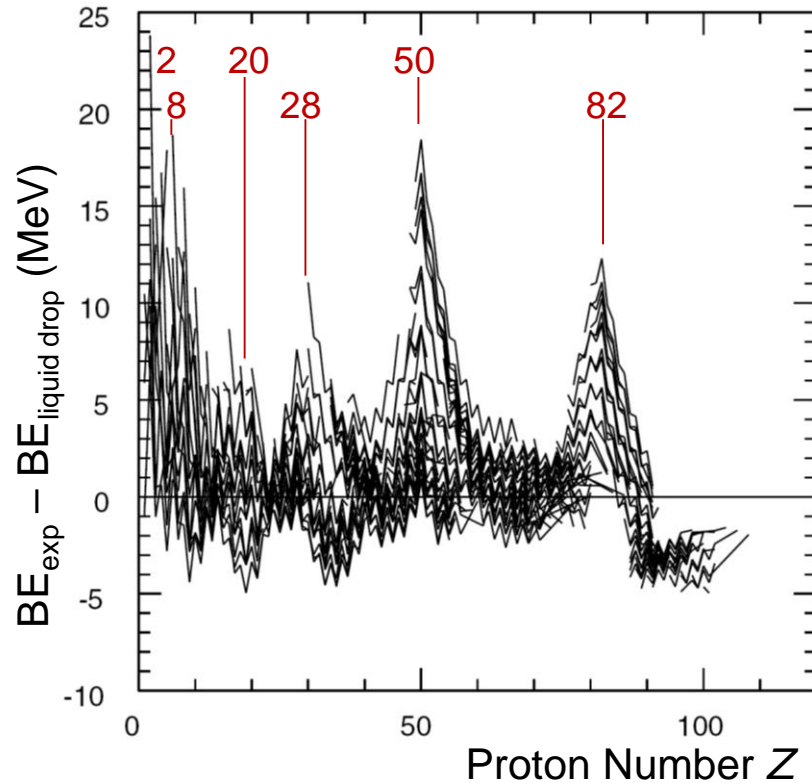
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Period																			
1	1 H																	2 He	
2	3 Li	4 Be											5 B	6 C	7 N	8 O	9 F	10 Ne	
3	11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl	18 Ar	
4	19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr	
5	37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe	
6	55 Cs	56 Ba	* 71 Lu	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn	
7	87 Fr	88 Ra	** 103 Lr	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn	113 Uut	114 Fl	115 Uup	116 Lv	117 Uus	118 Uuo	
*Lanthanoids			* 37 La	38 Ce	39 Pr	40 Nd	41 Pm	42 Sm	43 Eu	44 Gd	45 Tb	46 Dy	47 Ho	48 Er	49 Tm	50 Yb			
**Actinoids			** 89 Ac	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No			

A brief aside in atomic physics: electrons occupy shells.

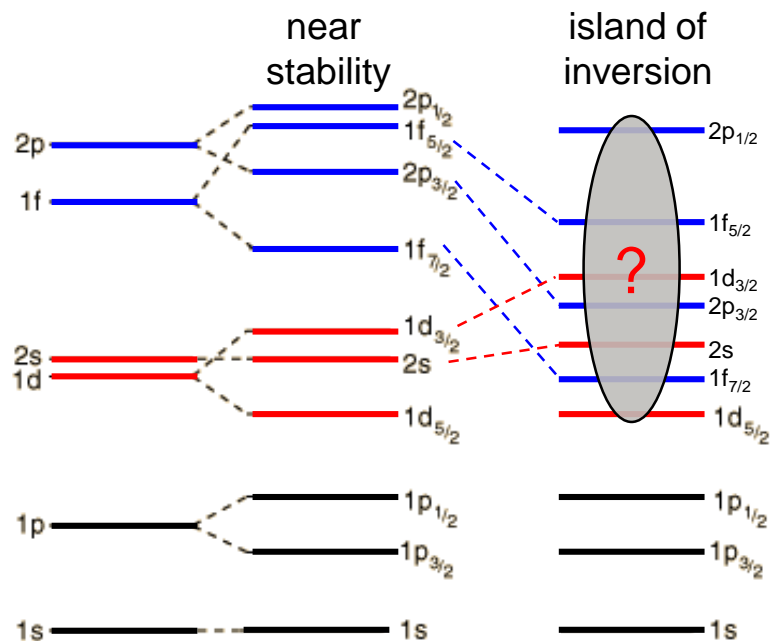
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Period																															
1	1	H																2	He												
2	3	Li	4	Be											5	B	6	C	7	N											
3	11	Na	12	Mg											13	Al	14	Si	15	P											
4	19	K	20	Ca	21	Sc	22	Ti	23	V	24	Cr	25	Mn	26	Fe	27	Co	28	Ni	29	Cu	30	Zn	31	Ga	32	Ge	33	As	
5	37	Rb	38	Sr	39	Y	40	Zr	41	Nb	42	Mo	43	Tc	44	Ru	45	Rh	46	Pd	47	Ag	48	Cd	49	In	50	Sn	51	Sb	
6	55	Cs	56	Ba	*	71	Lu	72	Hf	73	Ta	74	W	75	Re	76	Os	77	Ir	78	Pt	79	Au	80	Hg	81	Tl	82	Pb	83	Bi
7	87	Fr	88	Ra	**	103	Lr	104	Rf	105	Db	106	Sg	107	Bh	108	Hs	109	Mt	110	Ds	111	Rg	112	Cn	113	Uut	114	Fl	115	Uup
*Lanthanoids		*		37	La	38	Ce	39	Pr	40	Nd	41	Pm	42	Sm	43	Eu	44	Gd	45	Tb	46	Dy	47	Ho	48	Er	49	Tm		
**Actinoids		**		89	Ac	90	Th	91	Pa	92	U	93	Np	94	Pu	95	Am	96	Cm	97	Bk	98	Cf	99	Es	100	Fm	101	Md		



Neutrons & protons also occupy shells, which occur at “magic numbers.”



Nuclear shells may evolve as the ratio of Z/N grows more unstable,

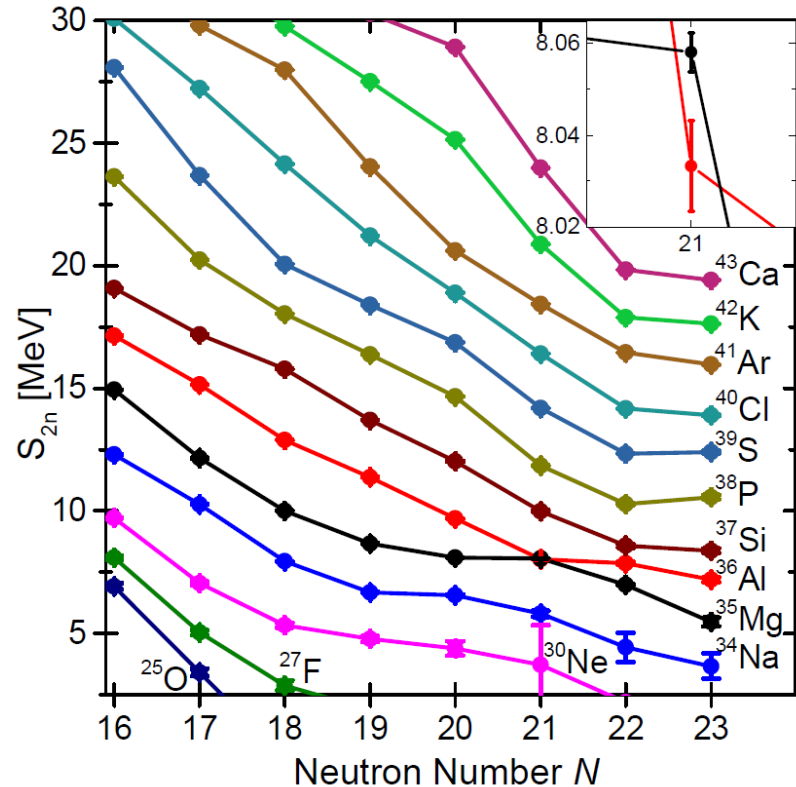
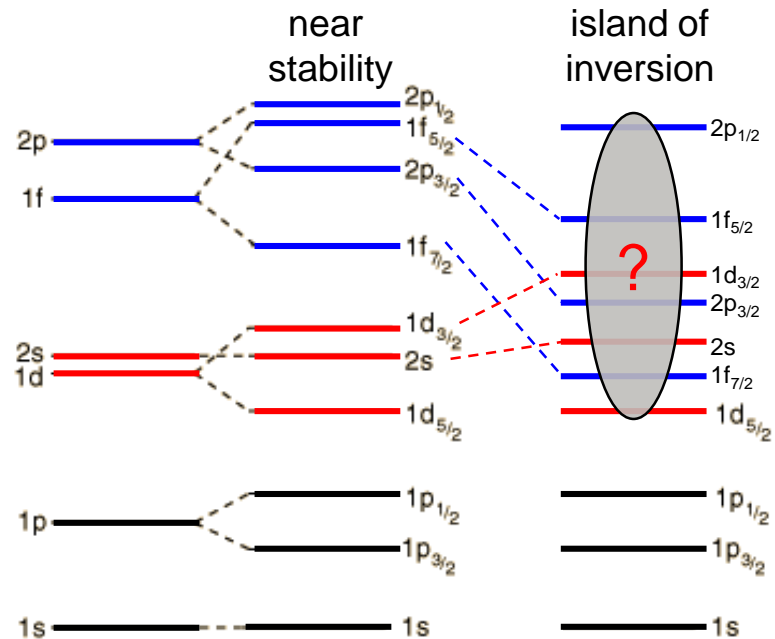


causing shells to disappear or new ones to emerge.

The TITAN campaign at $N = 20$ has measured masses of $^{29-32}\text{Na}$, $^{30-34}\text{Mg}$, $^{29-35}\text{Al}$ ($T_{1/2} \geq 12.9$ ms for ^{32}Na),

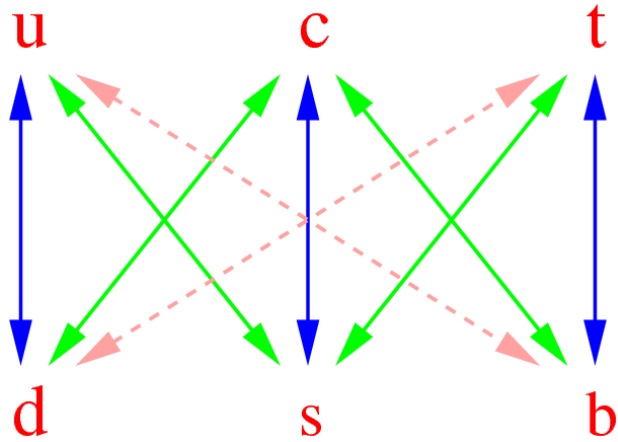
Improving precisions (often 10x) and finding deviations for Na & Mg from prior measurements.

TITAN's values indicate the disappearance of the $N = 20$ shell and large gains in correlation energy.



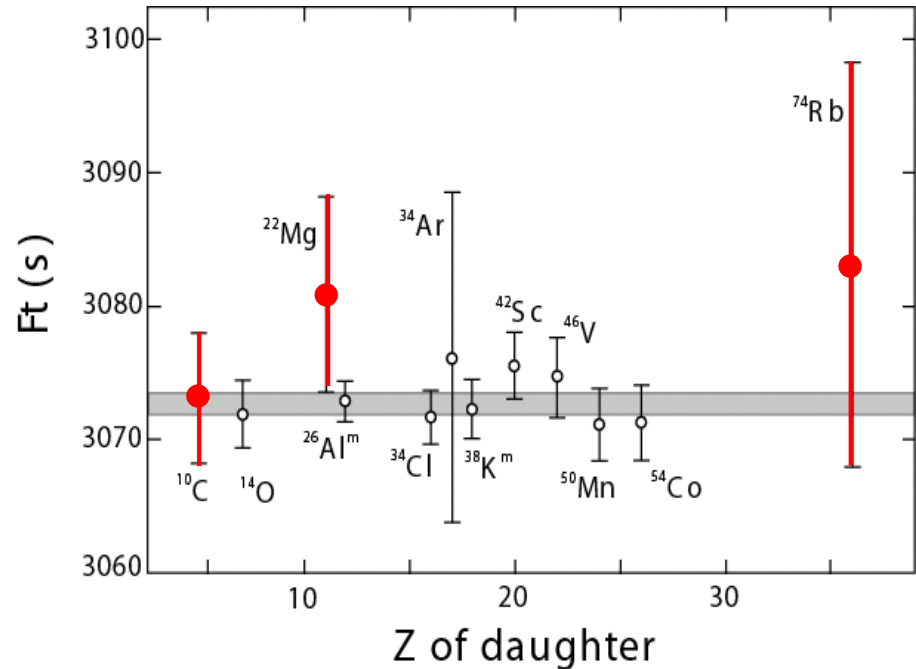
Precisions $\delta m/m$ of 10^{-7} are routine;
can we do better?

Testing the unitarity of the quark-mixing (CKM) matrix needs high-precision measurements.



Direct Q-values measured.

$$\frac{\delta m}{m} \sim 10^{-9} \text{ demonstrated.}$$



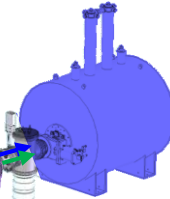
How can we purify the beam to the level required for high-precision experiments?

Two new traps will be added to extend the reach of TITAN.



UNIVERSITY
OF MANITOBA

CPET
sympathetic
cooling of HCl



MPET:
mass measurement
via $2\pi \nu_c = q/m \cdot B$

JUSTUS-LIEBIG-
UNIVERSITÄT
GIESSEN

MR-TOF MS
remove isobaric
contaminants

SCI

SCI

SCI

HCl

EBIT

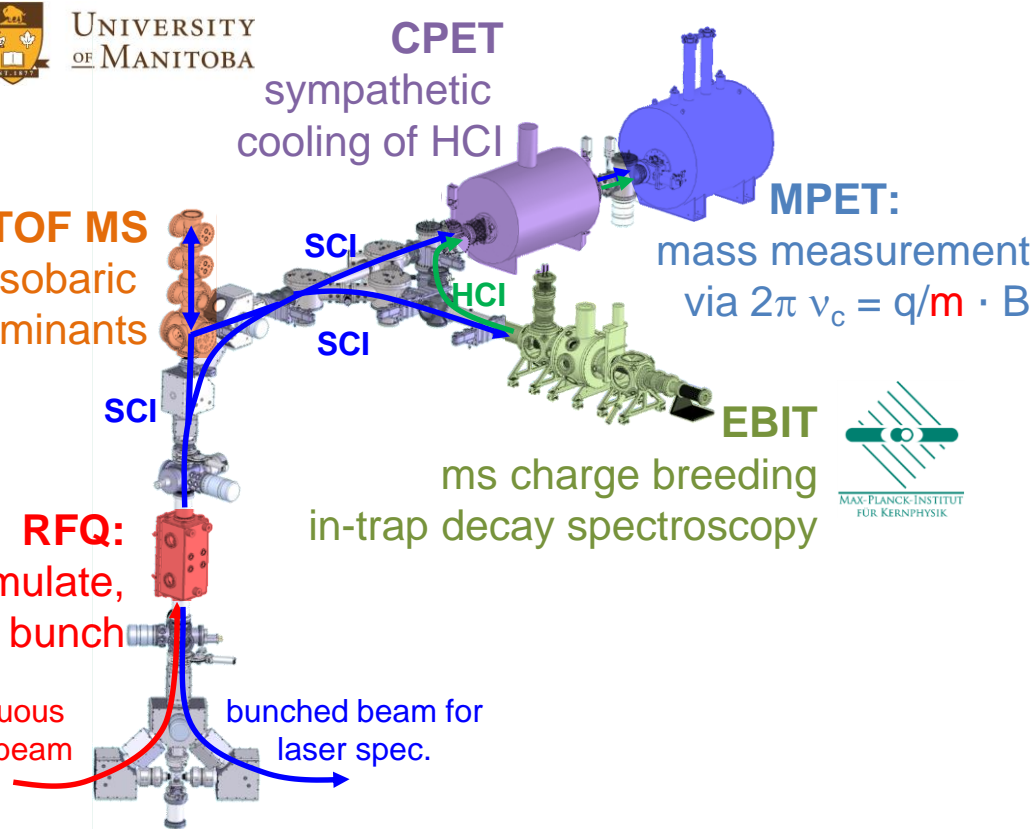
ms charge breeding
in-trap decay spectroscopy



RFQ:
accumulate,
cool, & bunch

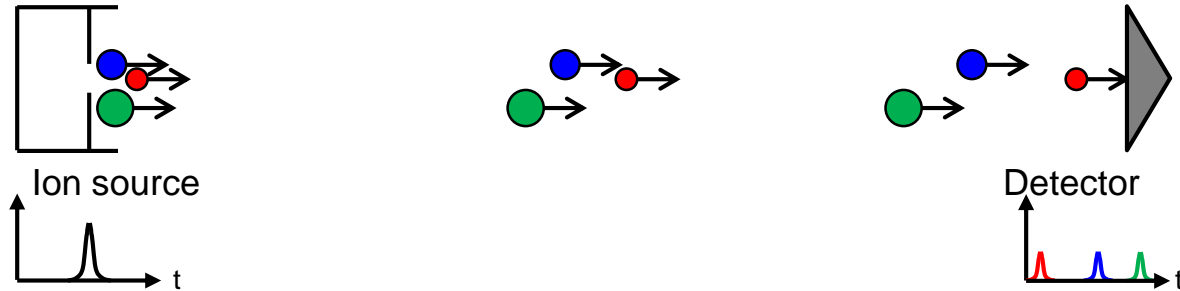
continuous
ISAC beam

bunched beam for
laser spec.



Multi-reflection time-of-flight mass spectrometers are based on simple kinematics.

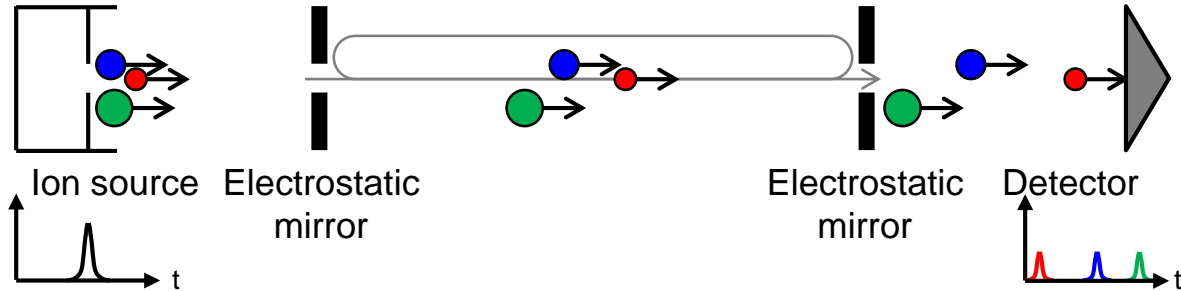
$$TOF = \frac{L}{v} = \frac{L}{\sqrt{2E}} = \sqrt{\frac{m}{q}} \int \frac{dz}{\sqrt{2V(z)}}$$



Separation increases with flight path \rightarrow longer path length

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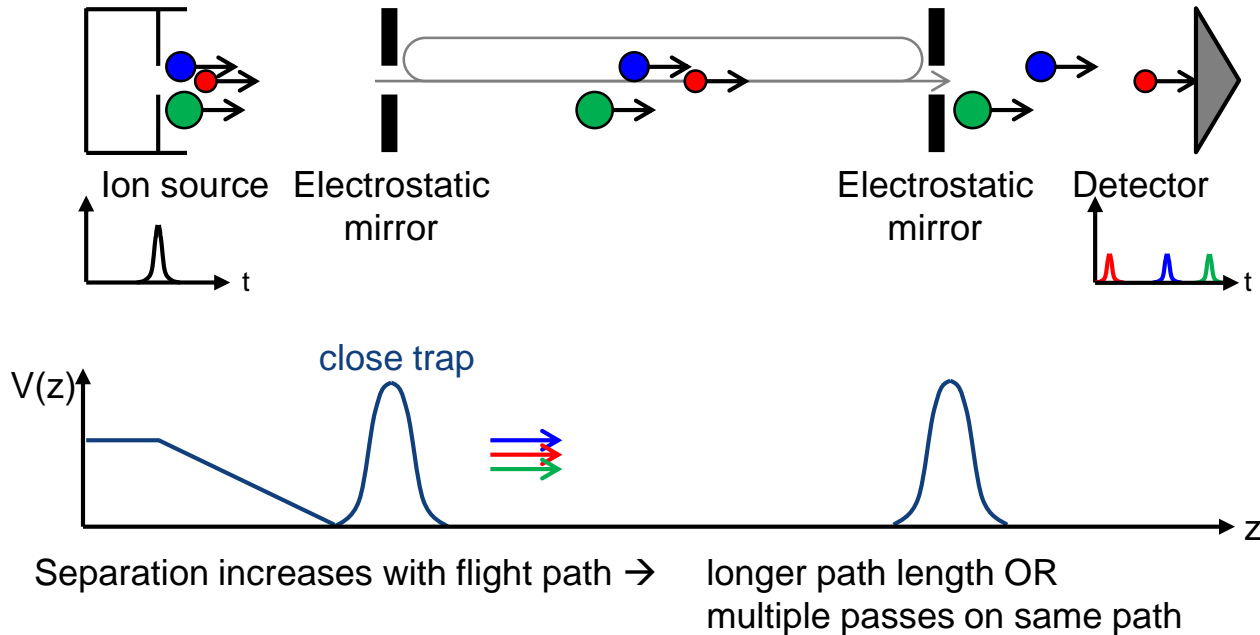
$$TOF = \frac{L}{v} = \frac{L}{\sqrt{2E}} = \sqrt{\frac{m}{q}} \int \frac{dz}{\sqrt{2V(z)}}$$



Separation increases with flight path → longer path length OR multiple passes on same path

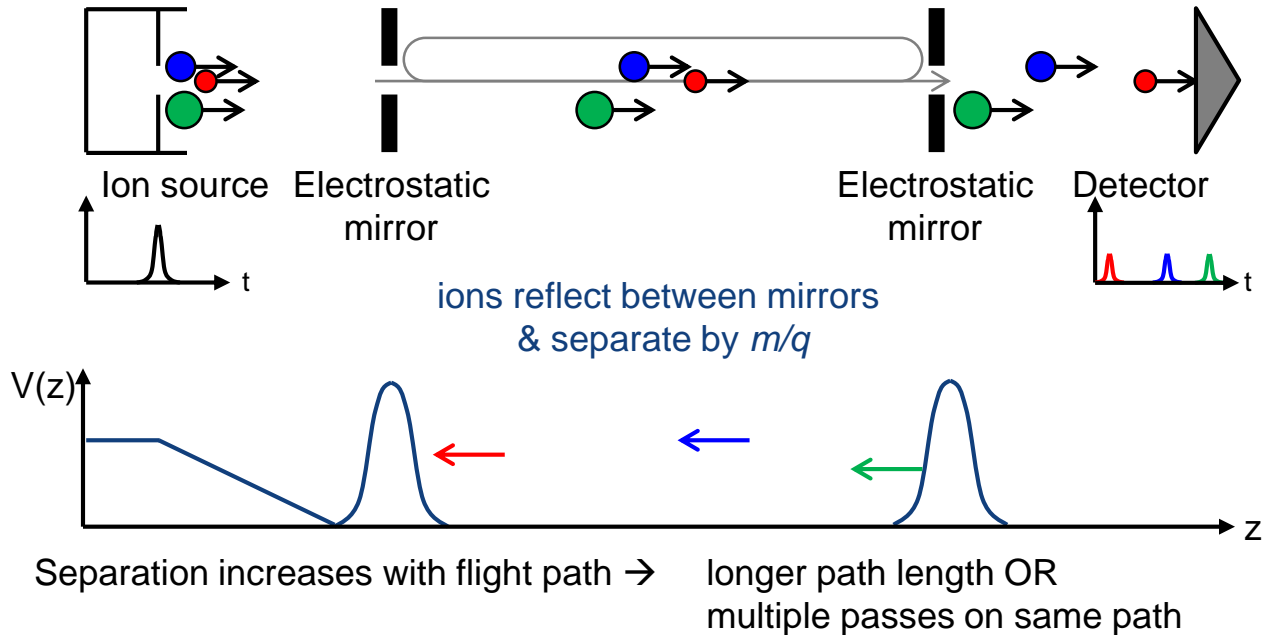
Multi-reflection time-of-flight mass spectrometers are based on simple kinematics.

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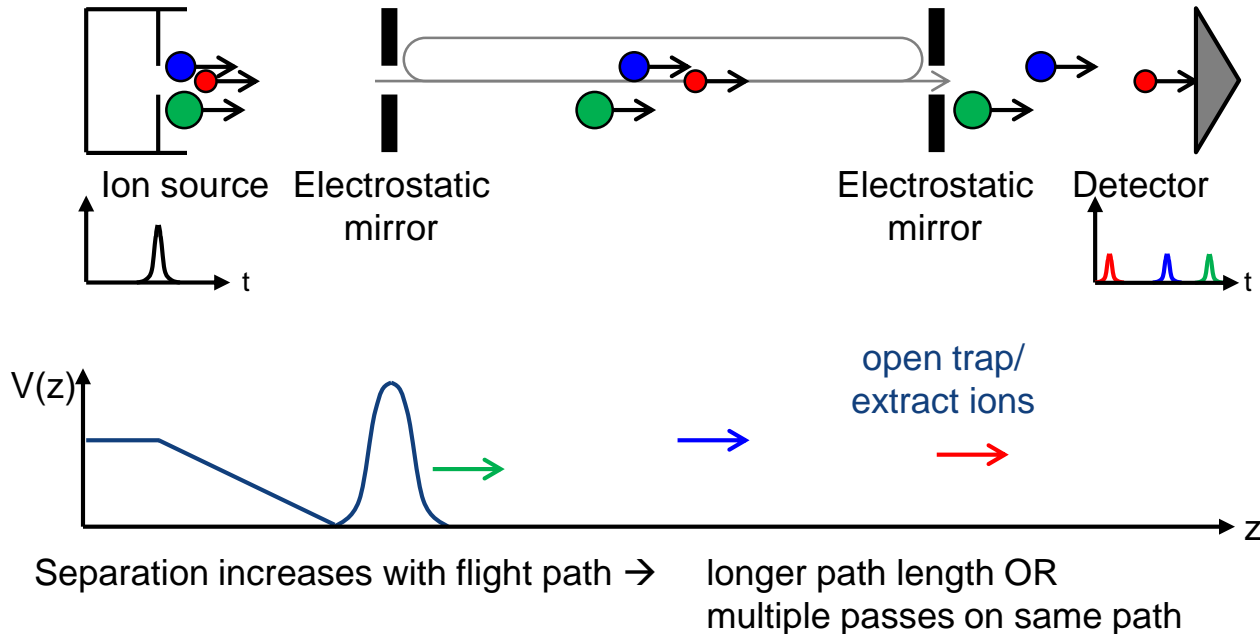
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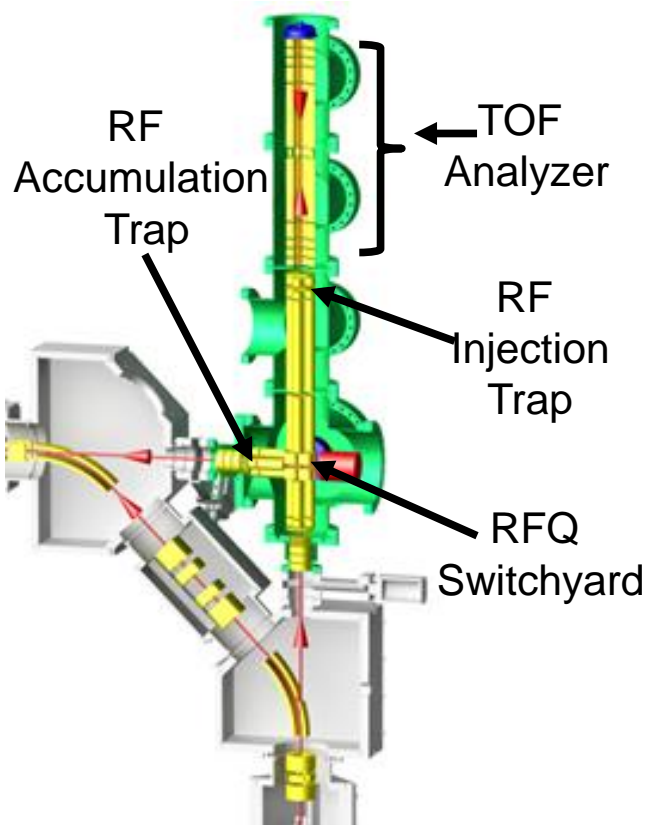


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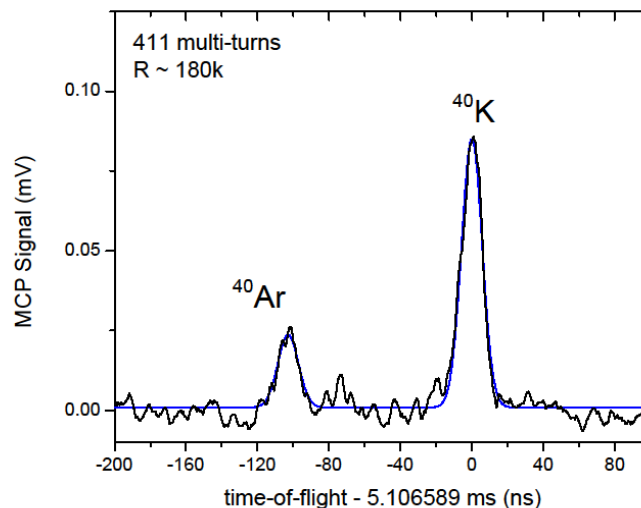
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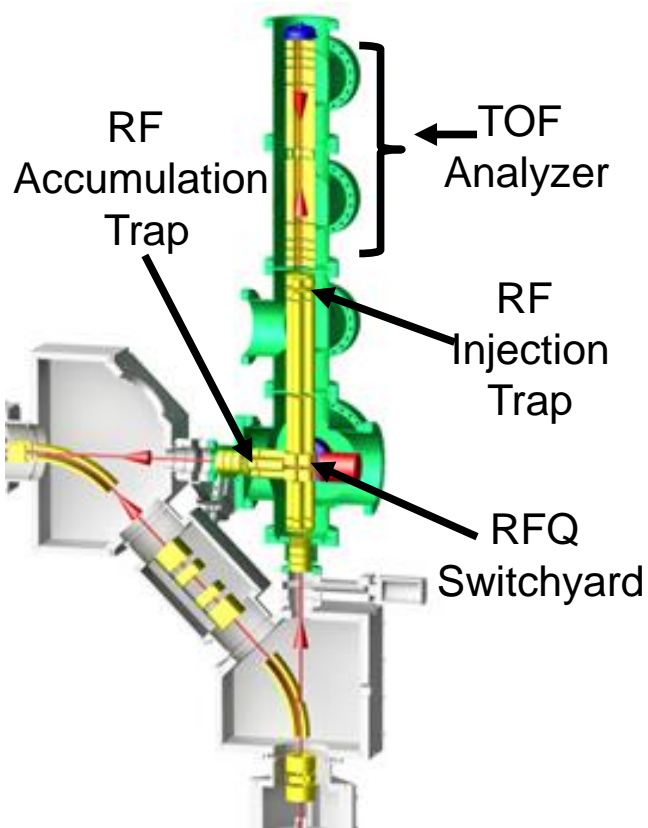
MR-TOF demonstrated to be a fast, broadband isobar separator off-line



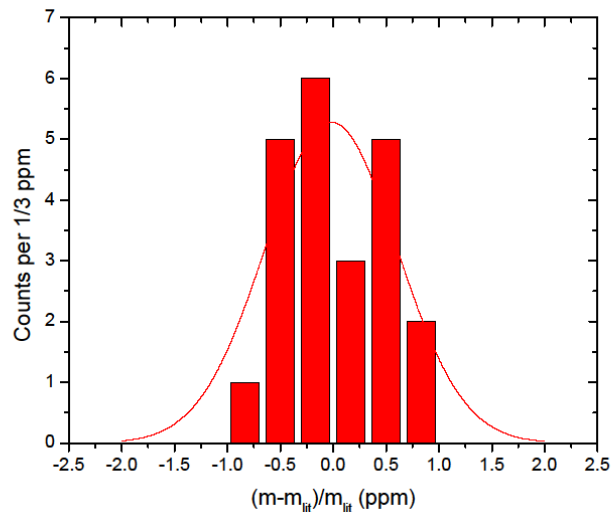
- Resolving power depends on number of passes or multi-turns
- $R \sim 10^5$ achieved in a few ms for $A = 40$



MR-TOF will be used for fast mass measurements.

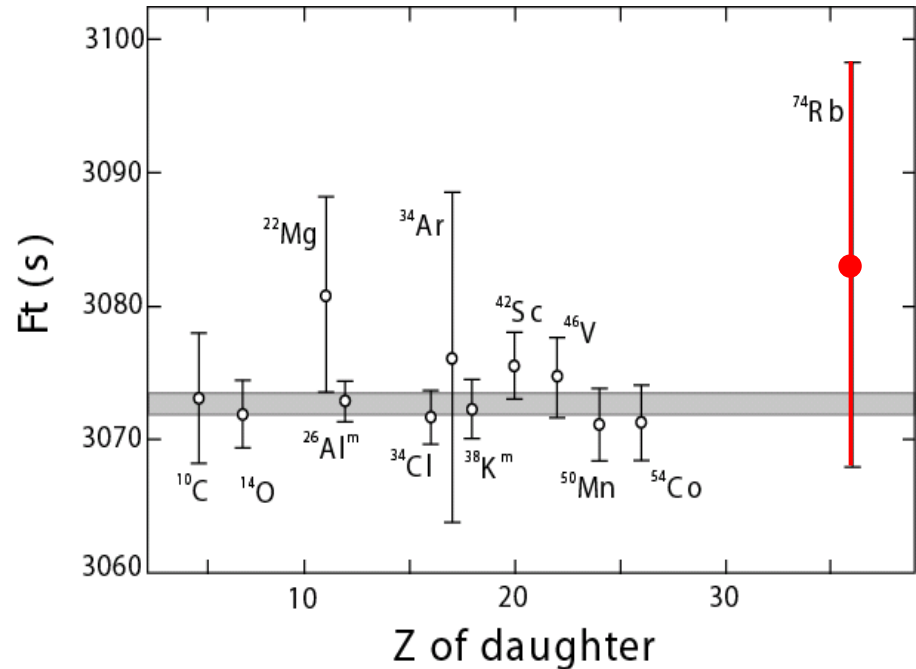
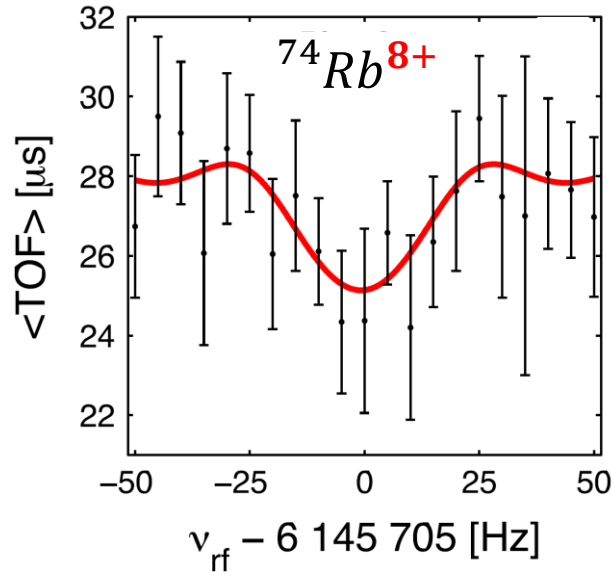


$^{40}\text{Ar}^+$ ($^{40}\text{K}^+$ for reference) agrees with AME 2012

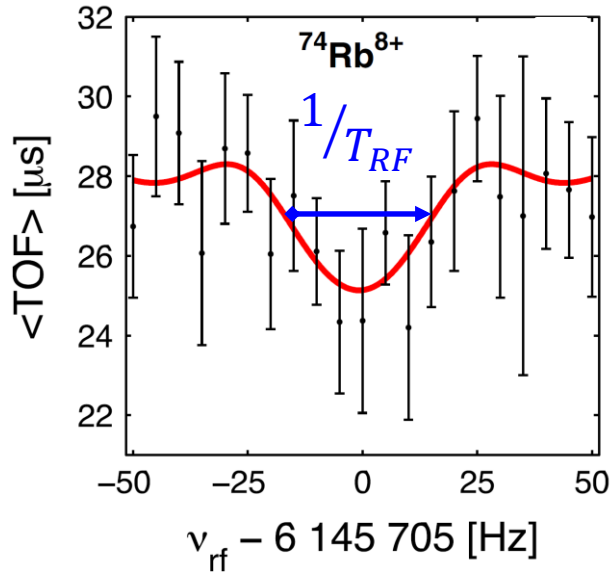


What else can be done to boost
measurement precision?

Testing the unitarity of the quark-mixing (CKM) matrix needs high-precision mass determinations.



Higher charge states can improve precision or reduce beam time requirements.



$$\frac{\delta m}{m} \propto \frac{m}{q e B T_{\text{RF}} \sqrt{N}}$$

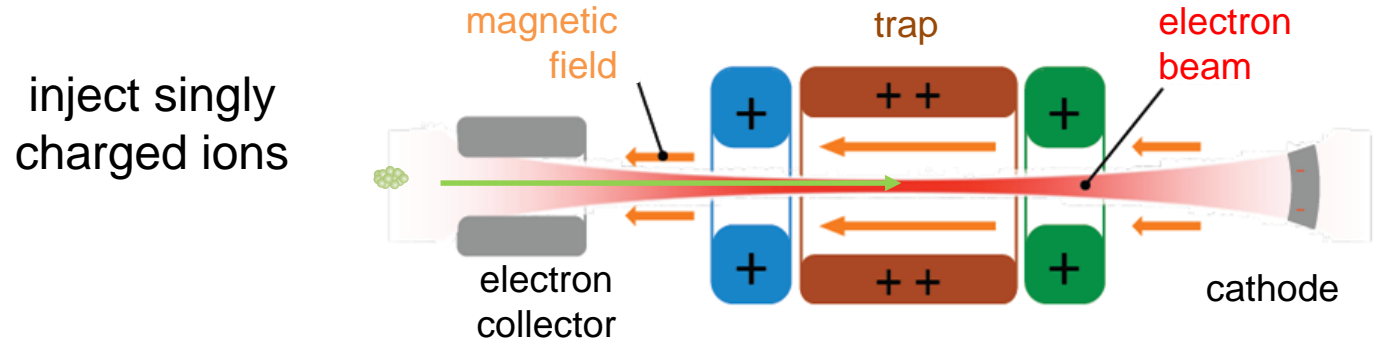
N = statistics \rightarrow limited by yield

T_{RF} = measurement time \rightarrow limited by half life

B = magnetic field \rightarrow limited by technology

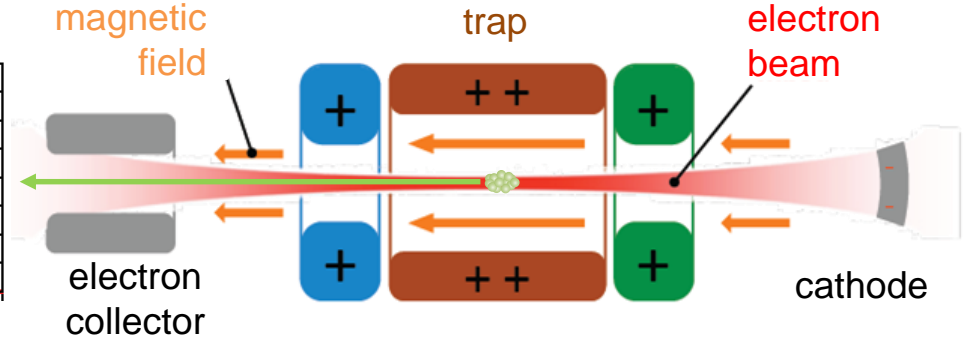
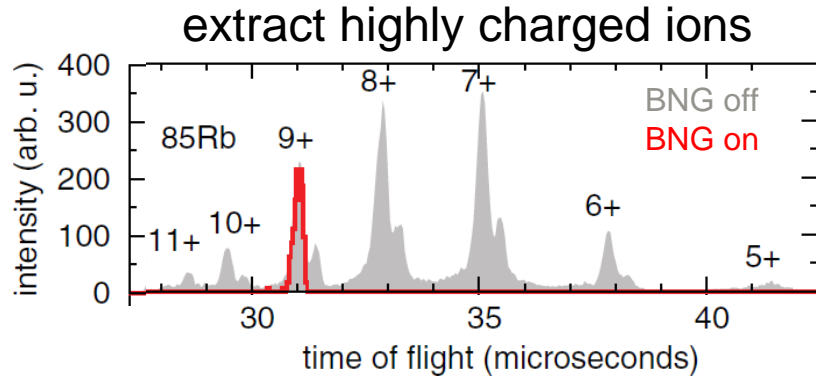
q = charge state \rightarrow limited by Z

Ions are charge bred in the Electron Beam Ion Trap, a Penning trap with an electron beam.



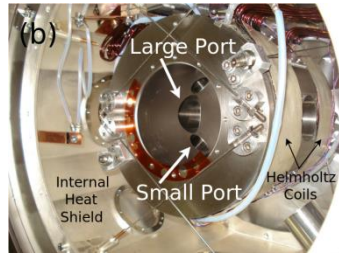
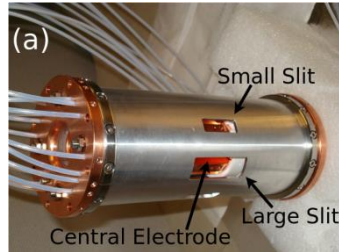
Maximum charge state depends on Z , electron beam energy, electron beam current, & charge breeding time

Ions are charge bred in the Electron Beam Ion Trap, a Penning trap with an electron beam.



Maximum charge state depends on Z , electron beam energy, electron beam current, & charge breeding time

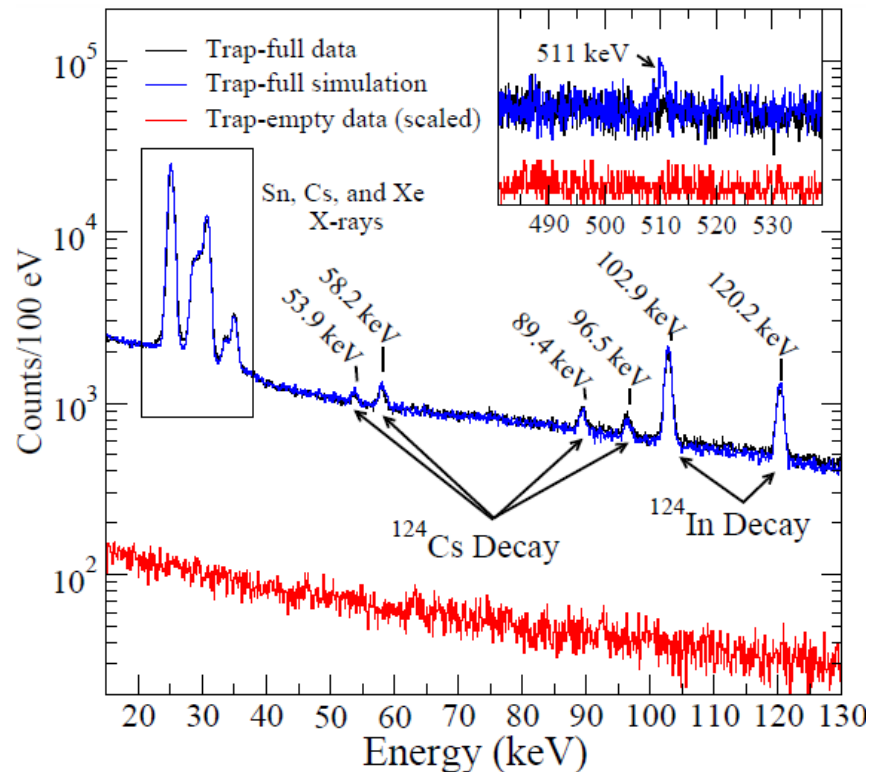
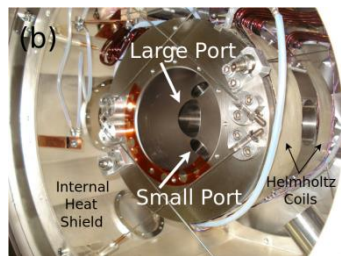
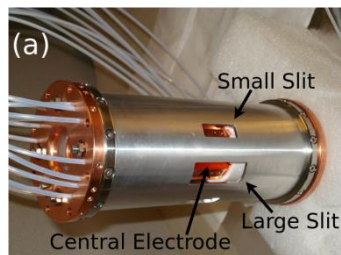
TITAN's EBIT provides optical access, allowing for in-trap decay spectroscopy.



Objective: benchmark $2\beta 2\nu$ nuclear matrix elements

Advantages: no backing material
 β 's directed away from γ detectors
trapper techniques/manipulation
high purity
HCl compatible

TITAN's EBIT provides optical access, allowing for in-trap decay spectroscopy.



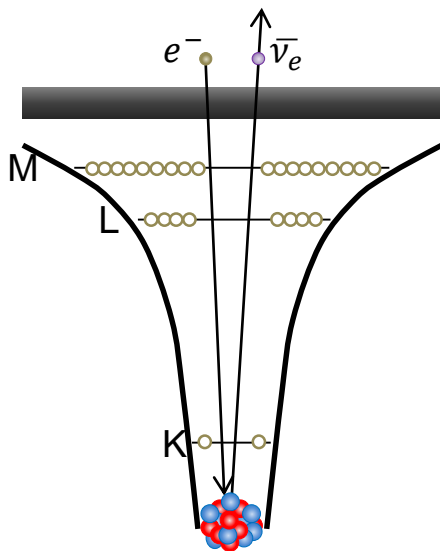
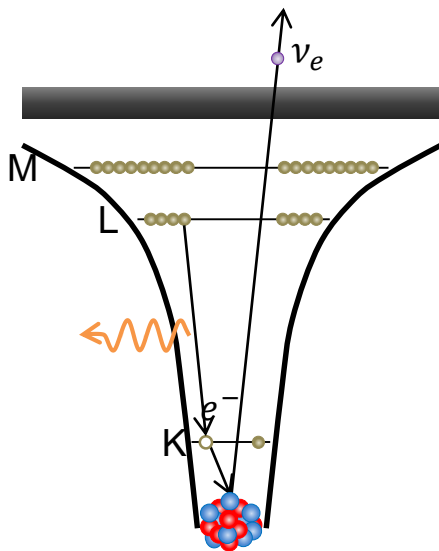
Highly Charged Ions have intrinsic scientific value.



Orbital Electron Capture

Free Electron Capture

In a neutral atom, the nucleus captures a K-shell e^- ; then the atom emits an x-ray as a higher e^- fills vacancy



For bare ions, orbital EC is forbidden

- decay only by β^+ decay if allowed
- or by capturing a free electron

The charge state impacts the decay and consequently nucleosynthesis.

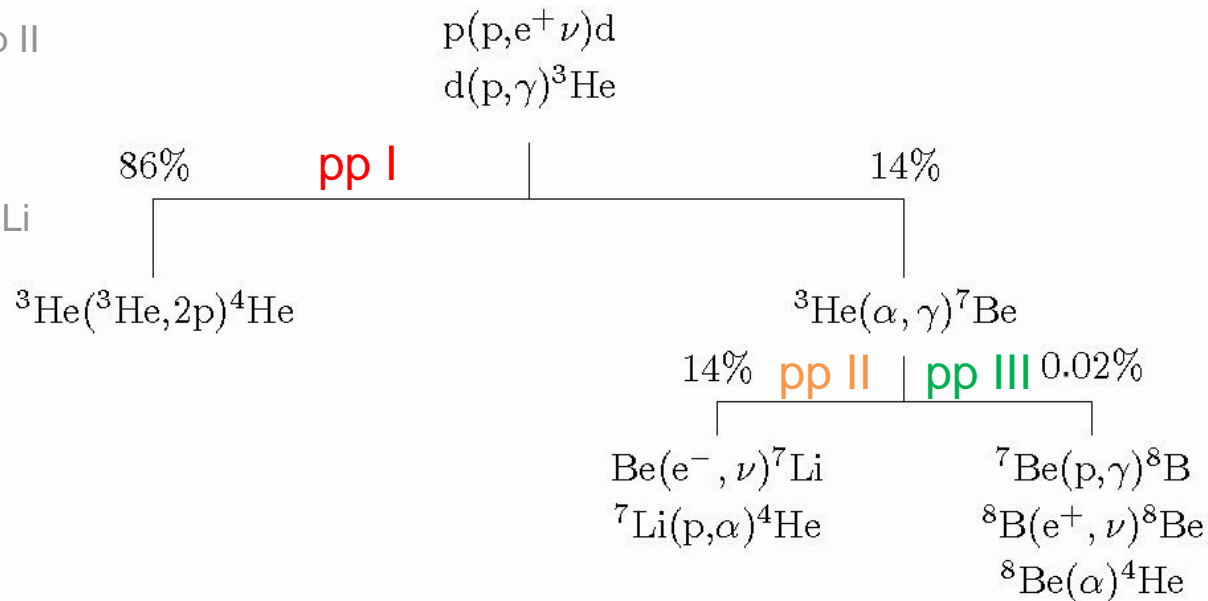
pp-chains describe how to form an α from protons

$T = 10\text{-}25 \text{ MK} \rightarrow \text{pp II}$

$T > 25 \text{ MK} \rightarrow \text{pp III}$

$T_{\odot} \approx 15 \text{ MK}$

\rightarrow ionize Be, Li



Ion traps at TITAN are used for

beam preparation

- cooling
- bunching
- charge breeding
- purifying
- in-trap decay and recapture

& precision measurements.

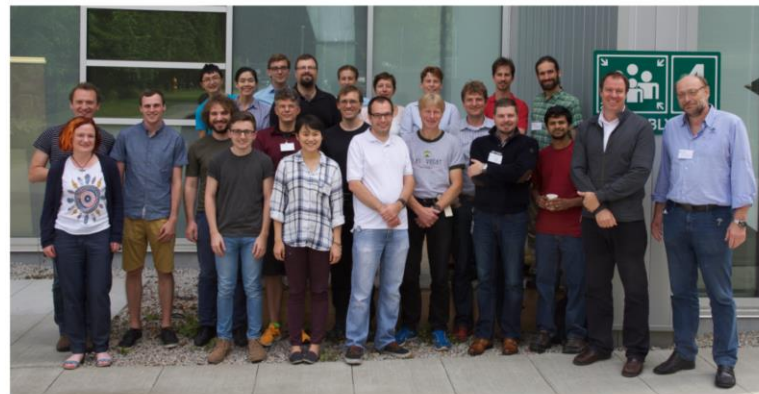
- Penning trap mass spectrometry $\rightarrow N = 20$ shell, testing CKM matrix, nuclear astrophysics
- in-trap decay spectroscopy $\rightarrow 2\nu 2\beta$ problem
- studies of highly charged ions \rightarrow radioactive decay, stellar evolution



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The TITAN collaboration



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